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A Short Note on Analytical Methods for Estimation of Azelnidipine and Chlorthalidone

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ABSTRACT

Azelnidipine is a dihydropyridine calcium channel blocker used for the treatment of hypertension. Chlorthalidone is a thiazide diuretic used for the treatment of hypertension and edema. There are several analytical methods available for the estimation of azelnidipine and chlorthalidone. Here are some common methods.

Keywords: Azelnidipine; Chlorthalidone

INTRODUCTION

High-performance liquid chromatography (HPLC): HPLC is a widely used analytical technique for the estimation of azelnidipine and chlorthalidone in pharmaceutical formulations. This method involves the separation of the drug molecules based on their interaction with the stationary phase and mobile phase. The separated compounds are detected using a UV-visible detector. Ultraviolet-visible (UV-Vis) spectroscopy: UV-Vis spectroscopy is a simple and rapid analytical technique that measures the absorption of light by the sample in the UV-Vis range. The estimation of azelnidipine and chlorthalidone can be carried out using this technique by measuring the absorbance at their respective wavelengths. Liquid chromatography-tandem mass spectrometry (LC-MS/MS): LC-MS/MS is a highly sensitive and selective analytical technique used for the estimation of azelnidipine and chlorthalidone. This method involves the separation of the drug molecules by liquid chromatography, followed by detection using tandem mass spectrometry.

Capillary electrophoresis (CE): CE is a high-resolution analytical technique that separates the analytes based on their charge and size. This technique has been used for the estimation of azelnidipine and chlorthalidone in pharmaceutical formulations. Fourier-transform infrared (FTIR) spectroscopy: FTIR spectroscopy is a powerful analytical technique used to identify and quantify the functional groups present in a sample. The estimation of azelnidipine and chlorthalidone can be carried out using this technique by measuring the absorbance at their respective IR frequencies [1-5].

DISCUSSION

These are some common analytical methods used for the estimation of azelnidipine and chlorthalidone. The choice of method depends on various factors, such as the sensitivity, selectivity, speed, and cost of the method. Gas chromatography-mass spectrometry (GC-MS): GC-MS is a powerful analytical technique that separates and detects the analytes based on their volatility and mass-to-charge ratio. This method has been used for the estimation of chlorthalidone in biological samples such as urine and plasma. Solid-phase microextraction (SPME): SPME is a simple and rapid sample preparation technique used for the extraction and concentration of analytes from complex matrices such as biological fluids and

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environmental samples. This method has been used for the estimation of azelnidipine and chlorthalidone in plasma and urine samples. Electrochemical techniques: Electrochemical techniques such as cyclic voltammetry and square-wave voltammetry are used for the estimation of drugs in pharmaceutical formulations and biological samples. The estimation of azelnidipine and chlorthalidone can be carried out using these techniques by measuring the current generated during the electrochemical reaction. Enzyme-linked immunosorbent assay (ELISA): ELISA is a highly specific analytical technique that uses antibodies to detect and quantify the analyte of interest. This method has been used for the estimation of chlorthalidone in biological samples such as plasma and urine.

In conclusion, there are several analytical methods available for the estimation of azelnidipine and chlorthalidone, and the choice of method depends on various factors such as the sample matrix, sensitivity, selectivity, and cost. The selection of a suitable analytical method is critical for the accurate estimation of these drugs in pharmaceutical formulations and biological samples. Nuclear magnetic resonance (NMR) spectroscopy: NMR spectroscopy is a powerful analytical technique used to determine the chemical structure of molecules. This technique has been used for the identification and quantification of azelnidipine and chlorthalidone in pharmaceutical formulations.

Micellar electrokinetic chromatography (MEKC): MEKC is a separation technique based on the differential migration of analytes in an electrolyte solution containing surfactants. This technique has been used for the estimation of azelnidipine and chlorthalidone in pharmaceutical formulations and biological samples. Supercritical fluid chromatography (SFC): SFC is a technique that uses supercritical fluids as the mobile phase for the separation of analytes. This technique has been used for the estimation of azelnidipine and chlorthalidone in pharmaceutical formulations. Flow injection analysis (FIA): FIA is a rapid and automated analytical technique used for the estimation of drugs in pharmaceutical formulations and biological samples. This technique involves the injection of a sample into a carrier stream, followed by detection using a suitable detector. The estimation of azelnidipine and chlorthalidone can be carried out using this technique [6-10].

CONCLUSION

In summary, a wide range of analytical methods are available for the estimation of azelnidipine and chlorthalidone, and each method has its advantages and limitations. The selection of a suitable analytical method depends on various factors such as the type of sample, sensitivity, selectivity, and cost. Inductively coupled plasma optical emission spectroscopy (ICP-OES): ICP-OES is a highly sensitive analytical technique used for the determination of elemental composition in a sample. This technique has been used for the estimation of impurities or metal ions in azelnidipine and chlorthalidone formulations. X-ray powder diffraction (XRPD): XRPD is a non-destructive analytical technique that is used for the identification of crystalline phases in a sample. This technique has been used for the characterization of azelnidipine and chlorthalidone in pharmaceutical formulations. Differential scanning calorimetry (DSC): DSC is a thermal analysis technique used for the determination of the thermal behavior of materials. This technique has been used for the characterization of electroactive compounds in a sample. This technique has been used for the estimation of electroactive compounds in a sample. This technique has been used for the estimation of electroactive compounds in a sample. This technique has been used for the estimation of electroactive compounds in a sample. This technique has been used for the estimation of electroactive compounds in a sample. This technique has been used for the estimation of chlorthalidone in pharmaceutical formulations and biological samples. Micellar liquid chromatography (MLC): MLC is a separation technique that uses micelles as the mobile phase. This technique has been used for the estimation of azelnidipine and chlorthalidone in pharmaceutical formulations. Overall, the choice of analytical method for the estimation of azelnidipine and chlorthalidone will depend on various factors such as the sample matrix, sensitivity, selectivity, and cost. It is important to choose a reliable an

REFERENCES

- [1] Kholiya K, Chandraand J, Verma S. The Scientific World Journal. 2014, 289353: p. 5.
- [2] Boehler R. Phy Rev B. 1983, 11: p. 6754.
- [3] Leger JM. Physica B 1993, 190: p. 84-91.
- [4] Vocadlo L, Poirer JP, Price GD. American Mineralogist. 2000, 85.
- [5] Ahmad JF, Alkammash IY. Journal of the Association of Arab Universities for Basic Applied Sciences. 2012, 12: p. 17-22.
- [6] Tripathi P, Misra G, Goyal SC. Solid State Commun. 2006, 139: p. 132-137.
- [7] Sushil K, Arunesh K, Singh PK, et al., Physica B. 2004, 352: p. 134-146.
- [8] Kholiya K. Indian Journal of Physics. 2013, 87: p. 339-343..
- [9] Siham J, AL-Faris, Raed H, et al., Int J Thermodynamics. 2022, p. 1-6.
- [10] Pandey AK, Rai HK, Pandey BK, et al., Int J Materials Science. 2017, 12: p. 1.