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3D-QSAR analysis on 6-(1-benzyl-1H-pyrrol-2-yl)-2, 4-dioxo-5-hexenoic acid derivatives as recombinant HIV-1 integrase inhibitors

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ABSTRACT

HIV-1 integrase is a fascinating target for designing of novel HIV-1 Integrase inhibitors. Due to the development of resistance by the use of already developed inhibitors the novel inhibitors are designed that can target Integrase with higher selectivity and less toxicity profiles. The present work describes the 3D QSAR studies on series of 6-(1-Benzyl-1H-pyrrol-2-yl)-2, 4-dioxo-5-hexenoic acids for establishing quantitative relationship between biological activity and their physicochemical properties. This study was performed with 47 compounds (data set) using manual selection method and simulated annealing algorithm for the division of the data set into training and test set. In this analysis, three statistical significant models were obtained using PLS as the statistical method. The most significant model is having ($q^2 = 0.608$) and ($pred_r^2 = 0.699$). Model showed that steric (S_{881}), (S_{184}) and electrostatic (E_{496}) interactions play important role in modulating the HIV-1 integrase inhibitory activity.

INTRODUCTION

Acquired immuno deficiency syndrome (AIDS) is a major global public health issue of the 21st century that is characterized by a cascade of infections resulting from the damage to the immune system caused by Lentivirus i.e. human immunodeficiency virus (HIV) belonging to the family Retroviridae [1,2]. AIDS is the most serious stage of HIV due to which the body becomes susceptible to numerous infections and diseases. The number of T cells that plays a major role in the immune system is declined to a severe level with the alteration in the immune system [3]. In the HIV life cycle the enzymes which play a vital role are reverse transcriptase (RT), protease (PR) and integrase (IN). With the help of the reverse transcriptase (RT) there is the transcription of the viral RNA genome into viral complementary DNA (cDNA). The IN helps in the insertion of viral cDNA into the host cell genome and the cleavage of newly synthesized polypeptides into single viral proteins occurs by viral PR [4]. Since last decades, the RT and PR are seen as the main targets for the inhibition of HIV replication.

The regimen known as highly active antiretroviral therapy (HAART) for the treatment of HIV includes the use of nucleoside RT inhibitors (NRTIs), non-nucleoside RT inhibitors (NNRTIs) and protease inhibitors (PRIs) resulting in the resistance due to the long term use. So a remedial intervention is needed to stop the epidemic completely from the roots [5]. Hence, as a result the enzyme IN has emerged as an attractive target for discovery of safe, efficient and effective HIV-IN inhibitors [6]. There is no known human counterpart to integrase hence integrase as a target is an attractive approach which will prove to be more advantageous over existing therapies due to high selectivity and less toxicity profiles [7]. Moreover, there is an assumption that the addition of the HIV-IN inhibitor into the regimen can

MATERIALS AND METHODS

Dataset and Data Selection

6-(1-Benzyl-1*H*-pyrrol-2-yl)-2, 4-dioxo-5-hexenoic acid derivatives and their HIV -1 IN inhibitory activities against ST were taken from the literature reported by Roberta Costi *et al.* [12]. The compounds included diketo acid (DKA) derivatives like 4-[5-(benzoylamino)thien-2-yl]-2,4-dioxobutanoic acid and DNA aptamers. The different kind of structurally different compounds as inhibitors were reported including tropolones (4), madurahydroxylactone derivatives (5) and 2-hydroxyisoquinolin-1, 3(2*H*, 4*H*)-diones (6) as shown in Fig.2.

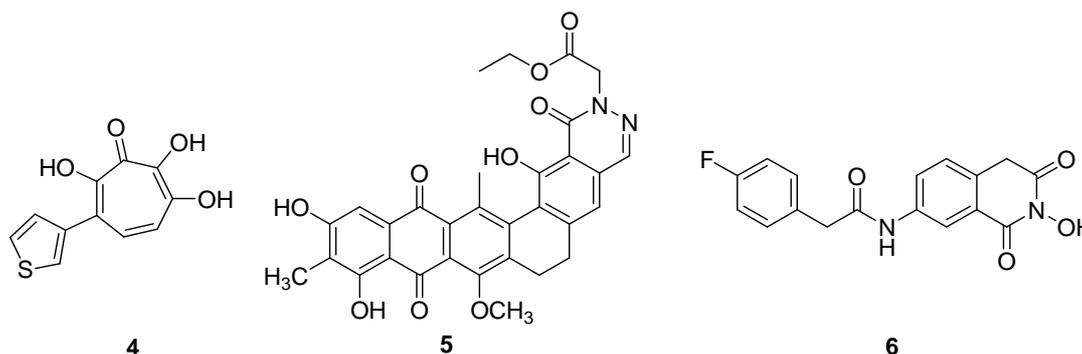


Fig. 2- Tropolones (4), Madurahydroxylactone (5), 2-hydroxyisoquinolin-1, 3(2*H*, 4*H*)-diones (6)

A data set of 47 molecules ($n=47$) are considered for the analysis along with their reported strand transfer inhibitory activity (IC_{50} , inhibitory concentration, in μM) expressed in terms of logarithm of inverse of their inhibitory concentration ($-\log IC_{50}$) as shown in **Table 1**. The activity data (IC_{50}) was given in μM value. All values have been converted to the logarithmic scale [pIC₅₀] and the data saved as .txt file. After the conversion, structures are saved as .mol2 file in QSAR Plus 3D window. The analogues as reported in **Table 1** have undergone 3D QSAR studies for drug design along with their general core structure using V Life Molecular Design Suite (MDS) Software. All the molecules in series were aligned based on atom overlapping alignment. The derived 3D-QSAR PLS model for the chemical series reported as anti-HIV integrase agents give insight into the influence of various interactive fields on the activity and thus help in designing the HIV integrase inhibitory activity of novel compounds.

The core 3D structures have been created and generated using Molecular Operating Environment (MOE) which were subjected to energy minimization using force field: Merck Molecular Force Field (MMFF94x) and Austin Model 1 (AM1) with 0.01 and 0.001 kcal/mol as the energy gradients. The force field was applied in order to obtain and ensure the minimum energy conformer. All the other structures mentioned in **Table 1** were generated in MOE using the force fields as mentioned above with the same gradients by modifying the core structure. All these energy minimized structures of compounds reported in **Table 1** have been harbored to the 3D-QSAR of QSARPLUS module for computational studies, hence computing the parameters.

Descriptors Evaluation

In these steric, electrostatic and hydrophobic parameters have been calculated using V Life MDS Software.

Model Development

In 3D-QSAR analysis for model development three methods i.e. Random selection method, Manual data selection method, Sphere Exclusion method were used for creation of training and test set. In this paper we have selected Manual data selection method and Random Selection method by conflicting disparate values. After the creation of training and test set, the statistical method like PLS (Partial Least Square) have been used for model building. PLS method helps in the establishment of relationship between the dependent variables (biological activity) and the independent parameters (electrostatic, steric and hydrophobic fields).

Variable Selection

In PLS method Stepwise variable selection is done. Simulated annealing method is used for 3D-QSAR analysis.

RESULTS AND DISCUSSION

The various regression methods were performed on all the 47 molecules of the selected series for QSAR analysis. The following considerable and significant 3D-QSAR models with equations were obtained using the alignment of energy minimized confirmations of diketo acid derivatives and PLS method as mentioned above.

Model 1

$$-\log IC_{50} = (8.368) S_{716} + (0.719) E_{496} + (19.810) S_{861} - (3.926) H_{285} + 11.01$$

$n = 38, r^2 = 0.712, q^2 = 0.624, r^2 T = 0.625, F\text{-Test} = 43.294$

Here n is the number of compounds in training set and $r^2 T$ is the r^2 of test set compounds. The equation of model 1 shows the effects of different parameters at grid points 716, 496, 861 and 285. The steric and electrostatic parameters give the positive contribution at points 716, 861 and 496 whereas the contribution by the hydrophobic parameter is negative at point 285. Figure 3a shows that S_{861} correspond to the substitution at X of $-COOX$ and S_{716} signifies the position 2nd of the diketo acid chain. The E_{496} indicates the electrostatic effect at the position linking N of the pyrrol and the phenyl ring whereas the negative contribution by hydrophobic parameter is unfavorable at R_4 substitution of the phenyl ring. The compounds in the test set were 9. The linear regression is obtained by fitting a best fit straight line to the data. The input parameters included cross correlation limit 0.8; iteration at given temperature 10; seed 1; decrease in temperature 1; perturbation limit 2; terms in model 4; variance cut off 0; auto scaling and four number of components.

Model 2

$$-\log IC_{50} = (92.520) S_{881} + (35.297) S_{184} + (0.065) E_{496} - (9.640) H_{71}$$

$n = 38, r^2 = 0.686, q^2 = 0.6990, r^2 T = 0.608, F\text{-Test} = 38.352$

The equation of the model shows the effects of different parameters at grid points 71, 881, 184 and 496. In this equation also the steric and electrostatic parameters the positive contribution is provided at points 881, 184 and 496 whereas the contribution by the hydrophobic parameter is negative at point 71. Figure 3b shows that S_{881} correspond to the substitution at X of $-COOX$ and S_{184} signifies the substitution at R_4 of the phenyl ring. The E_{496} indicates the electrostatic effect at the position linking N of the pyrrol and the phenyl ring whereas the negative contribution by hydrophobic parameter is unfavorable at R_5 substitution of the phenyl ring. The compounds in the test set were 9. The linear regression is obtained by fitting a best fit straight line to the data. The input parameters included cross correlation limit 0.5; iteration at given temperature 10; seed 2; decrease in temperature 2; perturbation limit 2; terms in model 4; variance cut off 0; auto scaling and four number of components.

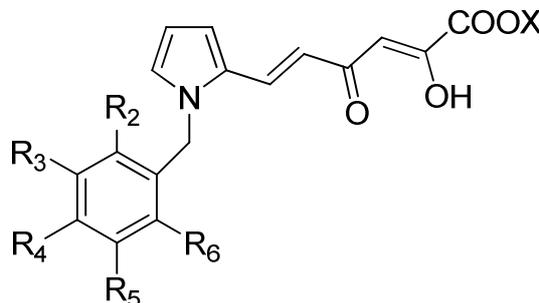
Model 3

$$-\log IC_{50} = (69.42) S_{94} + (290.61) S_{891} + (0.065) E_{496} - (5.420) H_{192}$$

$n = 38, r^2 = 0.687, q^2 = 0.598, r^2 T = 0.831, F\text{-Test} = 37.473$

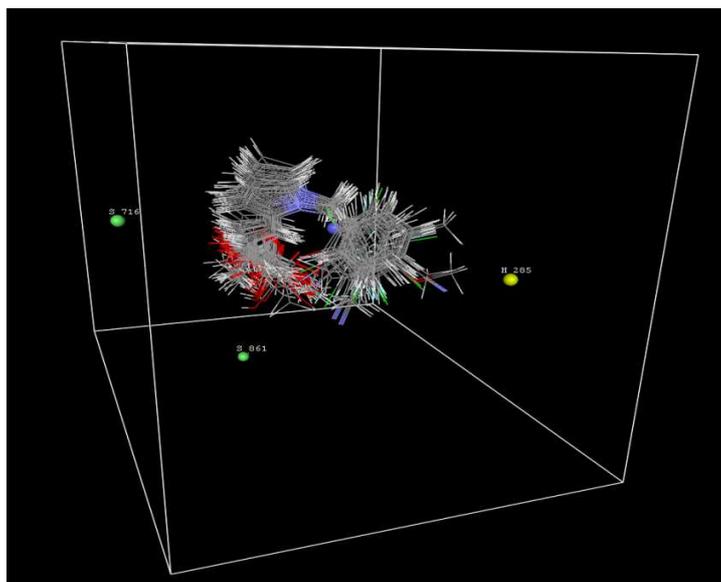
The equation of model 3 shows the effects of different parameters at grid points 94, 891, 496 and 192. This equation signifies the positive contribution by steric and electrostatic parameters at points 94, 891 and 496 whereas the contribution by the hydrophobic parameter is negative at point 192. Figure 3c shows that S_{891} correspond to the substitution at X of $-COOX$ and S_{94} signifies the substitution at R_4 of the phenyl ring. The E_{496} indicates the electrostatic effect at the position linking N of the pyrrol and the phenyl ring whereas the negative contribution by hydrophobic parameter is unfavorable at R_5 substitution of the phenyl ring. The linear regression is obtained by fitting a best fit straight line to the data. The compounds in the test set were 10. The input parameters included cross correlation limit 0.7; iteration at given temperature 10; seed 2; decrease in temperature 1; perturbation limit 2; terms in model 4; variance cut off 0; auto scaling and four number of components. The test set prediction by equations 1, 2 and 3 is in agreement with their experimental values as shown in Table 1.

Table 1- Observed and Predicted activity of diketo acid derivatives

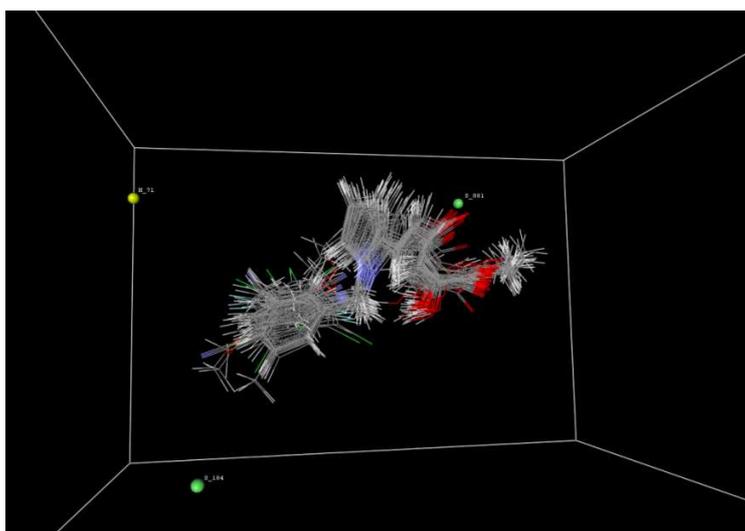


| Compound | Substitutions | | | | | | IC ₅₀ | -log IC ₅₀ | | | |
|---------------------|----------------|----------------|----------------|----------------|----------------|----|------------------|-----------------------|--------------|--------------|--------------|
| | R ₂ | R ₃ | R ₄ | R ₅ | R ₆ | X | | Observed | EMC (Model1) | EMC (Model2) | EMC (Model3) |
| 01 | H | H | H | H | H | Et | 15 | 4.823 | 4.778 | 5.0028 | 5.085 |
| 02 | Me | H | H | H | H | Et | 32 | 4.494 | 4.315 | 3.961 | 4.153 |
| 03 | H | Me | H | H | H | Et | 8 | 5.096 | 4.742 | 4.742 | 4.627 |
| 04 | F | H | H | H | H | Et | 1 | 6.000 | 5.086 | 5.444 | 5.670 |
| 05 ^{a,b,c} | H | F | H | H | H | Et | 11 | 4.958 | 5.152 | 5.440 | 5.355 |
| 06 | H | H | F | H | H | Et | 98 | 4.008 | 5.013 | 5.127 | 5.105 |
| 07 | H | Cl | H | H | H | Et | 6 | 5.221 | 4.916 | 5.149 | 4.139 |
| 08 | H | H | Cl | H | H | Et | 42 | 4.376 | 4.535 | 4.840 | 4.813 |
| 09 | CN | H | H | H | H | Et | 9 | 5.045 | 4.930 | 5.379 | 5.245 |
| 10 | H | CN | F | H | H | Et | 13 | 4.886 | 5.032 | 5.079 | 4.871 |
| 11 | OMe | H | H | H | H | Et | 23 | 4.638 | 4.938 | 4.856 | 4.676 |
| 12 ^{a,b,c} | H | H | OMe | H | H | Et | 110 | 3.958 | 4.733 | 4.011 | 3.855 |
| 13 | OEt | H | H | H | H | Et | 12 | 4.920 | 4.998 | 4.807 | 4.798 |
| 14 | H | Me | H | H | H | Et | 19 | 4.721 | 4.571 | 4.424 | 4.563 |
| 15 | F | F | H | H | H | Et | 0.53 | 6.275 | 6.168 | 5.974 | 5.938 |
| 16 | F | H | F | Me | H | Et | 1 | 6.000 | 6.012 | 5.764 | 5.817 |
| 17 ^{a,b,c} | F | H | H | H | H | Et | 0.45 | 6.346 | 5.546 | 5.577 | 5.924 |
| 18 | F | H | H | H | H | Et | 4.00 | 5.397 | 6.315 | 6.460 | 6.436 |
| 19 | H | F | F | F | H | Et | 0.60 | 6.221 | 6.011 | 5.928 | 5.934 |
| 20 | H | F | H | H | F | Et | 0.49 | 6.309 | 5.891 | 6.179 | 6.002 |
| 21 | Cl | H | H | H | H | Et | 1.70 | 5.769 | 5.729 | 5.588 | 5.490 |
| 22 ^{a,c} | H | Cl | H | F | H | Et | 8.00 | 5.096 | 5.167 | 5.412 | 5.352 |
| 23 | H | H | H | H | H | H | 0.09 | 7.045 | 6.408 | 6.602 | 6.078 |
| 24 | Me | H | H | H | Cl | H | 0.17 | 6.769 | 6.258 | 6.330 | 6.043 |
| 25 ^c | H | Me | H | Cl | H | H | 1.3 | 5.886 | 6.242 | 6.326 | 6.146 |
| 26 | H | H | Me | H | H | H | 1.2 | 5.920 | 6.003 | 5.782 | 5.777 |
| 27 ^{a,b,c} | F | H | H | H | H | H | 0.98 | 6.008 | 6.699 | 6.710 | 6.807 |
| 28 | H | F | H | H | H | H | 0.92 | 6.036 | 6.349 | 6.566 | 6.407 |
| 29 | H | H | F | H | H | H | 0.026 | 7.585 | 6.269 | 6.345 | 6.432 |
| 30 ^{a,b,c} | H | Cl | H | H | H | H | 0.31 | 6.508 | 5.928 | 5.989 | 6.009 |
| 31 | H | H | Cl | H | H | H | 4.1 | 5.387 | 5.663 | 5.794 | 5.827 |
| 32 | CN | H | H | H | H | H | 6 | 5.221 | 6.020 | 6.358 | 6.318 |
| 33 | H | CN | H | H | H | H | 0.75 | 6.124 | 5.974 | 6.146 | 6.083 |
| 34 ^{a,b,c} | H | H | CN | H | H | H | 1.7 | 5.769 | 6.136 | 5.802 | 5.685 |
| 35 | OMe | H | H | H | H | H | 0.53 | 6.275 | 6.066 | 5.930 | 5.974 |
| 36 | H | H | OMe | H | H | H | 4.1 | 5.387 | 5.971 | 5.263 | 5.283 |
| 37 | OEt | H | H | H | H | H | 0.31 | 6.508 | 6.544 | 6.473 | 6.694 |
| 38 | H | Me | H | Me | H | H | 1.6 | 5.795 | 5.715 | 5.573 | 5.770 |
| 39 ^{a,b,c} | F | F | H | H | H | H | 0.059 | 7.229 | 6.534 | 6.487 | 6.632 |
| 40 | F | H | F | H | H | H | 0.042 | 7.376 | 6.922 | 6.859 | 6.957 |
| 41 ^c | F | H | H | F | H | H | 0.052 | 7.283 | 7.310 | 7.207 | 7.339 |
| 42 | F | H | H | H | F | H | 0.15 | 6.823 | 6.569 | 6.577 | 6.570 |
| 43 | H | F | F | H | H | H | 1.60 | 5.795 | 6.428 | 6.320 | 6.367 |
| 44 ^{a,b,c} | H | F | H | F | H | H | 1.2 | 5.920 | 6.223 | 5.884 | 5.979 |
| 45 | Cl | H | Cl | H | H | H | 4.9 | 5.309 | 5.965 | 5.704 | 5.785 |
| 46 | Cl | H | H | H | Cl | H | 0.17 | 6.769 | 6.810 | 6.324 | 6.476 |
| 47 ^b | H | Cl | H | Cl | H | H | 0.97 | 6.013 | 6.037 | 6.010 | 6.279 |

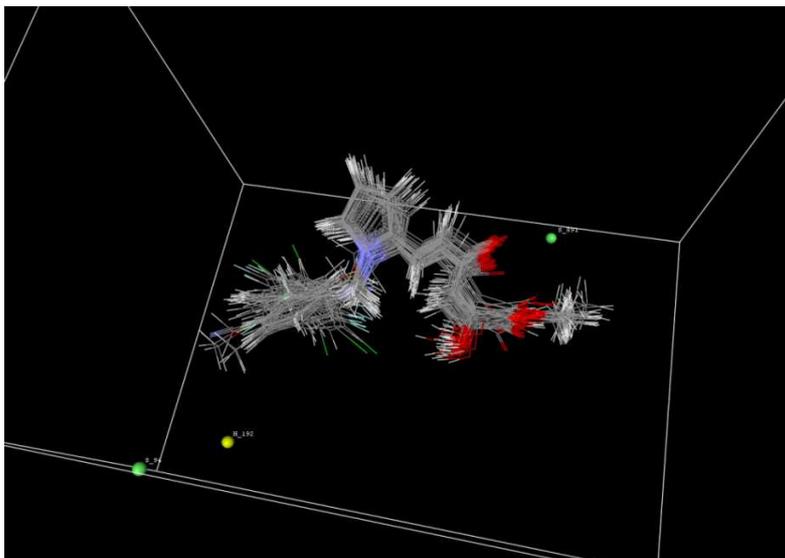
^a: test set compound in model 1; ^b: test set compound in model 2; ^c: test set compound in model 3.



Model 1 (Fig. 3a)



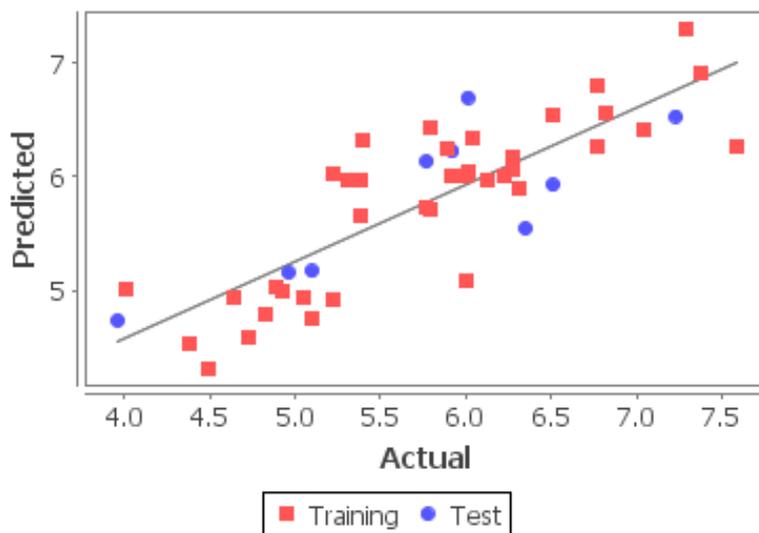
Model 2 (Fig. 3b)



Model 3 (Fig. 3c)

Fig.3- Molecular rectangular field grid displaying the effect of data points (s: steric effect, E: electrostatic effect, H:hydrophobic effect)

The fitness plot and the plot between the predicted and the actual biological activity for the test set and the training set for the equation 1, 2 and 3 are shown in figure 4 and 5 respectively.



Model 1

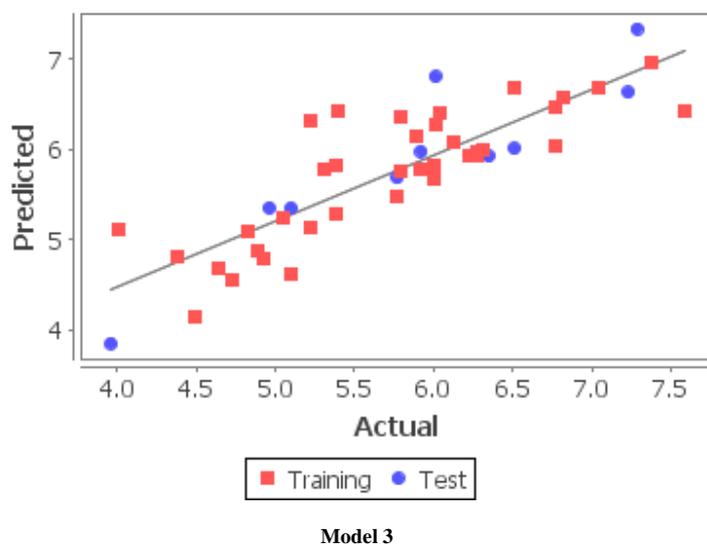
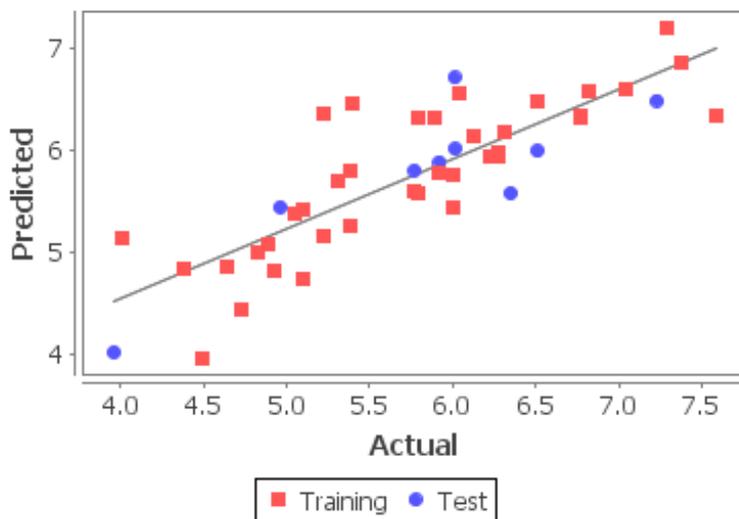


Fig.4- Fitness graphs between the actual and the predicted biological activities for equation 01, 02 and 03 respectively

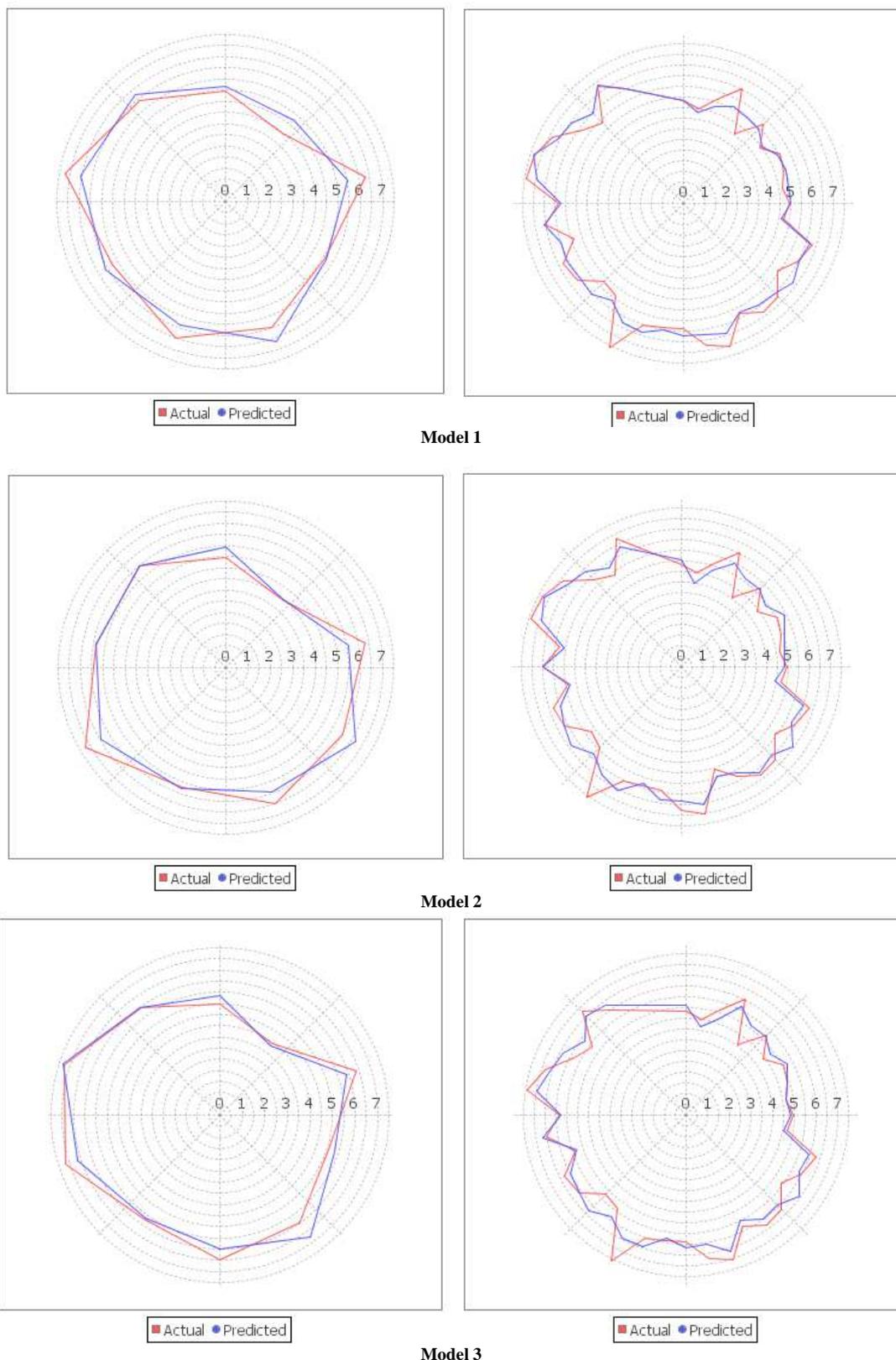


Fig.5- Graph between actual and predicted biological activity for test set and training set

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