



## An Efficient and Alternative Method for Synthesis of Tofacitinib

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### ABSTRACT

An effective synthetic process for the preparation of 3-Benzyl-6-methyl-7-oxa-3-aza-bicyclo[4.1.0]heptanes **7** and 3-Benzyl-6-methyl-3,7-diazabicyclo[4.1.0]heptane-7-carboxylic acid tert-butyl ester **9** has been developed from simple synthetic transformations, starting with extraordinarily simple materials (4-picoline and benzyl chloride) in five steps through reduction and epoxidation followed by ring opening, providing a new and convenient access to the key intermediates of Tofacitinib (CP-690,550). The process developed is highly efficient and can be applied for large scale synthesis also.

**Keywords:** Piperidines, Oxirane, Aziridine, Tofacitinib (CP-690,550).

### INTRODUCTION

The piperidine moiety is one of the privileged structures<sup>1</sup> in medicinal chemistry [1-6]. In particular, 3-aminopiperidines appear quite frequently in pharmaceutically compounds [7-18]. An additional substituent in the 4-position of such compounds can be introduced by nucleophilic ring opening of 3,4-aziridinopiperidines [19,20]. The compound Tofacitinib, (3-((3R,4R)-4-methyl-3-(methyl(7H-pyrrolo[2,3-d]pyrimidin-4-yl)amino)piperidin-1-yl)-3-oxopropanenitrile) is a small molecule designed by Pfizer, also known as CP-690,550 with the chemical structure of substituted piperidine linked to a deazapurine core, as shown in Figure 1. And it is a promising immunosuppressant compound, less toxic than predecessors and very effective, orally active, and described as a JAK3 specific inhibitor, which has demonstrated efficacy against psoriasis, rheumatoid arthritis, Chron's disease, kidney transplant rejection, and ulcerative colitis [21-24]. Due to its highly potent biological activities, many groups have tried to develop an efficient and alternative strategies for the synthesis of Tofacitinib. In this context, we considered orthogonally protected tert-butyl 4-benzyl-1-methyl-4,7-diaza-bicyclo[4.1.0]heptane-7-carboxylate **9** and 4-benzyl-1-methyl-7-oxa-4-aza-bicyclo[4.1.0]heptane-7-ol **14** [25] are an extraordinarily useful building blocks for the preparation of 4-substituted 3-aminopiperidine derivatives. Herein, we wish to report a straightforward synthesis of these racemic compounds **7** and **10**, are the key intermediates for making tofacitinib starting from readily available starting materials.

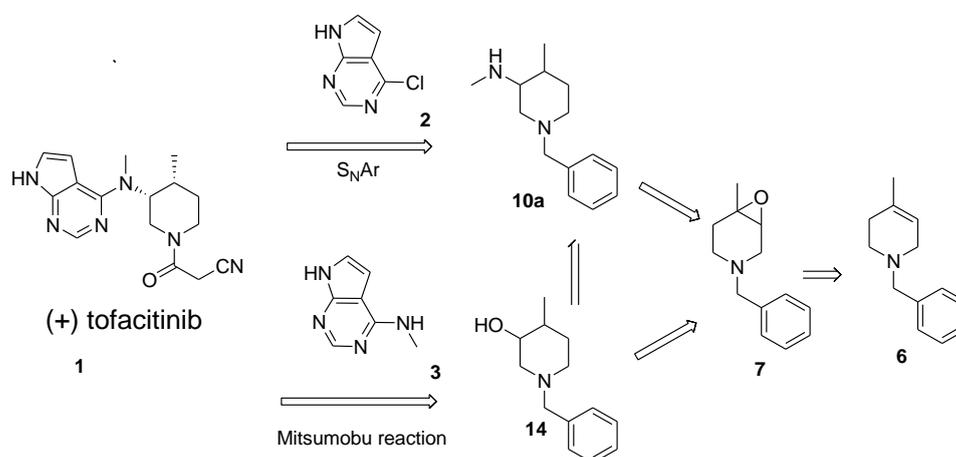
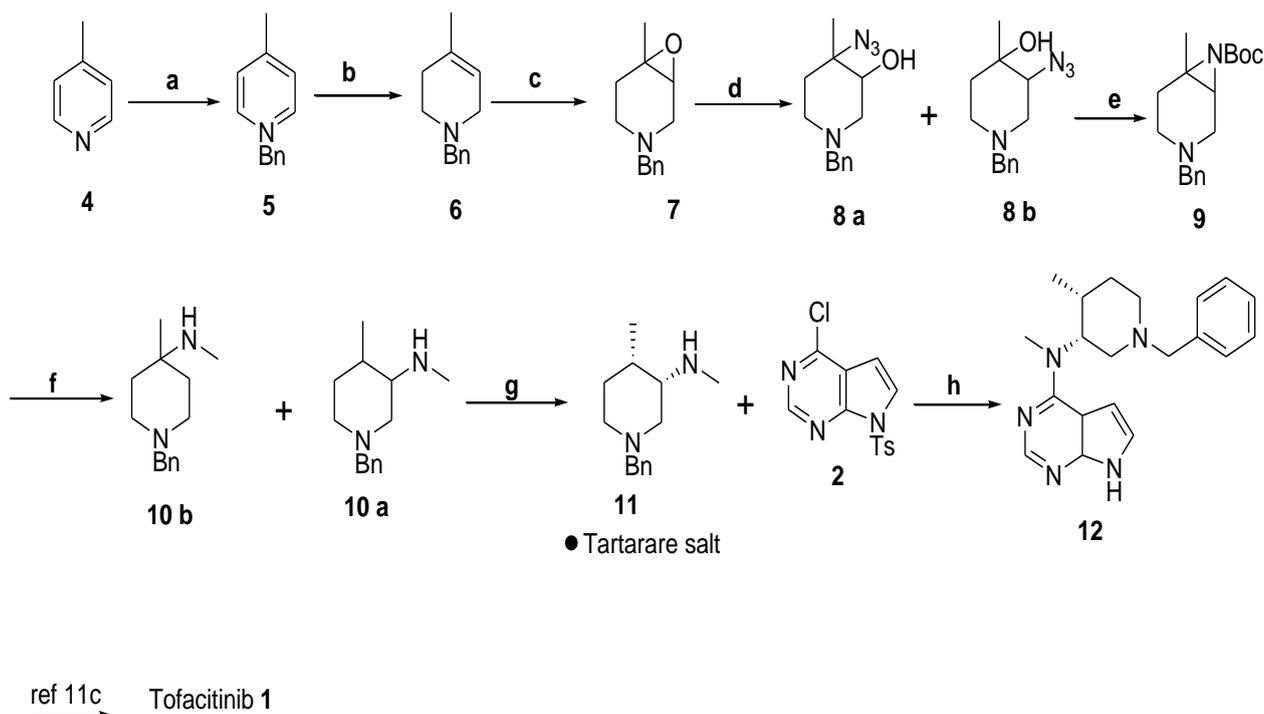


Figure 1: Retrosynthetic strategies for the synthesis of tofacitinib

Although several methods for the synthesis of tofacitinib are known in the literature, asymmetric synthesis is rare [26-28]. Most of reported syntheses require resolution of racemic 3,4-disubstituted piperidines 11 [29,30] or 14 [31], chiral 3,4-disubstituted piperidine 11 is reacted with 6-chloro-7-deazapurine 2 through an  $S_NAr$  mechanism or a Mitsunobu reaction [32,33] of 14 with 6-amino-7-deazapurine 3 to give the advanced intermediates for the synthesis of (+)-tofacitinib (Figure 1).

## RESULTS AND DISCUSSION

Based on its pharmaceutical importance, the synthesis of tofacitinib was extensively studied, and we propose a complete new approach for the synthesis of this new immunosuppressive compound. From a retrosynthetic point of view, orthogonally protected aziridine 9 and 1-Benzyl-4-methyl-piperidin-3-ol [31] 14 can be synthesized from olefin 6 *via* the corresponding epoxide 7 (Figure 1) [25]. Tetrahydropyridine derivative 6 might be prepared by addition of methyl lithium to 1-Benzyl-piperidin-4-one, followed by tertiary chloride formation and elimination provided olefin 6 with 60% yield [31]. Alternatively, an aqueous Diels-Alder reaction produced the material from isoprene and favored route to intermediate 6 [34], however, these routes have some drawbacks like long reaction time, low yields, handling of methyl lithium and isolation of product from diels-alder reaction.



Scheme 1: Reagents and conditions

(a) BnCl (1.0 equiv), 90°C, 4 h, 90%; (b) EtOH: H<sub>2</sub>O, NaBH<sub>4</sub> (3.0 equiv), 25°C-30°C, 15 h; 89.6; (c) i. Trifluoroacetic acid. (3.0 equiv), *m*-CPBA (3.3 equiv), CH<sub>2</sub>Cl<sub>2</sub>, 0°C-23°C, 1.5 h; 86% (d) NaN<sub>3</sub> (2 equiv), acetic acid 25°C-30°C, 18 h; 85% (e) i. Ph<sub>3</sub>P (2.0 equiv), toluene, 100°C, 10 h; ii. Boc<sub>2</sub>O (1 equiv), 22°C-30°C, 2 h, 74%; (f) LiAlH<sub>4</sub> (1.2 eq), THF, 0-5°C, 1 h and 65°C, 5-6 h, 44.4%; (g) L-DTTA, MeOH: H<sub>2</sub>O (1: 1); (h) i. 2K<sub>2</sub>CO<sub>3</sub>, H<sub>2</sub>O reflux; ii. Aq. NaOH solution.

As an alternative to above reactions, we envisioned the reduction of 4-picoline to give 4-methyl- tetrahydropyridine, probably the most inexpensive approach to this target is the solvent free benzylation of 4-picoline itself and subsequent treatment of the benzylpyridinium chloride with sodium borohydride in protic solvents. This reaction was previously reported in the literature to give 73% yield [35]. We were able to isolate compound 6 in 90% yield. Attempts at subsequent epoxidation of tertiary amine 6 were unfortunately not chemoselective. The conversion was practically quantitative, but the purification of epoxide 7 became very difficult because of its instability, gave the respective amine-*N*-oxide, 1-benzyl-4-methyl 3,4-dihydroxypiperidine as major products after chromatography over both silica gel and alumina. By careful adjustment of reaction condition and detailed chromatographic monitoring, In the overall yield of 15% were obtained the target epoxide [25] using trifluoroacetic acid, trifluoroacetic anhydride and H<sub>2</sub>O<sub>2</sub> in dichloromethane. All the above results indicated that the necessity for the synthetic conditions and appropriate epoxidizing agent to successful preparation of 1-benzyl-4-methyl-3,4-epoxy piperidine. We turned our attention to the exploration of epoxidation (Table 1) protocol by the addition of different oxidants, acids and solvents. Among all the protocol condition tertiary amine 6 could be converted into trifluoroacetic acid salt with excess TFA at 25°C-30°C in dry dichloromethane, Subsequent epoxidation with *m*-chloro-peroxybenzoic acid addition at 0°C-5°C portion wise gave oxirane 7 with high selectivity and in almost quantitative yield (85%-90%) after 2-3 h at 25°C-30°C. The crude material was pure enough to be submitted to the next step without purification. The ring opening with sodium azide was achieved in acetic acid water, and yielded a 93: 07 mixture of two regioisomers 8a and 8b, respectively, in an overall yield of 85%. These isomers could, but did not need to be separated by column chromatography. Staudinger reaction of this mixture of azides 8a and 8b was performed with triphenylphosphane in absolute toluene at 100°C for 10 h. As reported in the literature for the respective cyclohexane derivative [36], activation of the hydroxy group as recommended by others [37], was not necessary, since it directly reacts with the iminophosphorane to form the aziridine and triphenylphosphane oxide (Ph<sub>3</sub>P=O). It turned out that the parent aziridine (without Boc protection) was difficult to isolate and, moreover, not a stable compound under ambient conditions (we presume decomposition by oligomerization). Therefore, the crude reaction mixture was directly treated with a stoichiometric amount of di-*tert*-butyl dicarbonate (Boc<sub>2</sub>O), which resulted in quantitative Boc-protection of the aziridine within two hours at 25°C-30°C. Compound 9 was isolated in 73% after chromatographic purification, as material with seemingly unlimited storability; we were not able to detect any decomposition even at elevated temperature and in solution.

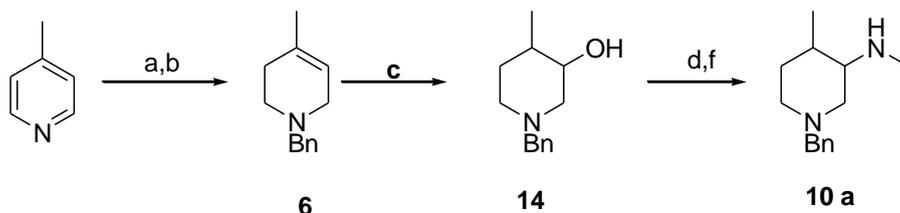
Table 1: Optimization of epoxydation with peracids<sup>a</sup>

S. No.	Solvent	Acid	Peracid	Time (h)	Temp (°C)	Yield (%) <sup>b</sup>
1	Dichloromethane	TFA	H <sub>2</sub> O <sub>2</sub>	12	0-30	No product
	Dichloromethane	AcOH	Ac <sub>2</sub> O/H <sub>2</sub> O <sub>2</sub>	12	0-30	No product
2	Dichloromethane	TFA	TFAA/H <sub>2</sub> O <sub>2</sub>	2	<10	25%-30%
3	Dichloromethane	TFA	Oxone	3-Feb	0-30	20-25
4	Dichloromethane	TFA	m-CPBA	3-Feb	0-30	85%-90%
5	Dichloromethane	Formic acid	m-CPBA	3-Feb	0-30	30-35

<sup>a</sup>Reaction conditions: All the reaction were carried out 1-Benzyl-4-methyl-1,2,3,6-tetrahydro-pyridine (1 mmol), Trifluoroacetic acid (3.0 mmol), peracid (2.5 mmol) dichloromethane (50 ml) at room temperatures; <sup>b</sup> Isolated yields

In comparison to the huge number of reports on the ring opening of aziridines by other nucleophiles [38,39], their ring opening by hydrides has received very limited interest in literature despite the synthetic potential of this approach. The reduction of Boc aziridine 9 with LiAlH<sub>4</sub> in THF at 0°C-10°C for 1 h, reaction continued for 5-6 h at reflux and yielded a mixture (30: 70) of ring opened methyamines derivatives (derived from hydride attack at both the more and the less hindered carbon atom in 2: 1 ratio), 10a and 10b respectively, in an overall yield of 62%. These isomers were separated by column chromatography.

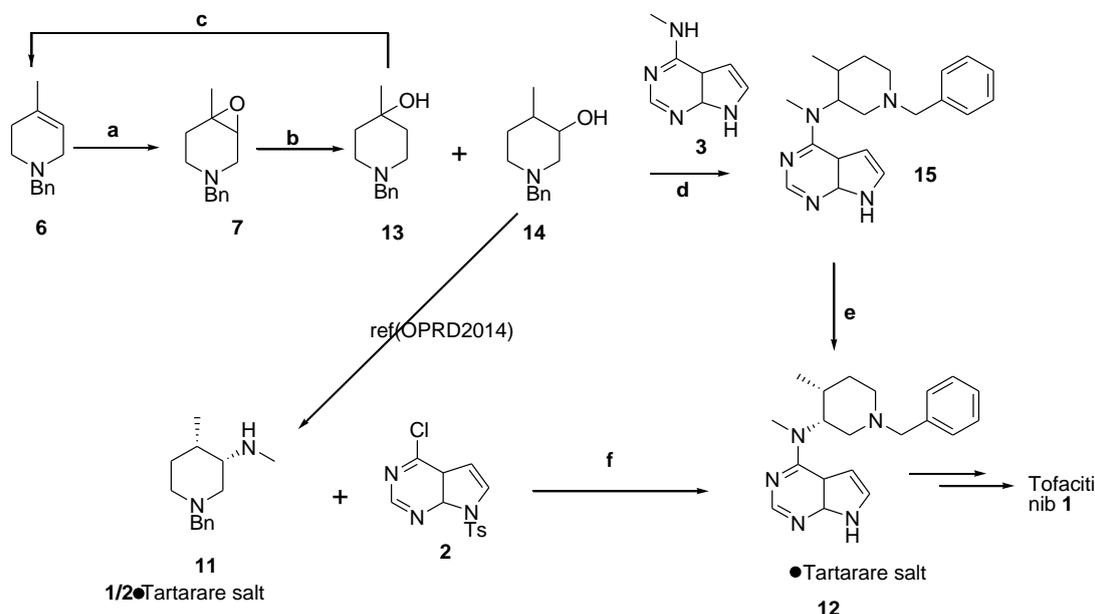
After chiral resolution of compound 10a with di-para tolyltartarate, optically pure compound 11 was successfully obtained by recrystallisation in methanol-water with 43% yield. As shown in Scheme 1 the nucleophilic substitution of commercially available 4-chloro-7-tosyl-7H-pyrazolo[2,3-d]pyrimidine compound with compound 11 underwent very smoothly in shorter reaction time and an improved yeield. Without tosylprotection, nucleophilic substitution reaction was very slow and low yielded. This indicated that tosyl group effectively reduced the electron density of pyrrolopyrimidine ring by its strong electron-withdrawing and conjugative effect. Subsequently, the tosyl group was hydrolyzed in aqueous solution of sodium hydroxide to afford compound 3 in a good yield. Thus, a combination of nucleophilic substitution and hydrolysis was successfully carried out in one pot under alkaline condition.



Scheme 2: Pfizer route for synthesis of 1-benzyl-N,4-dimethylpiperidin-3-amine

Scheme 2 Reagents and conditions: (a) BnCl, acetone, 73%; (b) NaBH<sub>4</sub>, EtOH, 73%; (c) (i) BF<sub>3</sub>·OEt<sub>2</sub>, (ii) BH<sub>3</sub>, (iii) MeOH, CaCl<sub>2</sub>, H<sub>2</sub>O, (iv) H<sub>2</sub>O<sub>2</sub>, (v) NaOH, (vi) TsOH, 88%; (d) SO<sub>3</sub>·py, DMSO, Et<sub>3</sub>N; (e) MeNH<sub>2</sub>, NaBH(OAc)<sub>3</sub>; (f) HCl, 53% (3 steps).

We proposed an alternative strategy based on the precursor 3-hydroxy-4-methylpiperidine 14 (Scheme 3) which has been developed including two from Pfizer [29], and another by A. Marican et al. [26-28]. Pfizer's route started with the benzylation of 4-pipecoline followed by reduction with sodium borohydride to afford 6, which underwent a hydroboration followed by oxidation to yield hydroxy piperidine 14 (Scheme 2). Unfortunately, the extreme reaction conditions required, and potentially hazardous nature of the reagents and releases toxic hydrogen fluoride makes these preparations unattractive for operation on a manufacturing scale and A. Marican et al. route started with chiral 5-Hydroxy-piperidin-2-one and it was too long to industrialize and employed costly raw materials and air sensitive reagents which gave a poor product yield.



Scheme 3: Strategy based on the precursor 3-hydroxy-4-methylpiperidine 14

Scheme 3 Reagents and conditions: (a) i. Trifluoroacetic acid (3.0 equiv), m-CPBA (3.3 equiv), CH<sub>2</sub>Cl<sub>2</sub>, 0-23°C, 1.5 h; 90% (b) i. LiAlH<sub>4</sub> (1.1 equiv), THF 0-25°C, 1 h; 45 % (c) i. SOCl<sub>2</sub>, DCM, 25-30°C, 15 h; ii. KOH, EtOH; 40% (d) DIAD (1.1 equiv), Ph<sub>3</sub>P (1.1 equiv), 1,4-dioxane, 100°C, 3 h; 56% (e) L-DTTA, H<sub>2</sub>O, IPA, MeOH (f) i. K<sub>2</sub>CO<sub>3</sub>, H<sub>2</sub>O reflux; ii. Aq. NaOH solution.

According to our retrosynthetic analysis applied to the target CP-690,550 (Figure 1), compound 14 was prepared *via* selective reduction of epoxide compound 7, after several parameters were evaluated in an effort to optimize the selectivity and reproducibility of the reduction of epoxide 7 with different reducing agent as shown in Table 2. However, we found that sodium borohydride and other reducing agents were not suitable for this conversion. Lithium aluminum hydride was chosen as the reducing agent as it afforded a high yield of compound 14 and the reaction was simple to perform and work up. Finally compound 14 was obtained with Lithium Aluminium Hydride (LiAlH<sub>4</sub>) in THF at 0-5°C for 1 h, the regeoisomer 13 and 14 was separated by column chromatography with 40-45% yield, And the compound 13 converted to compound 6 with 45% yield as described in the literature [31].

**Table 2: Optimization of reducing agents for the synthesis of 1-Benzyl-4-methyl-piperidin-3-ol<sup>a</sup>**

S. No.	Solvent	Reagent	Temp (°C)	Time (h)	Yield (%)	
					13	14
1	THF	NaCNBH <sub>4</sub>	25-30	12	-	-
2	THF	NaBH <sub>4</sub>	25-30	12	-	-
3	THF	LiAlH <sub>4</sub>	0-5	1	40	41
4	THF	LiAlH <sub>4</sub>	25-30	<0.5	45	40
5	Ether	LiAlH <sub>4</sub>	0-5	1	42	45

<sup>a</sup>Reaction conditions: All the reactions were carried out 3-Benzyl-6-methyl-7-oxa-3-aza-bicyclo[4.1.0]heptanes (1.0 mmol.), Reducing agent(1.1 mmol.), and THF (10 ml) at room temperatures; <sup>b</sup> Isolated yields

The reaction between 14 and commercially available Methyl-(7H-pyrrolo[2,3-d]pyrimidin-4-yl)-amine proceeded under Mitsunobu conditions similar to those described in the literature [26-28]. But required a shorter time and gave an acceptable yield of advanced tofacitinib intermediate 15, which was resolved with di-para tolyltartarate as described in the literature [40].

## CONCLUSION

We have developed an alternate route for the synthesis of 10a and 14, which are important intermediate for the synthesis of tofacitinib from N-benzyl 3,4-epoxypiperidine, this synthetic strategy provides reliable new entries for the synthesis of tofacitinib, as well as synthesis of piperidine based bioactive compounds. The key reaction was the reductive ring opening of Boc activated aziridine as well as epoxide utilizing LiAlH<sub>4</sub> has been reported for the first time in high regioselective way and it allows to exclude an operation with a toxic hydrogenfluoride and oxidation to make key intermediate of tofacitinib.

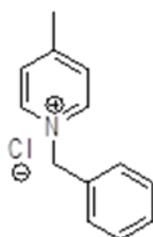
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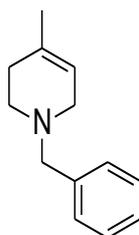
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**General Information**

All reagents were used as received from commercial sources without further purification or prepared as described in the literature. Reactions were stirred using Teflon-coated magnetic stirring bars. TLC plates were visualized by ultraviolet light. Chromatographic purification of products was carried out by flash column chromatography on silica gel (100-200 mesh). Melting points were determined using an electro thermal melting point apparatus and are uncorrected. NMR spectra were measured in  $\text{CDCl}_3$  and  $\text{DMSO-d}_6$  (all with TMS as internal standard) on a Varian Gemini 400 MHz FT NMR spectrometer magnetic resonance spectrometer. Chemical shifts ( $\delta$ ) are reported in ppm, and coupling constants ( $J$ ) are in Hz. The following abbreviations were used to explain the multiplicities: s=singlet, d=doublet, t=triplet, q=quartet, m=multiplet. LCMS were recorded on an Agilent HP-5989A quadrupole mass spectrometer.

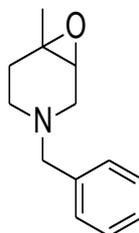
**Experimental Procedures and Analytical data***Synthesis of benzyl pyridinium chloride (5)*

To a stirring solution of 4-methyl Pyridine **4** (25.0 g, 268.81 mmol, 1.0 eq.) cooled to 0-5°C and benzyl chloride (36.96 g, 261.81 mmol, 1.0 eq.) was added slowly drop wise, after complete addition, stirring was continued at 90°C for 3-4 h. After completion of reaction on TLC, reaction mass brought to rt, stirred for 1 hour, filtered the precipitated white crystalline solid (53.0 g, 242.00 mmol, 90.0%).

*Synthesis of 1-benzyl-1,2,3,6-tetrahydro-4-methylpyridineride(6)*

Procedure: To a stirring solution of benzyl pyridinium chloride **5** (30.0 g, 136.98 mmol, 1.0 eq.) in a EtOH: H<sub>2</sub>O (9: 1) (300 ml) cooled to 0-10°C and NaBH<sub>4</sub> (15.61 g, 410.95 mmol, 3.0 eq.) was added Portion wise, after complete addition, stirring was continued at RT for 16 h. reaction was monitored by TLC, After completion of reaction diluted with water and extracted with Ethyl acetate (2 × 250 ml), combined organic layer was washed with brine solution, dried over sodium sulphate, the solvent was evaporated under reduced pressure to get desired compound as pale yellow liquid which was purified by column chromatography using 100-200 silica mesh with 10% ethyl acetate in hexane (23.0 g, 122.90 mmol, 89.9%).

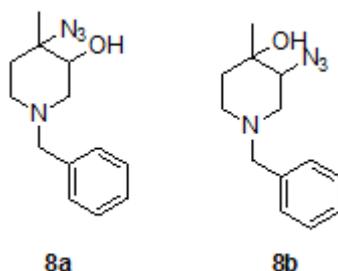
For compound 6: <sup>1</sup>H-NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm)=7.35-7.33 (m, 5H), 5.36-5.34 (m, 1H), 3.56 (s, 3H), 2.92 (t, J=3.2 Hz, 2H), 2.54 (t, J=5.6 Hz, 2H), 2.06 (d, J=7.6 Hz, 2H), 1.67 (s, 3H). LC-MS: tRet=0.608 min, M+1: 188.2.

*Synthesis of 4-benzyl-1-methyl-7-oxa-4-aza-bicyclo [4.1.0] heptane (7)*

Procedure: To stirred solution of 1-benzyl-1,2,3,4-tetrahydro-4-methylpyridine (30.0 g, 160.42 mmol), in dichloromethane, was added trifluoroacetic acid (30.89 ml, 401.06, 2.5 eq) at 05°C and stirred for 30 min same temperature and added m-CPBA (82.8 g, 481.28 mmol, 3.0 eq, 77% purity)portion wise at same temperature and reaction continued for 2-3 h at 25-30°C.Reaction progress was monitired by TLC, after completion of reaction, diluted with cooled water, excess m-CPBA quenched with saturated sodium sulfite, Organic layer was separated and washed with sat. NaHCO<sub>3</sub> followed by brine solution, dried over sodium sulphate, filtered the solvent, evaporated under reduced pressure to get crude compound which was purified by column chromatography using 100-200 silica mesh with 10% ethyl acetate in hexane as a colorless liquid. (28.2 g, 203.50 mmol, 86%).

Compound 7 : <sup>1</sup>H-NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm)=7.3-7.24 (m, 5H), 3.04 (d, J=4.1 Hz, 1H), 3.01 (d J=9.2 Hz, 1H), 2.41 (d, J=12.8 Hz, 1H), 2.32 (d, J=10.0 Hz, 1H), 2.01 (s(broad), 1H), 1.83(d, J=10.0 Hz, 2H), 1.26 (s, 3H); <sup>13</sup>C-NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  (ppm)=20.25, 52.57, 57.85, 62.29, 68.25, 122.58, 127.11, 128.21, 128.90, 134.97, 137.82. LC-MS: tRet=2.187; min, M+1: 204.2.

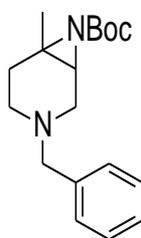
Synthesis of *trans* 4-azido-1-benzyl-4-methylpiperidin-3-ol (8a) and *trans* 3-azido-1-benzyl-4-methylpiperidin-4-ol (8b)



To a stirred solution of mixture of epoxide 8 (25.0 g, 122.54 mmol, 1.0 eqt.), in water acetic acid (6: 4) 250 ml, to this  $\text{NaNO}_3$  (22.5 g, 367.62 mmol 3.0 eqt.) was added at RT and stirred for overnight at same temperature. Reaction was monitored by TLC. After completion of reaction, neutralized with Sodium bicarbonate solution, and extracted with ethyl acetate ( $2 \times 200$  ml), combined organic layer was washed with brine solution, dried over sodium sulphate, the solvent was evaporated under reduced pressure and the residue was purified by column chromatography using 100-200 silica mesh with ethyl acetate in hexane to give a mixture of both regioisomers (11a/11b = 96: 04 by LCMS) as a pale yellow liquid. The isomers could be separated by repeated chromatography. Compound 8a was eluted at 20% EtOAc in hexane (18.2 g, 73.68 mmol, 60.6%) and compound 8b was eluted 30% EtOAc in hexane (2.4 g, 9.716 mmol, 8.0%).

For compound 8a:  $^1\text{H-NMR}$  (400 MHz,  $\text{CDCl}_3$ ): 7.33-7.34 (m, 5H), 3.52 (s, 2H), 3.36(s, 1H), 2.66-2.56 (m, 3H), 2.32-2.26 (m, 1H), 2.87-2.81 (m, 1H), 1.60-1.57 (m, 1H), 1.38 (s, 3H). LC-MS:  $t_{\text{Ret}}=1.62$  min,  $M+1$ : 247.2.

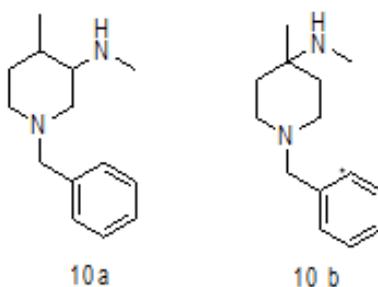
Synthesis of *tert*-butyl 4-benzyl-1-methyl-4,7-diaza-icyclo[4.1.0]heptane-7-carboxylate(9)



A mixture of both regioisomers of azide 8a/8b (20.00 g, 82.3 mmol 1.0 eqt.),  $\text{PPh}_3$  (62.0 g, 164.50 mmol 2.0 eqt.), and toluene (200 ml) was stirred for 12 h at  $100^\circ\text{C}$  under an  $\text{N}_2$ -atmosphere. Reaction was monitored by TLC. The mixture was cooled to  $25-30^\circ\text{C}$  and a solution of Boc<sub>2</sub>O (26.9 g, 123.45 mmol 1.5 eqt), was added. After stirring the mixture for 2 h at  $25-30^\circ\text{C}$ , the solvent was removed under vacuum and the residue was purified by column chromatography using 100-200 silica mesh with 20% ethyl acetate in hexane as a colorless liquid, (18.0 g, 59.40 mmol, 74%).

For Compound 9:  $^1\text{H-NMR}$  (400 MHz,  $\text{CDCl}_3$ ): 7.32-7.23 (m, 5H), 3.40 (s, 2H), 2.96-2.92 (m, 1H), 2.50-2.42 (m, 1H), 2.25-2.22 (m, 1H), 2.05-2.03 (m, 1H), 1.86-1.82 (m, 1H), 1.69-1.64 (m, 1H), 1.38 (s, 9H), 1.22 (s, 3H);  $^{13}\text{C-NMR}$  (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  (ppm)=25.20, 25.5, 35.4, 37.7, 45.5, 46.5, 57.85, 62.29, 78.25, 127.11, 128.21, 128.90, 134.97, 137.82. LC-MS:  $t_{\text{Ret}}=1.75$  min,  $M+1$ : 303.2.

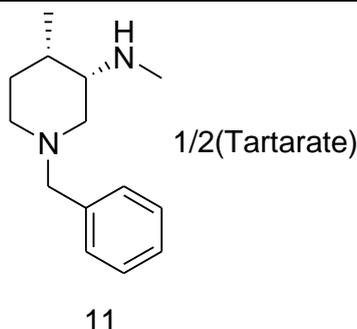
Synthesis of 1-benzyl-N,4-dimethylpiperidin-3-amine (10a)



To a suspension  $\text{LiAlH}_4$  (3.76 g, 99.00 mmol) in THF (100.0 ml) was added 9 (10.0 g, 33.00 mmol) in THF (20.0 ml) at  $0-5^\circ\text{C}$ . The mixture was stirred for 1 h at room temperature and then refluxed for 4-5. After completion of reaction monitored by TLC, reaction mass brought to  $0-5^\circ\text{C}$ , the mixture was quenched with ethyl acetate followed by sat.  $\text{Na}_2\text{SO}_4$  and filtered. The filtrate was diluted with water (100 ml) and extracted with EtOAc ( $3 \times 100$  ml). The combined organic layers were washed with brine and dried over  $\text{Na}_2\text{SO}_4$ . The solvent was filtered and concentrated under reduced pressure to get crude regioisomers, which were purified by column chromatography using 100-200 silica mesh using Methanol in dichloromethane to give desired products 10a as colorless oil (3.2 g, 19.26 mmol, 44.4%) and 10b as colorless liquid (1.2 g, 16.6%), compound 10 a was directly used in chiral resolution.

For compound 10a:  $^1\text{H-NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm)=7.20-7.28 (m, 5H), 5.25 (d,  $J=8.8$  Hz, 1H), 3.70 (d,  $J=9.2$  Hz, 1H), 3.52-3.60 (m, 1H), 3.43 (ABq,  $J=13.2$  Hz, 2H), 2.76 (d,  $J=10.4$ , 1H), 2.12 (d,  $J=10.8$  Hz, 1H), 1.90 (td,  $J=11.2, 3.2$  Hz, 1H), 1.48-1.64 (m, 1H), 1.42 (s, 9H), 1.25-1.40 (m, 2H), 0.90 (d,  $J=8.4$  Hz, 3H). LC-MS:  $t_{\text{Ret}}=1.75$  min,  $M+1$ : 218.34.

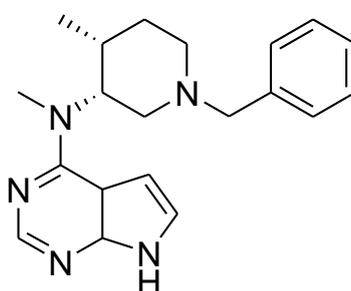
Synthesis of (3*S*, 4*S*)-1-benzyl-N,4-dimethylpiperidine-3-amine. Tartarate salt (11)



Compound 10a (3.2 g, 14.67 mmol) was added in the mixture of methanol (5.0 ml) and isopropyl alcohol (20.0 ml), followed by the addition of water (20.0 ml) and di-p-toluoyl-L-tartarate (2.8 g, 7.33 mmol). The reaction mixture was heated to reflux until homogeneous. At this temperature reaction stirred for 1 h, then slowly reaction brought to rt and filtered the target compound 11 (2.2 g, mmol, 50.2%) as white solid.

For compound 13a:  $^1\text{H-NMR}$  (400 MHz,  $\text{CD}_3\text{OD}$ )  $\delta$  (ppm)=8.04 (d,  $J=8.0$  Hz, d), 7.20-7.28 (m, 7H), 5.85 (s, 1H), 3.70 (d,  $J=12.7$  Hz, 1H), 3.42 (d,  $J=12.7$  Hz, 1H), 3.09 (s, 1H), 2.91 (d,  $J=23.4$  Hz, 2H), 2.49 (s, 1H), 2.22 (d,  $J=18.3$ , 2H), 1.91 (q,  $J=3.4$  Hz, 1H), 1.47-1.62 (m, 2H), 1.02 (d,  $J=7.1$  Hz, 3H).

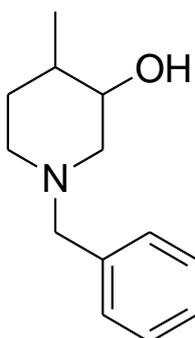
Synthesis of *N-((3R,4R)-1-benzyl-4-methylpiperidin-3-yl)-7,7a-dihydro-N-methyl-4aH-pyrrolo[2,3-d]pyrimidin-4-amine (12)*



To a stirred solution of 11 (2.0 g, 3.31 mmol), in 20 ml of water, was added potassium carbonate (1.37 g, 9.93 mmol), and 4-chloro-7-tosyl-7H-pyrrolo[2,3-d]pyrimidine 2 (2.04 g, 6.62 mmol). The reaction mixture was stirred at reflux for 10 h. After completion of reaction monitored by TLC. Then, the reaction mixture was cooled to room temperature, saturated NaOH solution (25 ml) was added at below  $30^\circ\text{C}$ . The resulting reaction mixture was refluxed for 6 h. After being cooled to room temperature, the reaction mixture was stirred for 2 h at room temperature. The solid was isolated by filtration to afford off-white solid 12 (0.9 g, 81.0%).

For compound 12:  $[\alpha]_D^{25}=+29.4$  (c 1.01, MeOH); purity: 99.8% (C18 HPLC); 100% (chiralpak ID, 95:5 hexane/ isopropanol, retention time of 16.86 min);  $^1\text{H-NMR}$  (400 MHz,  $\text{DMSO-d}_6$ )  $\delta$  (ppm)=0.88 (d,  $J=6.8$  Hz, 3H), 1.59-1.63 (m, 1H), 1.70 (s, 1H), 2.13 (s, 1H), 2.28 (s, 1H), 2.54 (dd,  $J_1=4.0$  Hz,  $J_2=11.2$  Hz, 1H), 2.61 (s, 1H), 2.77 (dd,  $J_1=6.0$  Hz,  $J_2=11.2$  Hz, 1H), 3.44-3.52 (m, 5H), 5.09 (s, 1H), 6.53 (s, 1H), 7.08 (t,  $J=2.6$  Hz, 1H), 7.20-7.23 (m, 1H), 7.28-7.31 (m, 4H), 8.05 (s, 1H), 11.55 (s, 1H).

Synthesis of *1-benzyl-4-methylpiperidin-3-ol (14)*

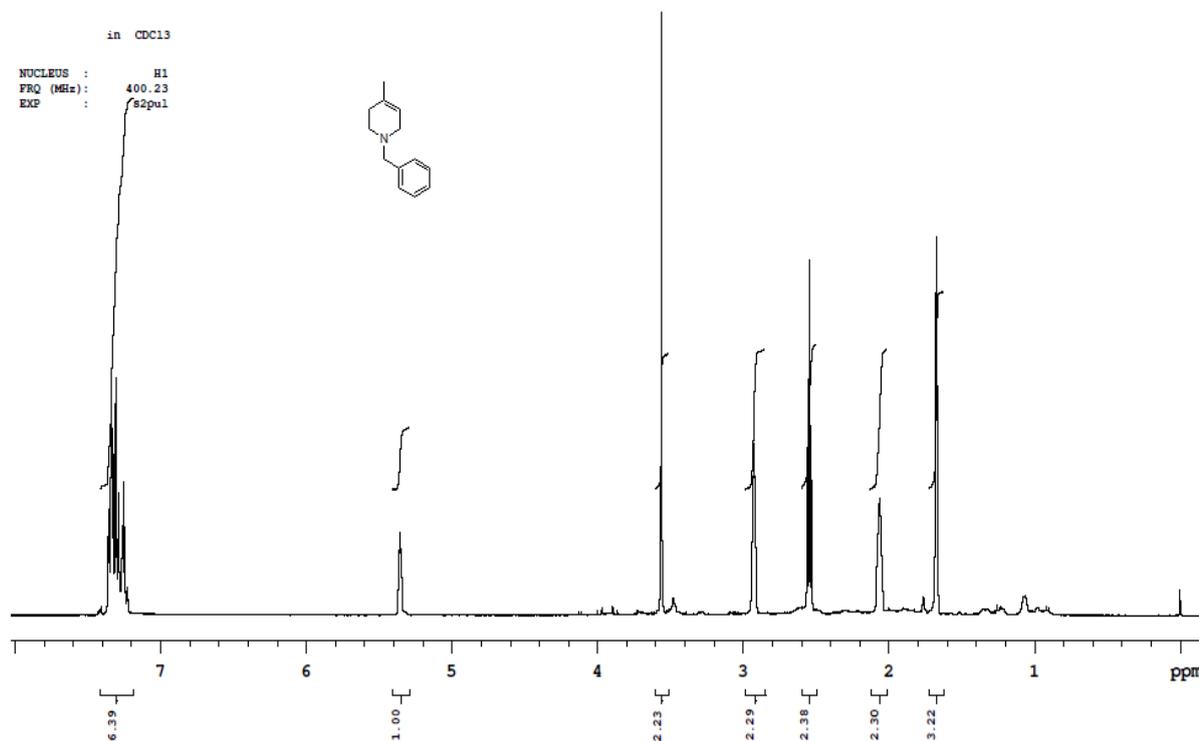


To a suspension  $\text{LiAlH}_4$  (1.87 g, 49.261 mmol) in THF (100.0 ml) was added 7 (10.0 g, 49.261 mmol) in THF (20.0 ml) at  $0-5^\circ\text{C}$ . The mixture was stirred for 1 h at same temperature. After completion of reaction monitored by TLC, the mixture was quenched with ethyl acetate followed by sat.  $\text{Na}_2\text{SO}_4$  paste at same temperature and filtered. The filtrate was diluted with water (100 ml) and extracted with EtOAc ( $3 \times 100$  ml). The combined organic layers were washed with brine and dried over  $\text{Na}_2\text{SO}_4$ . The solvent was filtered and concentrated under reduced pressure to get crude regioisomers, which were purified by column chromatography using 100-200 silica mesh using ethyl acetate and hexane to give desired products 14a as colorless oil (3.8 g, 18.536 mmol, 37.6%) and 10b as colorless liquid (4.1 g, 20.00 mmol, 40.6%).

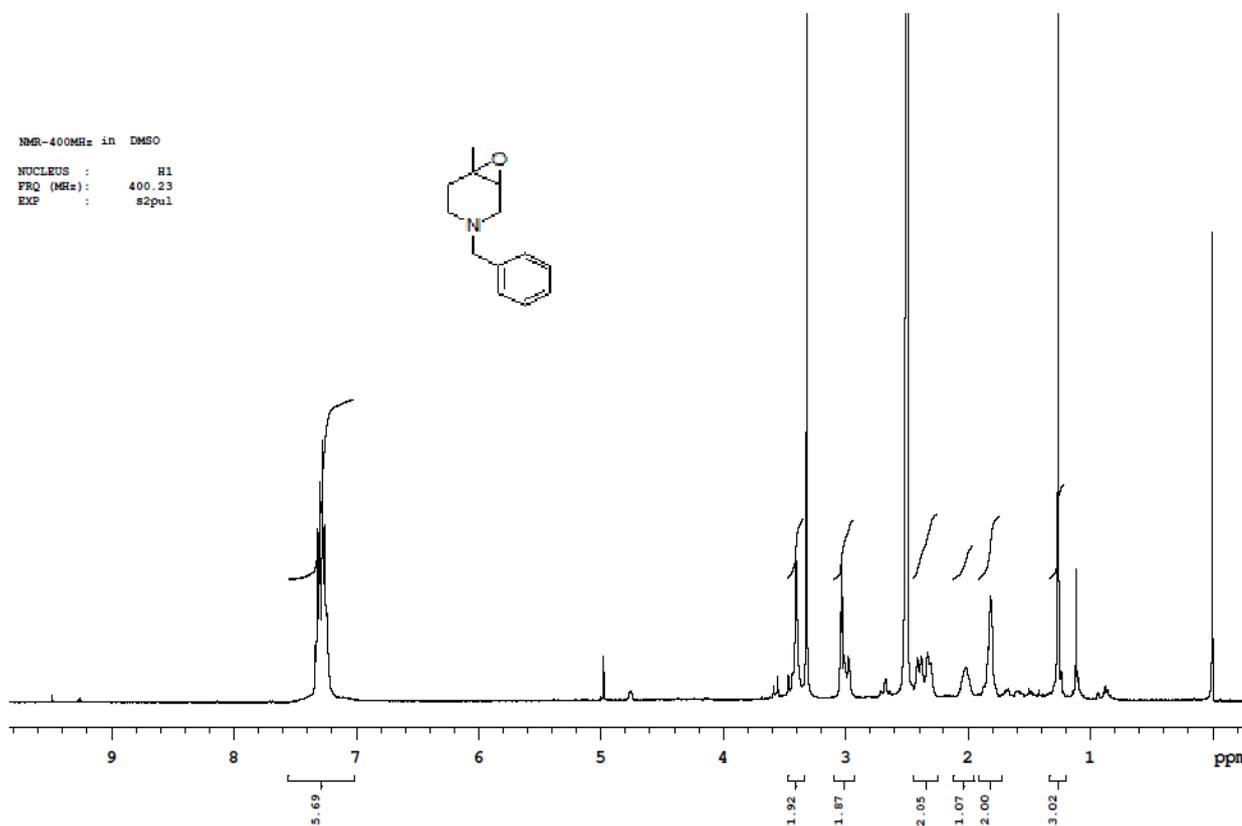
For compound 14:  $^1\text{H-NMR}$  (400 MHz,  $\text{CDCl}_3$ ): 7.37-7.22 (m, 5H), 3.56 (s, 1H bro), 3.50 (s, 2H), 2.96-2.81 (m, 1H), 2.72-2.62 (m, 1H), 2.60-2.16 (m, 1H), 2.13 (m, 1H), 1.54-1.40 (m, 3H), 1.01 (s,  $J=6.0$  Hz, 3H);  $^{13}\text{C-NMR}$  (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  (ppm)=17.81, 28.37, 34.86, 53.29, 60.06, 60.06, 65.66, 69.35, 77.09, 127.13, 128.21, 128.90, 132.99, 138.28. LC-MS: tRet=0.92 min, M+1: 206.1.

$^1\text{H}$ ,  $^{13}\text{C}$ -NMR and LCMS spectra

<sup>1</sup>H-NMR and LCMS of Compound 5

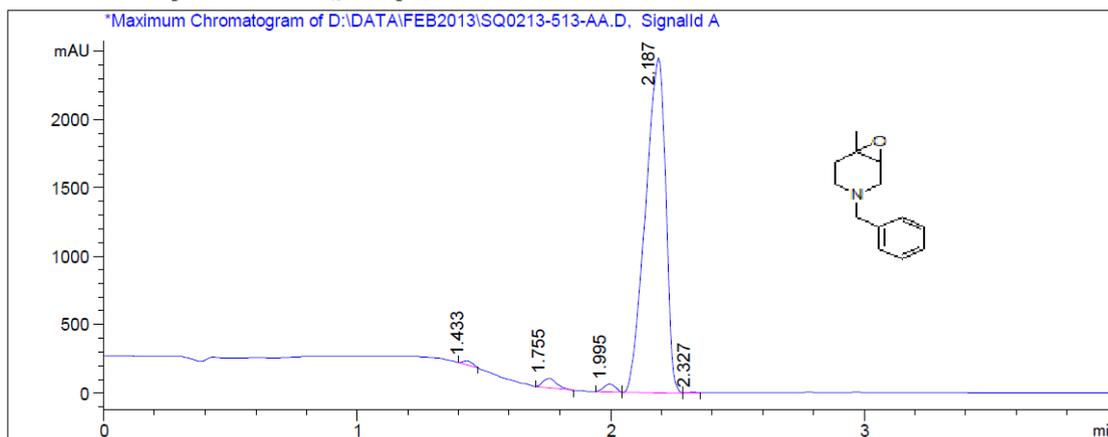


CMS Compound 6



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 Acq. Method : D:\METHODS\KINETEX\_AA1.M

Column: Kinetex C-18, 4.6x30mm, 2.6 um, 100 A  
 Mobile Phase A: 5mM Ammonium Acetate in Aq B: ACN  
 T/%B : 0/5, 1.5/90, 4/90, 4.1/5  
 Flow rate: 1.0ml/min  
 Instrument: Agilent 6120 Qudrapole LCMS

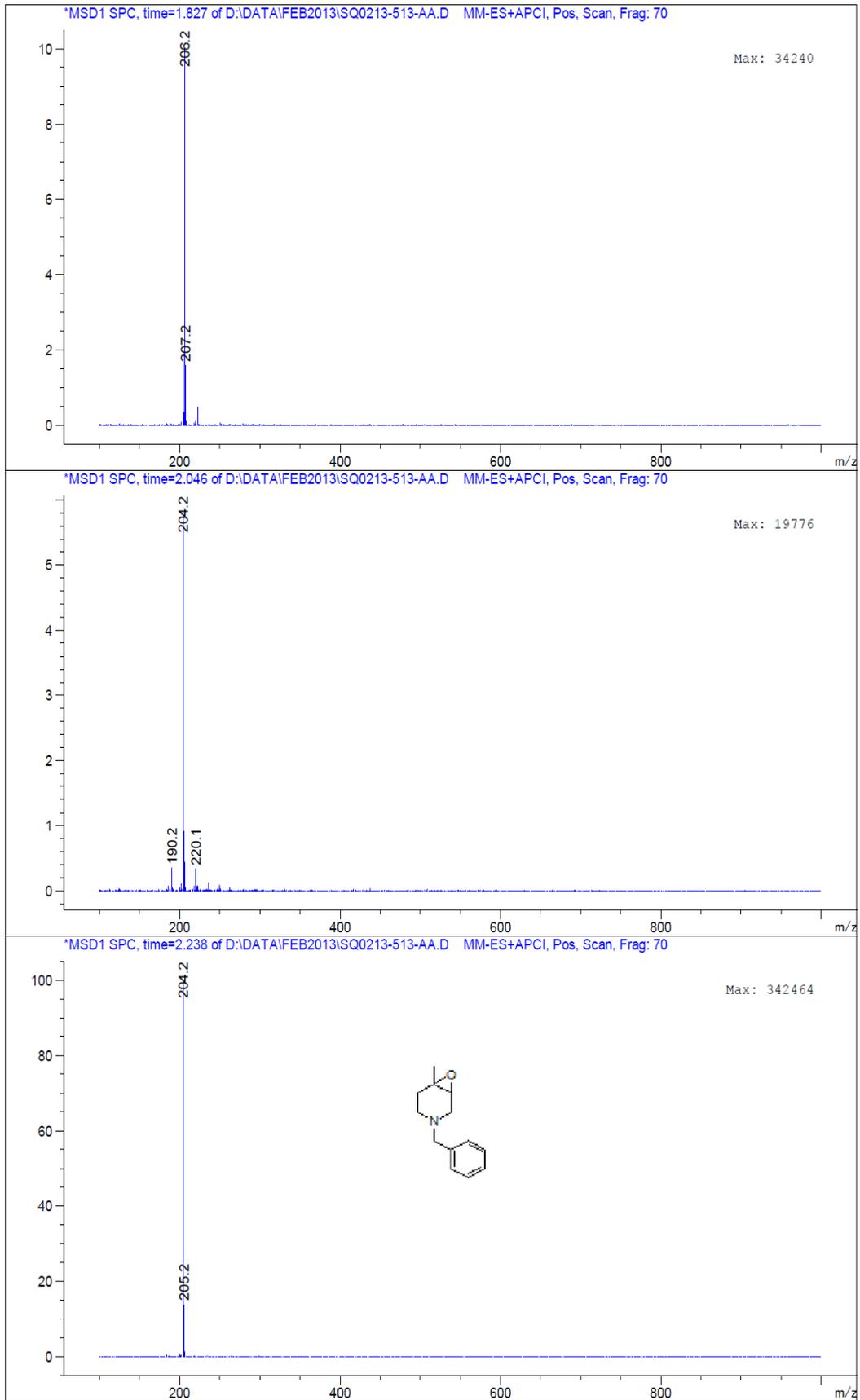


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2	1.755	2.665e+002	1.889
3	1.995	1.816e+002	1.287
4	2.187	1.356e+004	96.100
5	2.327	2.625e+001	0.186

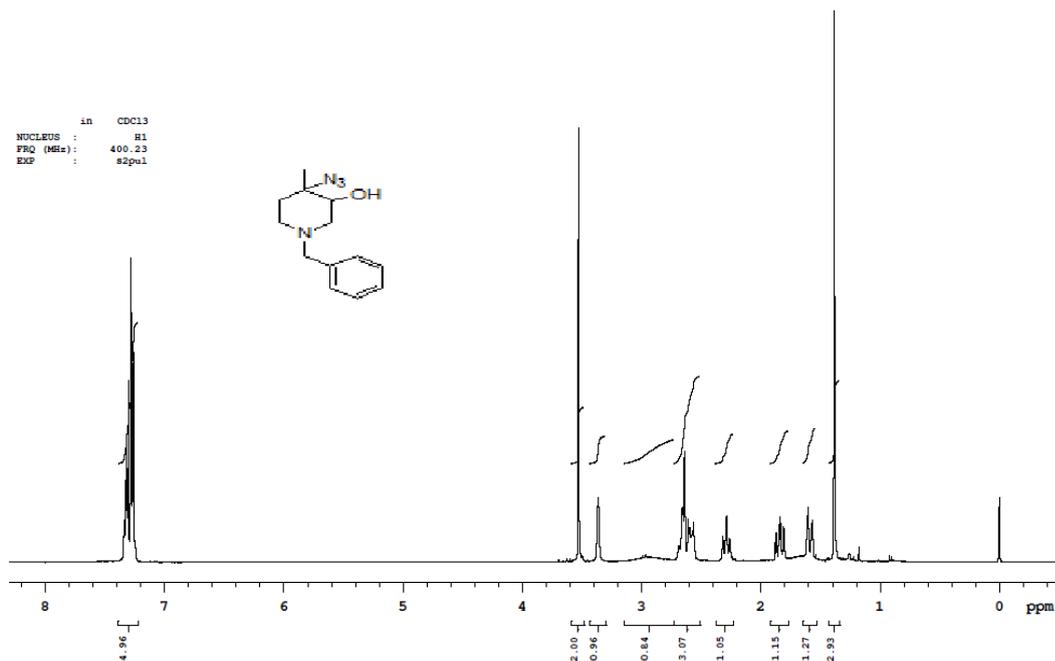
Analysed by :

Print of window 80: MS Spectrum

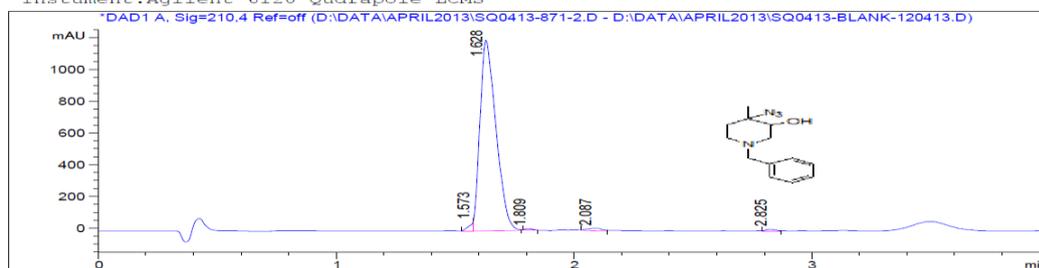
MS Spectrum



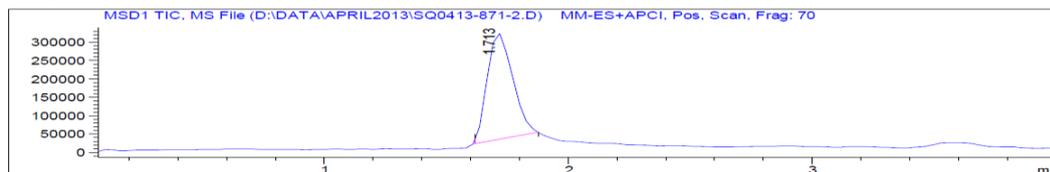
<sup>1</sup>H-NMR and LCMS of Compound 7



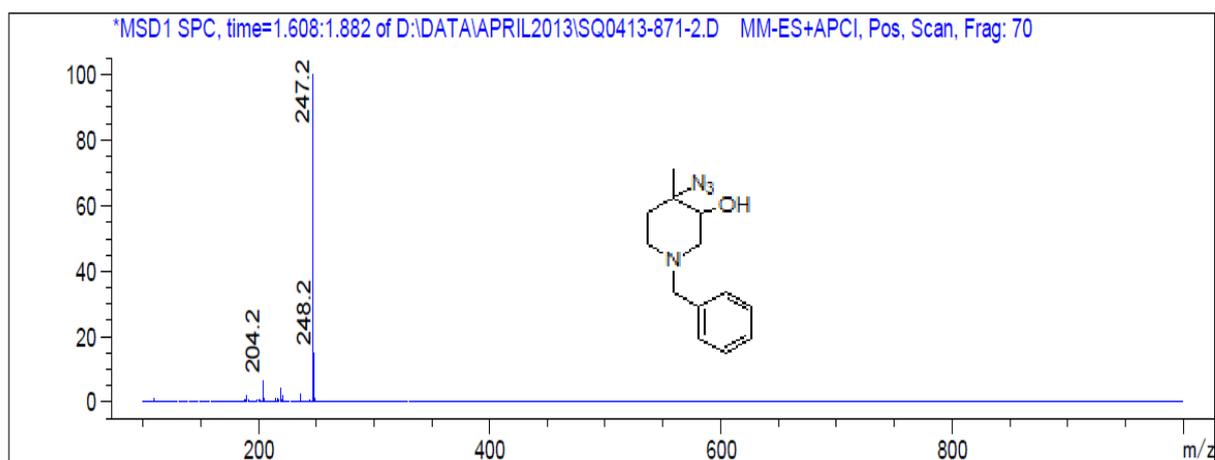
Data File Name : D:\DATA\APRIL2013\SQ0413-871-2.D  
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 Mobile Phase A: 0.1% Formic Acid in Aq B: ACN  
 T/%B : 0/5, 1.5/90, 4/90, 4.1/5  
 Flow rate: 1.0ml/min  
 Instrument: Agilent 6120 Quadrupole LCMS



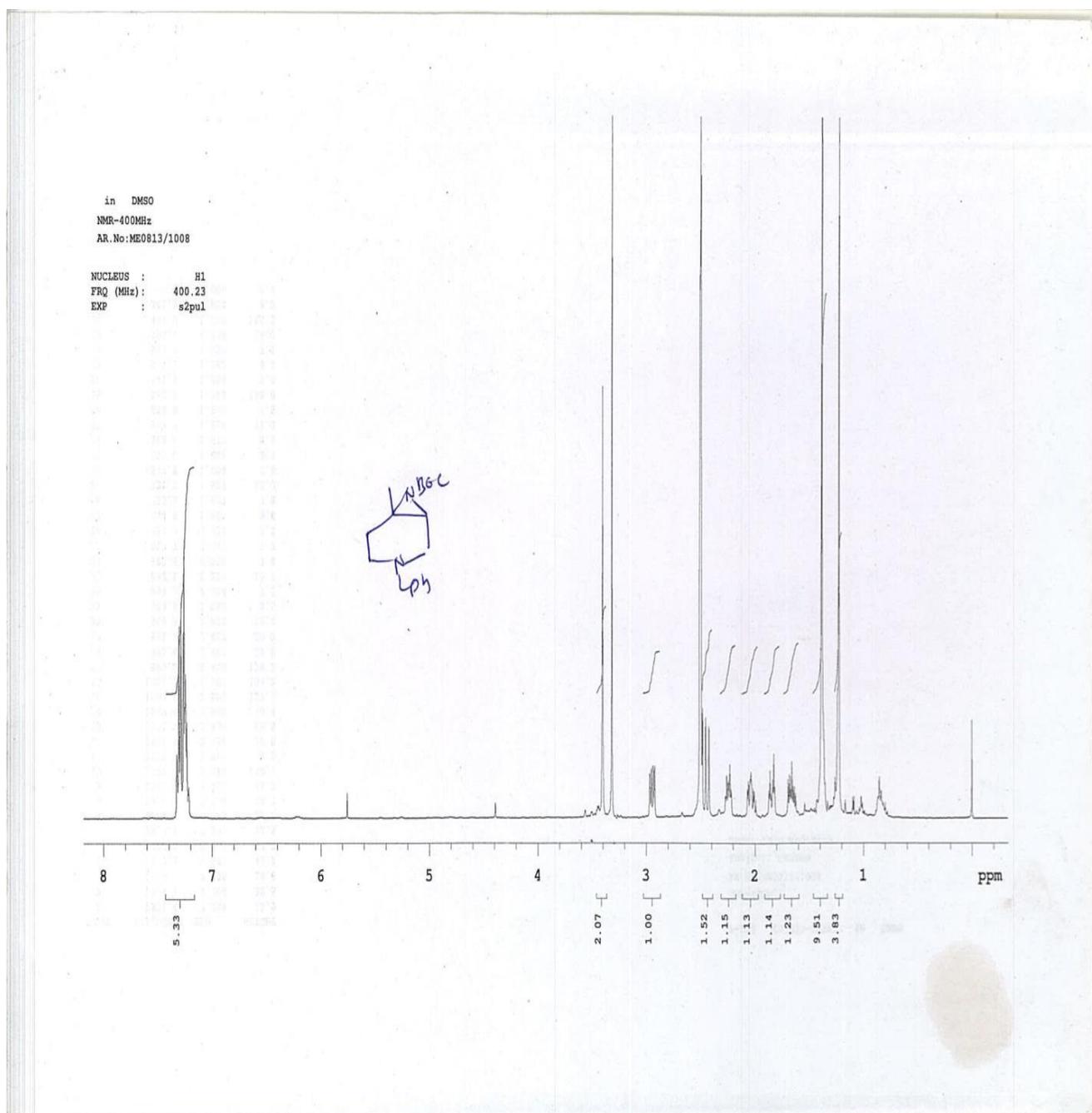
Peak No	RT min	Area	Area %
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2	1.628	5.595e+003	96.236
3	1.809	1.795e+001	0.309
4	2.087	6.622e+001	1.139
5	2.825	4.350e+001	0.748



Analysed by :

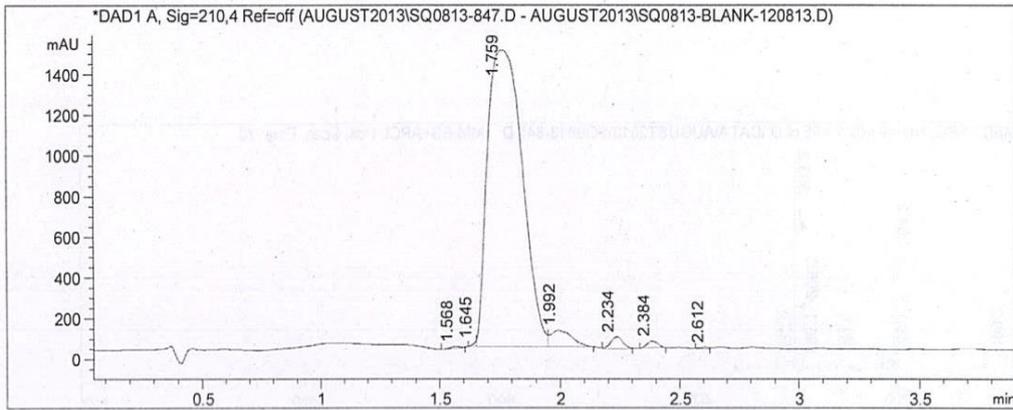


<sup>1</sup>H-NMR and LCMS of Compound 9

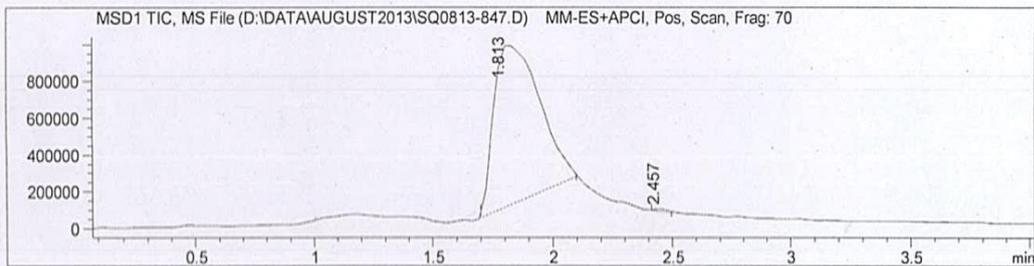
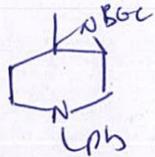


Acq. Method : D:\METHODSS\KINETEX\_FA1.M

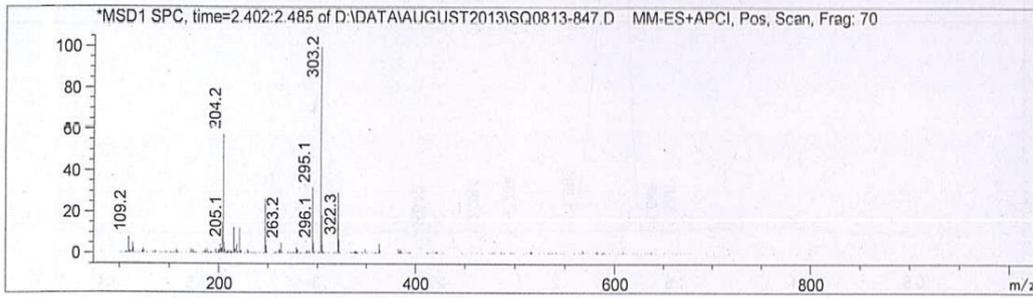
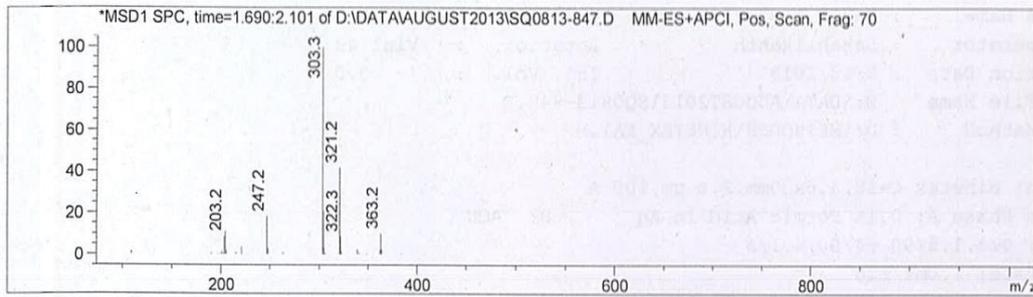
Column: Kinetex C-18, 4.6x30mm, 2.6 um, 100 A  
 Mobile Phase A: 0.1% Formic Acid in Aq B: ACN  
 T/%B : 0/5, 1.5/90, 4/90, 4.1/5  
 Flow rate: 1.0ml/min  
 Instrument: Agilent 6120 Quadrupole LCMS



Peak No	RT min	Area	Area %
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2	1.645	2.760e+001	0.178
3	1.759	1.467e+004	94.554
4	1.992	4.773e+002	3.076
5	2.234	2.075e+002	1.337
6	2.384	9.613e+001	0.620
7	2.612	1.507e+001	0.097

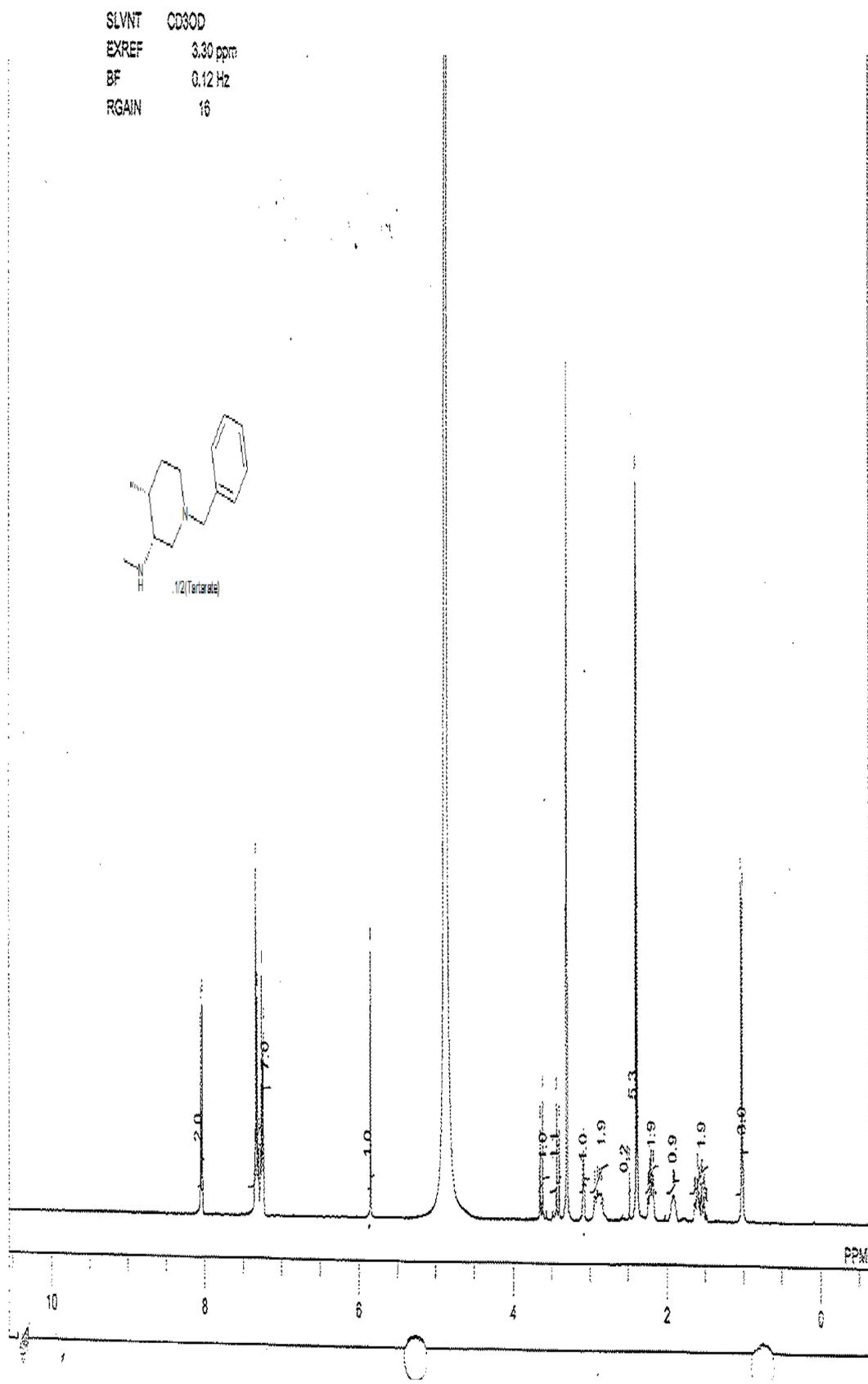


Analysed by :



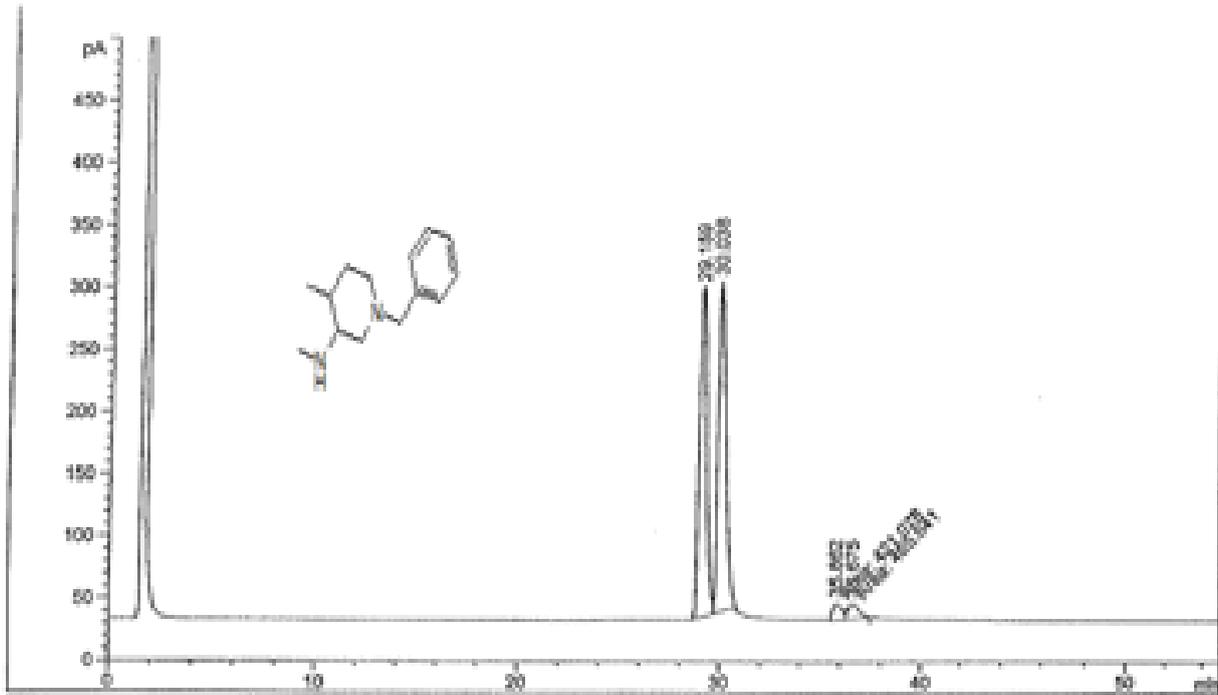
Analysed by :

<sup>1</sup>H-NMR and HPLC of Compound 10



Seq. Line : 5  
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Acq. Method : F:\PRAC01\METHOD\TFC-CHE.M  
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 Analysis Method : F:\HPCHEM\1\METHODS\REF\_TEA.M



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 Area Percent Report  
 =====

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 Use Multiplier & Dilution Factor with ISTDs

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2	30.038	VB	0.3120	6999.34570	48.39448	Cis (R,R) Isomer
3	35.862	MF	0.5116	421.07809	2.91139	Trans Isomer
4	36.675	FM	0.6198	465.64130	3.21951	Trans Isomer

Totals : 1.44631e4

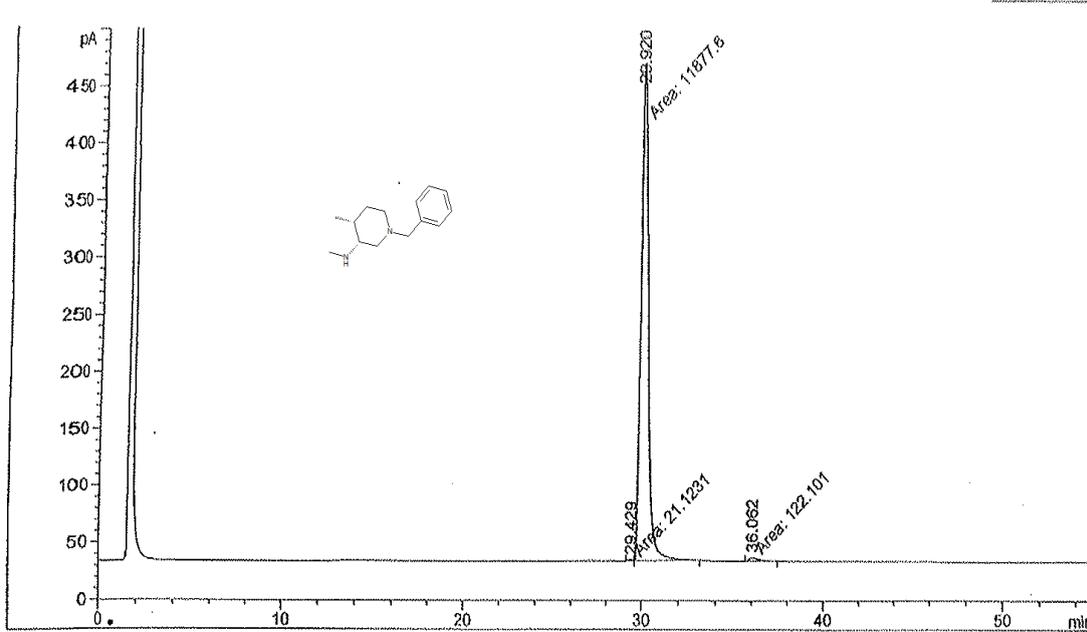
Results obtained with enhanced integrator!  
 1 Warning or Errors :

Warning : Calibration warnings (see calibration table listing)

4/109 DTT

Seq. Line : 3  
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 Inj : 1  
 Inj Volume : 1 µl

Acq. Method : F:\PRAGC01\METHOD\TFC-CHI.M  
 Analysis Method : F:\HPCHEM\1\METHODS\WRF\_TEA.M



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 Area Percent Report  
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Sorted By : Signal  
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 Use Multiplier & Dilution Factor with ISTDs

Signal 1: FID1 A,

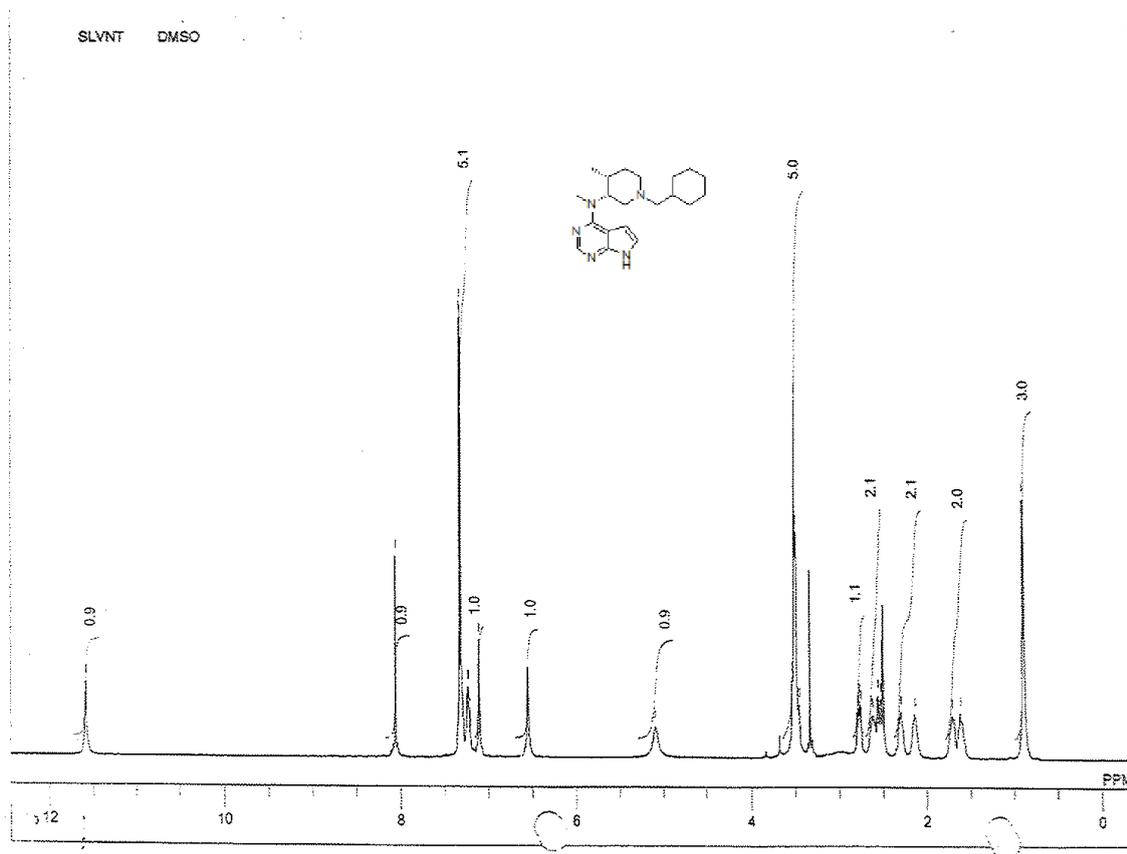
Peak #	RetTime [min]	Type	Width [min]	Area [pA*s]	Area %	Name
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2	29.920	FM	0.4536	1.18776e4	98.80853	Cis (R,R) Isomer
3	36.062	MM	0.6534	122.10132	1.01575	Trans Isomer
4	36.675		0.0000	0.00000	0.00000	Trans Isomer

Totals : 1.20208e4

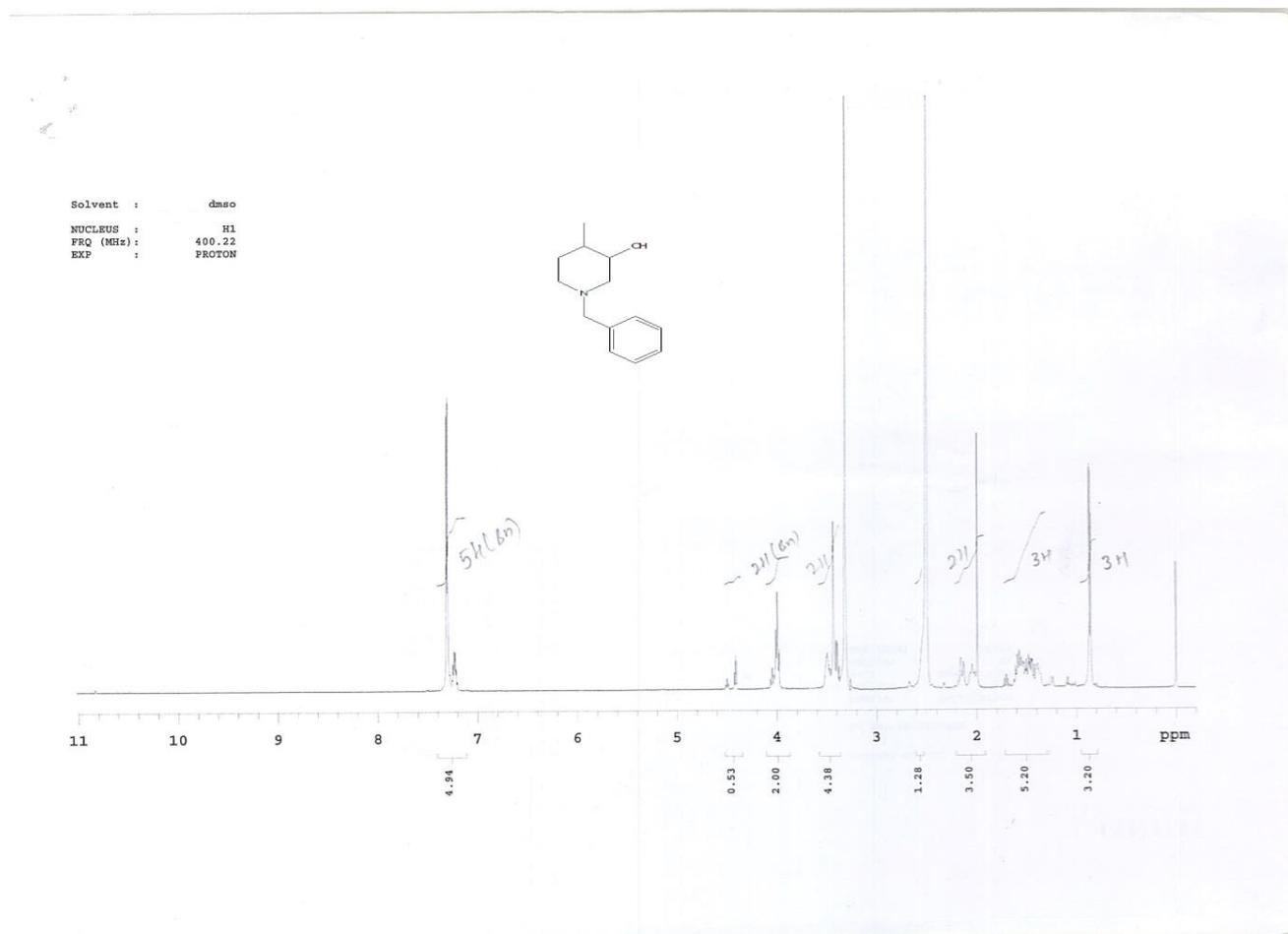
Results obtained with enhanced integrator!  
 2 Warnings or Errors :

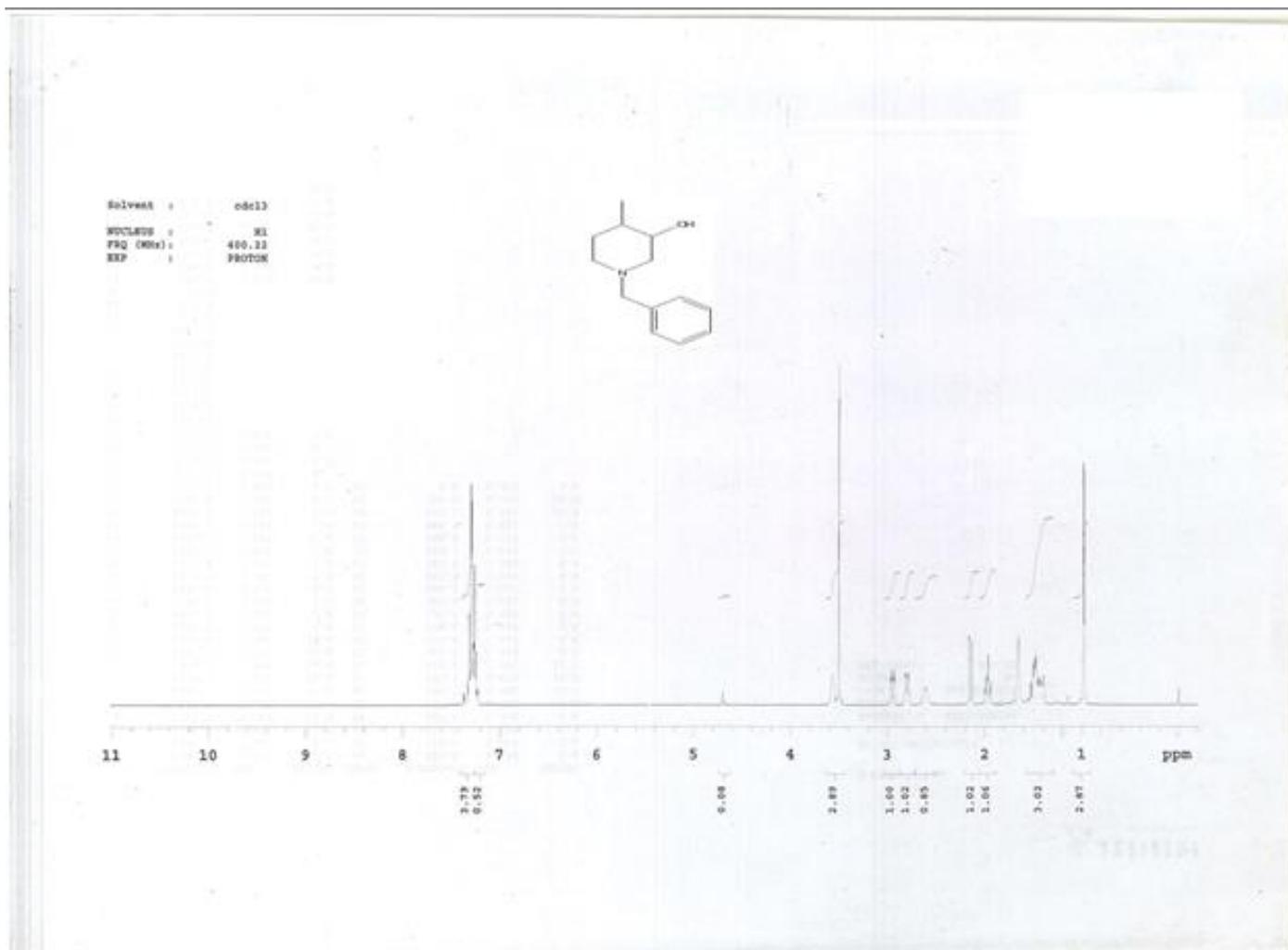
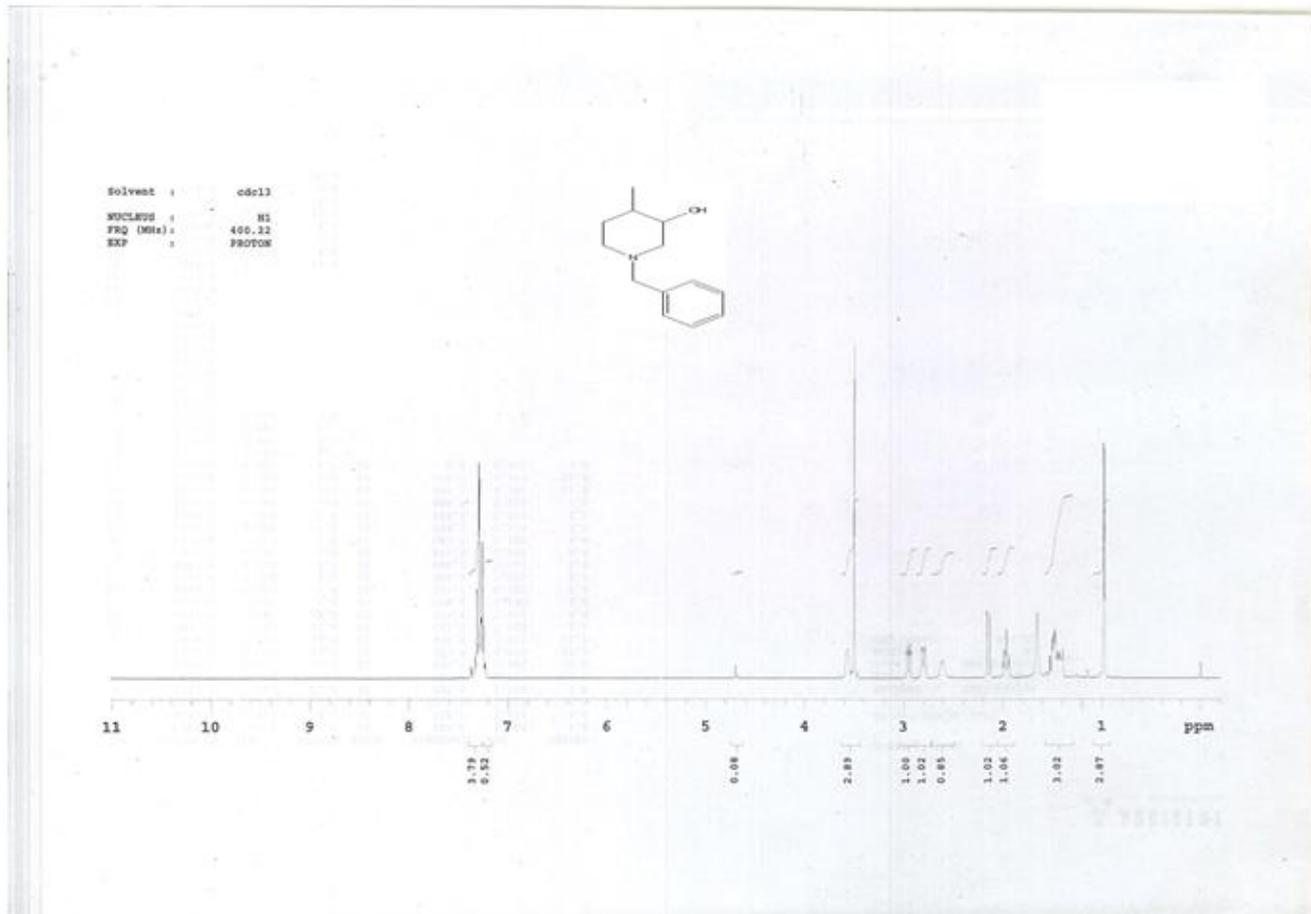
Warning : Calibration warnings (see calibration table listing)  
 Warning : Calibrated compound(s) not found

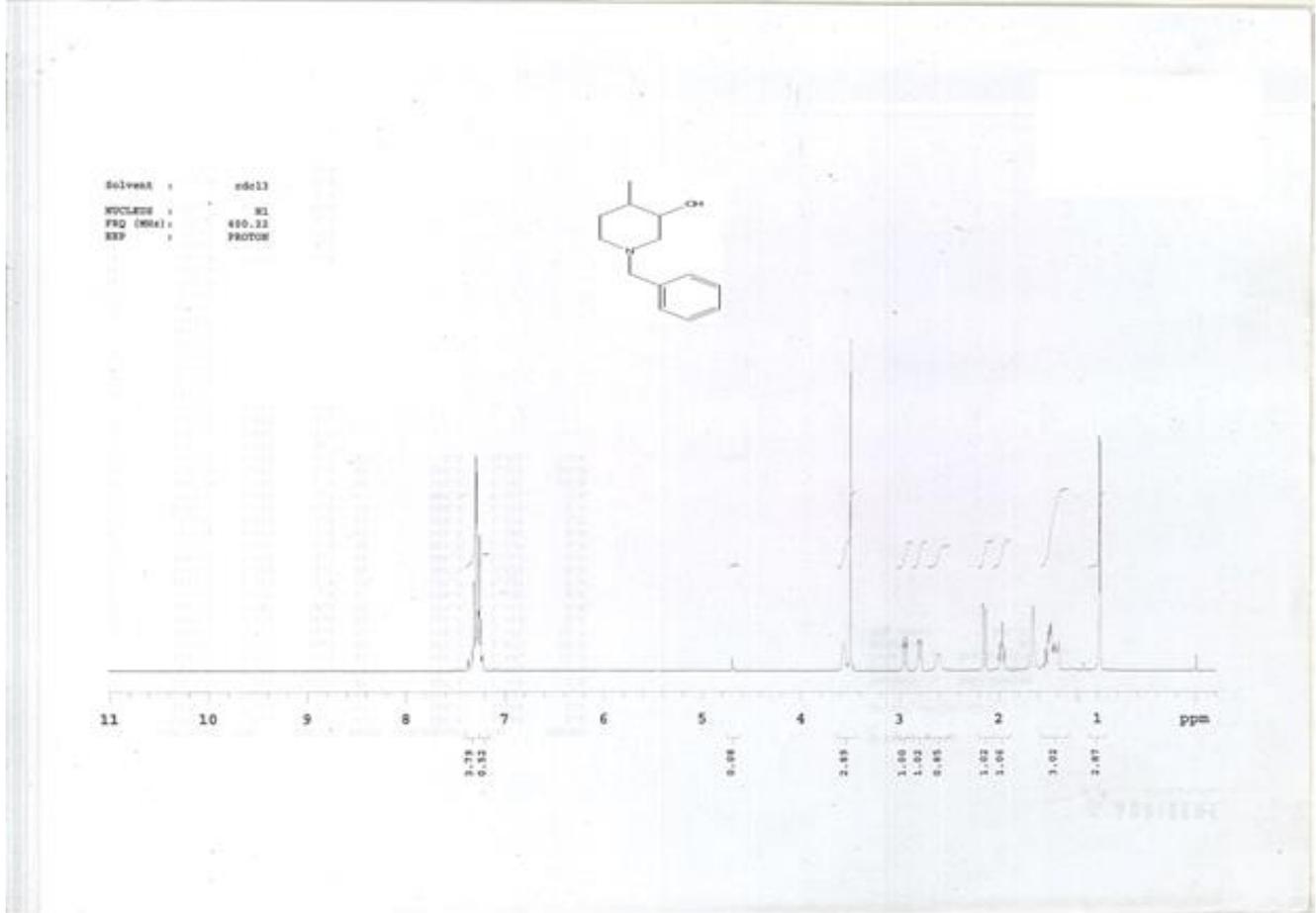
<sup>1</sup>H-NMR and HPLC of compound



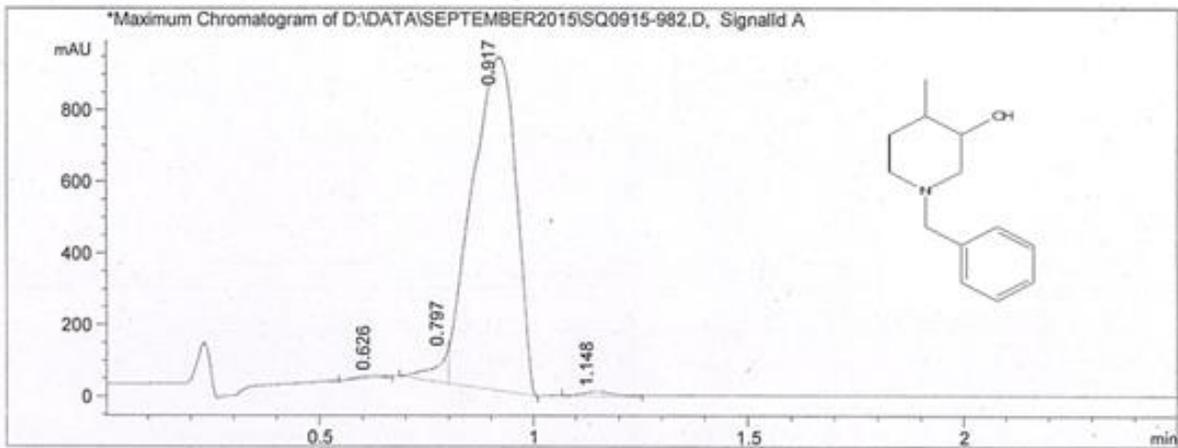
<sup>1</sup>H-NMR in DMSO, CDCl<sub>3</sub>, <sup>13</sup>C and LCMS of compound







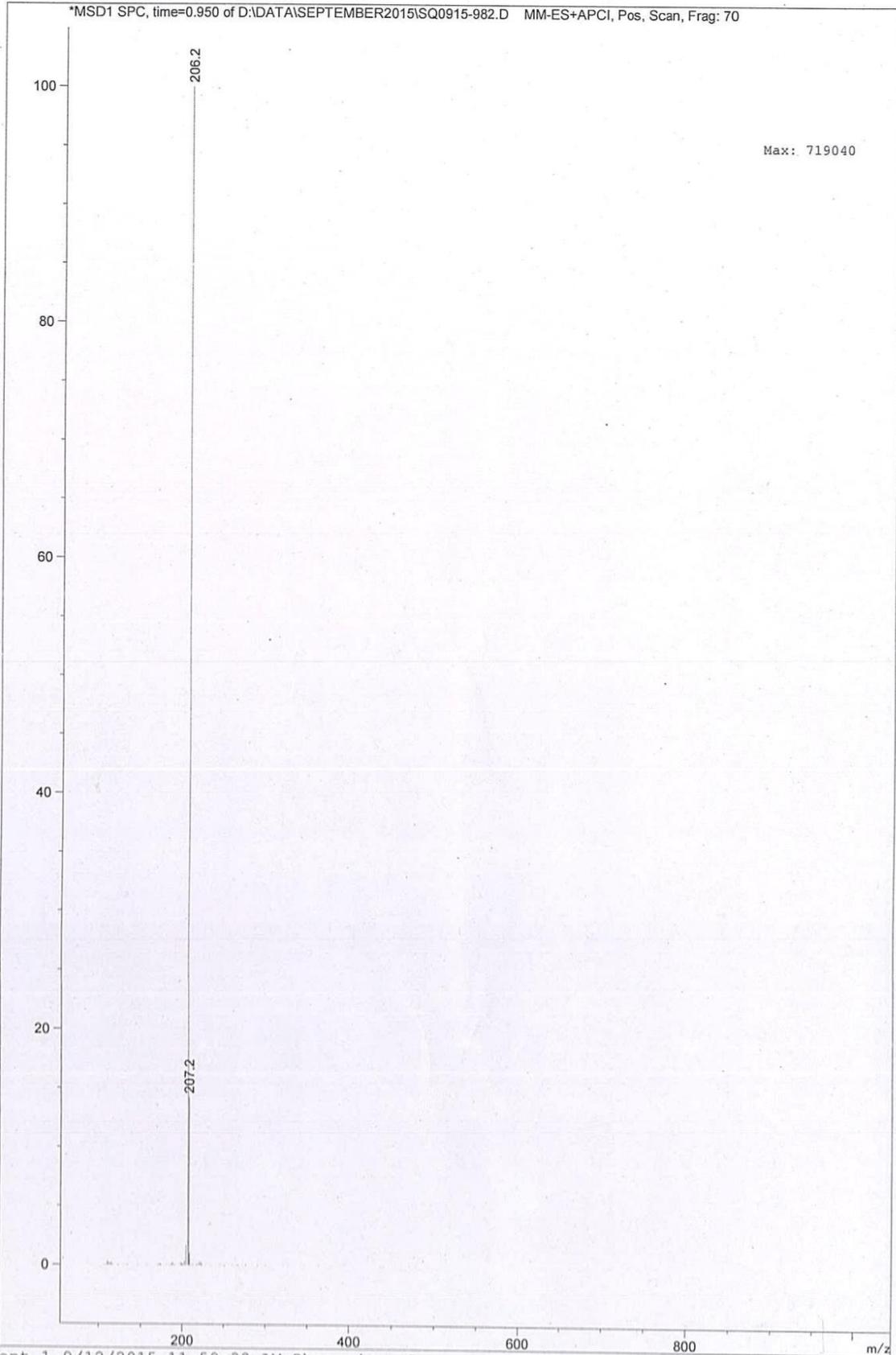
Acq. Method :  
 Column: Kinetex C-18, 4.6x30mm, 2.6 um, 100 A  
 Mobile Phase A: 0.1% Formic Acid in Aq B: ACN  
 T/%B : 0/5, 0.5/95, 2.4/95, 2.5/5  
 Flow rate: 1.5ml/min  
 Instrument: Agilent 6120 Quadrupole LCMS



Peak#	RT(min)	Area	Area %
1	0.63	28.81	0.41
2	0.80	227.85	3.26
3	0.92	6651.89	95.23
4	1.15	76.72	1.10

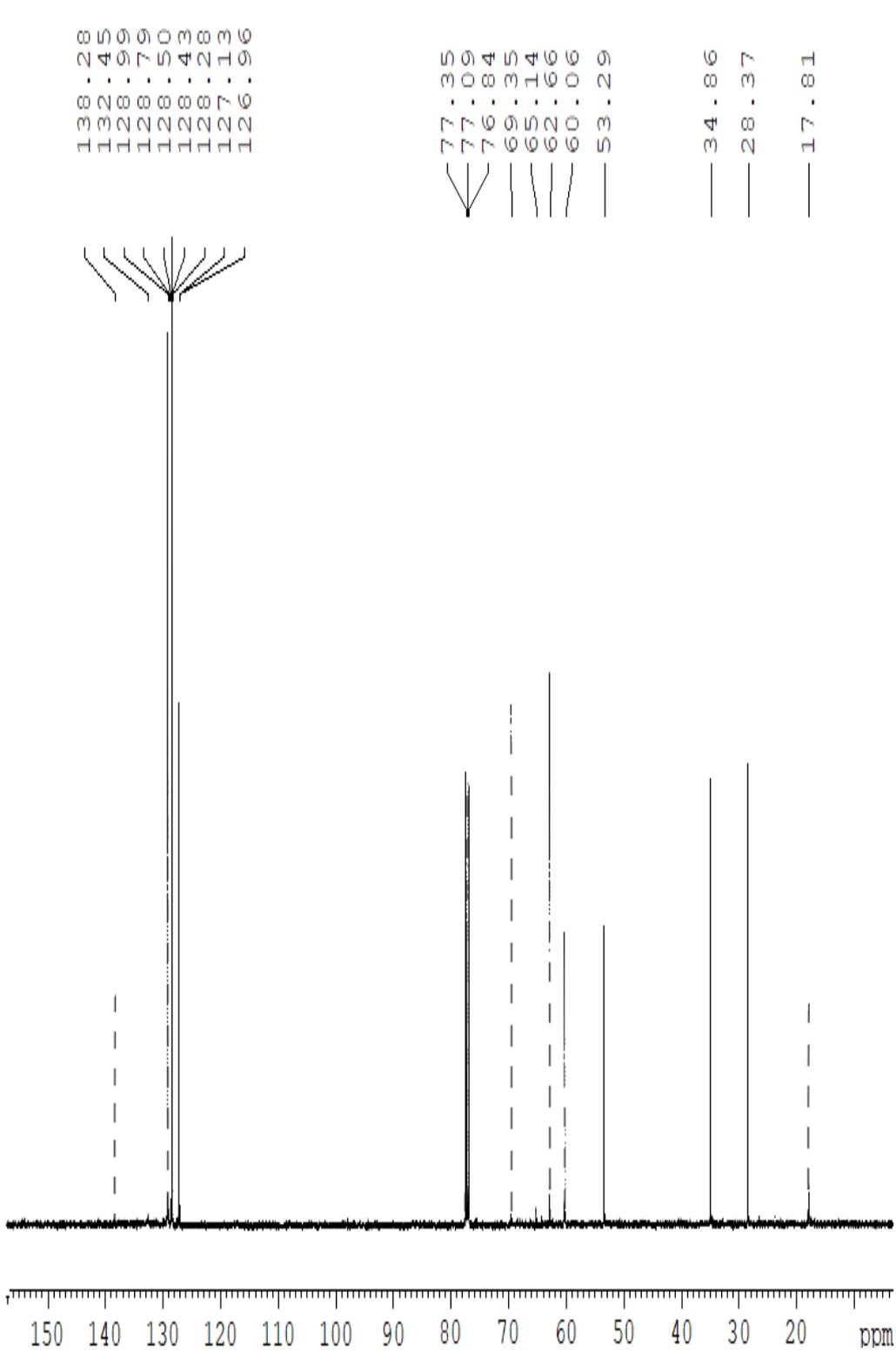
Print of window 80: MS Spectrum

MS Spectrum



Instrument 1 9/12/2015 11:59:30 AM Dhanunjaya Rao

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Current Data Parameters  
 NAME Mohan  
 EXPNO 669  
 PROCNO 1

F2 - Acquisition Parameters  
 Date\_ 20170228  
 Time 12.05  
 INSTRUM spect  
 PROBHD 5 mm PABBO BB-  
 PULPROG zgpg30  
 TD 65536  
 SOLVENT CDC13  
 NS 152  
 DS 4  
 SWH 29761.904 Hz  
 FIDRES 0.454131 Hz  
 AQ 1.1010048 sec  
 RG 203  
 DW 16.800 usec  
 DE 6.50 usec  
 TE 299.5 K  
 D1 2.0000000 sec  
 D11 0.0300000 sec  
 TDO 1

===== CHANNEL f1 =====  
 SFO1 125.7955112 MHz  
 NUC1 13C  
 P1 8.65 usec  
 PLW1 120.5000000 W

===== CHANNEL f2 =====  
 SFO2 500.2320009 MHz  
 NUC2 1H  
 CPDPRG[2] waltz16  
 PCPD2 80.00 usec  
 PLW2 27.16399956 W  
 PLW12 0.34685999 W  
 PLW13 0.22199000 W

F2 - Processing parameters  
 SI 32768  
 SF 125.7829335 MHz  
 WDW EM  
 SSB 0  
 LB 1.00 Hz  
 GB 0  
 PC 1.40