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# Stability Indicating HPLC Method Development and Validation of Lamotrigine in Bulk and Tablet Dosage Form

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### ABSTRACT

A simple, precise, linear stability indicating RP-HPLC method was developed and validated as per ICH guidelines for the determination of lamotrigine in the bulk drug and tablet dosage form. Chromatographic separation was performed by using agilant HPLC 1200 series include isocratic mode on a C18 Qualisil BDS (250 mm  $\times$  4.5 mm  $\times$  5 mm) column using mobile phase containing potassium dihydrogen ortho phosphate buffer (pH 7.4) and methanol in the ratio 60: 40 at flow rate 1.3 ml/min. The wavelength for detection was 305 nm at room temperature. The retention time for separation was 5.3 min and linearity of lamotrigine was perceived in the concentration range of 5-30 µg/ml with correlation coefficient of 0.999. Force degradation of lamotrigine was completed under acidic, basic, oxidation, dry heat and photolytic conditions. The method was found precise, linear, and accurate as per ICH guidelines and may be used as an alternative method in industry.

Keywords: Lamotrigine, HPLC, ICH guideline, Stability indicating, Validation.

#### INTRODUCTION

Lamotrigine is an anticonvulsant drug used to treat bipolar disorder as well as epilepsy [1,2]. It increases the action of Gamma-Aminobutyric Acid (GABA) and it act as inhibitory neurotransmitter in the central nervous system and decrease voltage-sensitive sodium channels It is the part of the sodium channel blocking class of anticonvulsant drugs, but it has an additional action which is broader spectrum antiepileptic action as compare to other sodium channel blocking anticonvulsant drugs like a phenytoin and carbamazepine. The chemical name of Lamotrigine is 3,5-diamino-6-(2,3-dichlorophenyl)-1,2,4-triazine, which has molecular formula  $C_9H_7Cl_2N_5$  with molecular weight is 256.091 gm/mol (Figure 1). The common side effects of the lamotrigine involve sleepiness, headache, vomiting, trouble with coordination, and rash. Lamotrigine is a white to pale cream-colored powder, freely soluble in methanol and acetonitrile and slightly soluble in water (0.17 mg/ml at 25°C) which has pKa is 5.7.

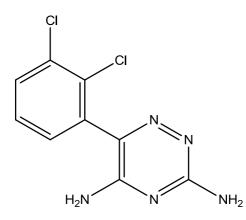


Figure 1: Structure of Lamotrigine

The United State of Pharmacopeia reported HPLC and UV methods for the determination of Lamotrigine [3]. Moreover British Pharmacopeia

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describes a nonaqueous acid base titration for the determination of pure drug [4]. A spectrophotometric, TLC and HPLC methods [5] were used for the determination of lamotrigine with its impurities [6]. Literature survey also reveals that RP-HPLC using solvent-demixing extraction, UV spectroscopic, spectrofluorimetric and stability indicating LC method were developed for the determination of lamotrigine in various matrices such as human plasma, pharmaceutical dosage forms, bulk, in whole blood, brain and urine samples [7-18].

Reviewing the literature in hand, no chromatographic method has been studied by using methanol and potassium dihydrogen ortho phosphate buffer (pH 7.4). A simple protein precipitation with methanol guaranteed high extraction yield values (>90%) and good purification from matrix interference. Therefore, this work was directed towards the development of a simple and sensitive stability-indicating HPLC that can be used to establish a stability profile of the drug and consequently assess its purity in bulk and tablet form.

## MATERIALS AND METHODS

#### Materials

Lamotrigine was gifted by Wokhardt Pharmaceuticals Ltd, Aurangabad. The lamictal tablet formulation of lamotrigine was purchased from the local market. The water of HPLC was prepared by using double distillation and filtration through 0.45 mm filters. Methanol of HPLC grade was obtained from E. Merck (India) Ltd., Mumbai. Potassium dihydrogen phosphate and sodium hydroxide of analytical grade were purchased from E. Merck (India) Ltd., Mumbai. Ortho-phosphoric acid was obtained from E. Merck (India) Ltd., Mumbai.

#### Instrumentation and condition

The liquid chromatographic system used was an isocratic HPLC Agilent Technology system involve pump (Quaternary, Agilent, Germany), PDA detector equipped with EZ Chrome elite software and a manual sample injector volume with a 50  $\mu$ l sample. The chromatographic separation was done on C-18 Qualisil BDS (250 mm × 4.5 mm × 5 mm) column. The mobile phase composition was taken 0.02 M potassium dihydrogen ortho phosphate buffer: methanol (60: 40 v/v) pH adjusted to 7.5 with dilute ortho phosphoric acid and filtered through Whatman filter paper. All the separations were done isocratically at a flow rate 1.3 ml/min at RT.

#### Preparation of stock and standard solutions

Stock solution of lamotrigine was prepared by dissolving accurately weighed 10 mg of lamotrigine standard in 100 ml volumetric flask, added 30 ml of methanol, sonicated to dissolve and make up the volume with methanol and mix well to obtain concentration of 100  $\mu$ g/ml. This was further diluted with the same solvent to obtain concentration of 10  $\mu$ g/ml.

#### Preparation of test solution for assay

Twenty tablets of lamotrigine were powdered. An amount of powdered drug equivalent to 10 mg of drug was weighed and transferred to the 100 ml volumetric flask containing 30 ml of methanol and sonicated for 20 min. After sonication, the final volume makes upto 100 ml by adding methanol to obtain concentration of 100  $\mu$ g/ml. This was further diluted with the same solvent to obtain concentration of 10  $\mu$ g/ml. The sample solution was filtered through 0.45  $\mu$ m nylon syringe filter

#### Method validation

#### Linearity

Linearity of drug was performed by preparing six sample of concentration range from 5, 10, 15, 20, 25, 30 µg/ml and injected into system. The peak areas versus concentration data were plotted to obtained calibration curve.

#### Precision

Intraday precision was determined by analyzing, the three different concentrations 5, 10 and 15  $\mu$ g/ml of lamotrigine, for three times in the same day. Day to day variability was assessed using above mentioned three concentrations analyzed on three different days.

#### Accuracy

Accuracy of the method was studied at three different levels i.e. 80, 100 and 120 % levels. To the pre-analyzed sample solution of lamotrigine tablet formulation 15  $\mu$ g/ml i.e. considered as 100% and a known amount of 12, 15, 18  $\mu$ g/ml of lamotrigine bulk stock standards solution i.e. 80%, 100% and 120% were added into formulation solution respectively.

#### Repeatability

Repeatability experiment was performed by analyzing sample solution 10  $\mu$ g/ml of lamotrigine into the system and measuring the peak area. It was repeated for six times.

#### Sensitivity

The LOD of an individual analytical procedure is the lowest amount of analyte in a sample. The LOQ is the concentration that can be quantified reliably with a specified level of accuracy and precision.

#### Ruggedness

Ruggedness of method was determined by doing the analysis from two different analysts person and the respective areas were noted. The result was designated by % RSD.

## Analysis of tablet

The amount lamotrigine was determined in tablet using linear regression equation. These procedures were repeated for six times.

## Forced degradation studies

Forced degradation studies were carried out for evaluation of stability indicating properties. All sample solution for stress studies was prepared at an initial concentration of 100 µg/ml of lamotrigine.

#### Acid hydrolysis studies

Acid degradation was carried out in 0.1 N HCl at concentration of 100 µg/ml. Lamotrigine was refluxed with 0.1 N HCl at 80°C for 3 h. The

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stressed sample was neutralized after cooling to room temperature and diluted with methanol as per requirement with diluents, filtered and injected.

## Alkali degradation studies

Alkali degradation was carried out in 1 N NaOH at concentration of 100  $\mu$ g/ml, Lamotrigine was refluxed with 1 N NaOH at 80°C for 3 h. The stressed sample was neutralized after cooling to room temperature and diluted with methanol as per requirement with diluents, filtered and injected.

## Oxidation

Oxidation degradation was carried out by using 15%  $H_2O_2$  solution and stored at room temperature for 24 h. Filtered and 10  $\mu$ g/ml dilution was prepared by taking 0.1 ml above solution in 10 ml volumetric flask and volume make up with methanol and injected.

## Neutral

Neutral degradation was carried out in water at concentration of 100 µg/ml, Lamotrigine was refluxed with water at 80°C for 3 h. The stressed sample was neutralized after cooling to room temperature and diluted with methanol as per requirement with diluents, filtered and injected.

#### Temperature stress study

A temperature stress study was done by weighing 100 mg of Lamotrigine sample and it was kept into hot air oven at 60°C for 6 h. After 6 h 10 mg sample transferred into 100 ml volumetric flask and volume make up with diluents, From above solution 10 µg/ml dilution was prepared and injected.

#### Photo stability

The photo stability study was done by weighing 100 mg of Lamotrigine and exposed in sun light for 7 days, 10 mg sample transferred into 100 ml volumetric flask and volume make up with diluents. From above solution  $10 \mu g/ml$  dilution was prepared and injected.

## **RESULTS AND DISCUSSION**

## Development and Optimization of the HPLC method

For the method development, various organic solvent were used for optimization of mobile phase. Different proportions of solvent were tried for the optimization of mobile phase to obtain the good resolution peak by using the C-18 column. Beginning of mobile phase optimization was tried by taking the methanol: water in 50: 50 v/v as a mobile phase but tailing was observed. After this acetonitrile: water was tried in proportion 50: 50 v/v then noise was observed and in 40: 60 v/v proportions no peak observed. Methanol was tried with different proportion of buffer with various pH and finally sharp peak obtainer at mobile phase methanol: Potassium dihydrogen ortho phosphate buffer pH adjusted 7.4 with ortho phosphoric acid at wavelength 305 nm, Retention time was 5.3 min with flow rate 1.3 ml/min. Figures 2 and 3 represent the chromatograms of standard and test preparation respectively.

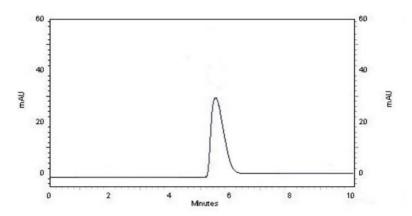


Figure 2: Chromatogram of standard preparation

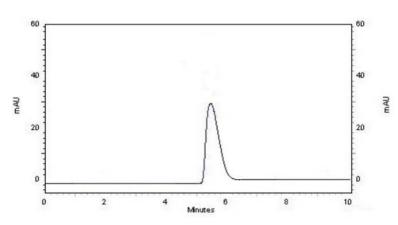
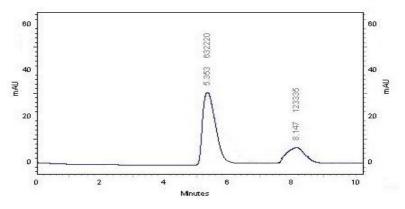


Figure 3: Chromatogram of test preparation

## Forced degradation studied (specificity)

Specificity is the ability of method to determine drug in presence of other impurities, degradation etc. In which the forcefully degradation was carried out by exposing drug with acid, base, hydrogen peroxide, water, thermally and photolytically (Figures 4-10).





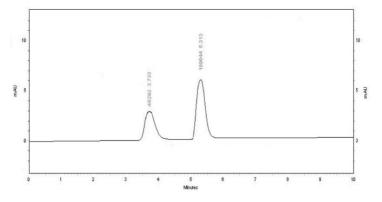


Figure 5: Chromatogram of base degraded sample

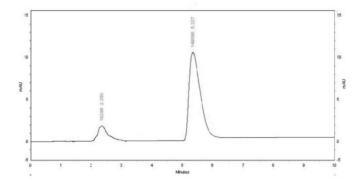


Figure 6: Chromatogram of H<sub>2</sub>O<sub>2</sub> degraded sample

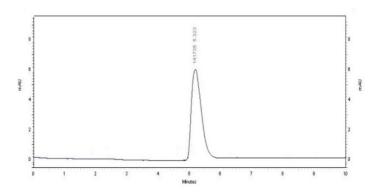


Figure 7: Chromatogram of water degraded sample

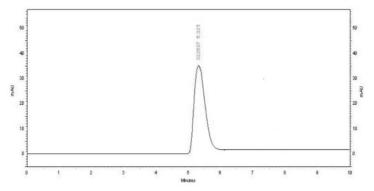


Figure 8: Chromatogram of thermally degraded sample

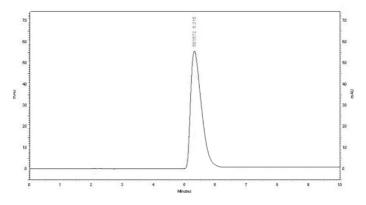


Figure 9: Chromatogram of photolytic degraded sample

The results of forced degradation study were obtained in Table 1.

Table	1:	Result	of forced	degradation	study
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Stress condition	% Degradation	% Active drug after degradation
Acidic Degradation	19.5	80.5
Basic Degradation	25.46	74.54
Oxidative Degradation	12.88	87.12
Neutral Degradation	No degradation	100
Thermal Degradation	No degradation	100
Photolytic Degradation	No degradation	100

## Linearity

From stock solution of Lamotrigine aliquots of 0.5 to 3 ml were taken in 10 ml volumetric flask diluted up to the mark with methanol such that the final concentration of Lamotrigine in the range 5-30  $\mu$ g/ml. The peak areas versus concentration data were plotted to obtained calibration curve. Result of linearity was obtained in (Table 2).

#### Table 2: Result of linearity of lamotrigine

Concentration (µg/ml)	Area
5	116711
10	174368
15	220213
20	272385
25	328328
30	382538

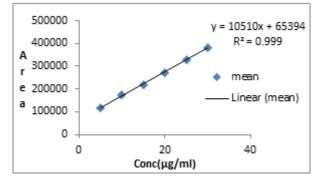


Figure 10: Calibration curve of Lamotrigine

## Precision

Intra-day precision was determined by analyzing, the three different concentrations 5  $\mu$ g/ml, 10  $\mu$ g/ml and 15  $\mu$ g/m of lamotrigine, for three times in the same day. Inter-day variability was assessed using above mentioned three concentrations analyzed on three different days, over a period of one week. Result of precision obtained in Table 3.

#### Table 3: Result of precision of lamotrigine

Common d		Intra-day Precision <sup>*</sup> (n=3)		Inter-day Precision <sup>*</sup> (n=3)	
Compound	Conc. (µg/ml)	Amount found	% RSD	Amount found	% RSD
Lamotrigine	5	4.988	0.5906	4.953	0.5904
	10	9.946	0.6284	10.23	0.5684
	15	14.74	0.3011	14.65	1.0519

## Accuracy

Recovery studied was performed by taking concentration in level of 80%, 100%, 120% and recoveries of spike drug were obtained in Table 4.

#### Table 4: Results of accuracy studies for Lamotrigine

Drug		Initial amount (µg/ml)	Amount added (µg/ml)	% Recovered	% RSD
	80%	15	12	99.62	0.593
Lamotrigine	100%	15	15	99.12	0.485
	120%	15	18	99.15	0.366

## Repeatability

Repeatability experiment was performed by injecting sample solution 10  $\mu$ g/ml of lamotrigine into the system and measuring the peak area. It was repeated for six times; results were shown in Table 5.

## Table 5: Results of repeatability studies for Lamotrigine

Amount taken (µg/ml) (n=6)	Amount found	%Amount found	± SD	%RSD
10	9.98	99.85	0.558	0.5588

#### Analysis of tablet

A 10 µg/ml dilution was prepared from tablet stock solution and injected into system. This procedure was performed for six times; results were shown in Table 6.

#### Table 6: Results of tablet assay studies for Lamotrigine

Amount taken (µg/ml)	Amount found (n=6)	%Amount Found	±SD	%RSD
10	9.94	99.48	0.551	0.554

## LOD and LOQ

Sensitivity of the proposed method was estimated in terms of Limit of Detection (LOD) and Limit of Quantitation (LOQ). The LOD and LOQ were calculated by the use of equation,  $LOD=SD/S \times 3.3$  and  $LOQ = SD/S \times 10$ , where SD is the residual standard deviation of the peak areas of the drugs (n=6) and 'S' is the slope of the line. Sensitivity was performed between 6-10 µg/ml, and results are shown in Table 7.

#### Table 7: Result of Sensitivity of Lamotrigine

LOD (µg/ml)	LOQ (µg/ml)
0.11	0.35

## CONCLUSION

Stability indicating method development and validation of Lamotrigine by HPLC was studied as per ICH guidelines and method was found to be linear, precise, accurate, specify and stability indicating. This method can be used as alternative method in industry for the determination of Lamotrigine in pharmaceutical formulation.

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#### REFERENCES

[1] W.O. Foye, Foye's principles of medicinal chemistry, Lippincott Williams & Wilkins, 2008.

[2] C. Dollery, S73, **1999**.

[3] United States Pharmacopeial Convention, Committee of R, The United States Pharmacopeia, 29<sup>th</sup> (Edn.), United States Pharmacopeial Convention, Incorporated, **2010**.

[4] British Pharmacopoeia C. British Pharmacopoeia Her Majesty's Stationary Office. London, 1998.

[5] J. Emami, N. Ghassami, F. Ahmadi, J. pharm. biomed. Anal., 2006, 40(4), 999-1005.

[6] N.F. Youssef, E.A. Taha, Chem. pharm. bull., 2007, 55(4), 541-5.

[7] M.T. Martins, C.S. Paim, M. Steppe, Chromatogr. Res. Int., 2011, 2011.

[8] N.M. El-Enany, D.T. El-Sherbiny, A.A. Abdelal, F.F. Belal, J. fluorescence., 2010, 20(2), 463-472.

[9] P. Srinivasulu, K. Mukkanti, B.R. Reguri, K.S.V. Srinivas, Chromatographia., 2009, 70(1-2), 271-6.

[10] S. Bompadre, A. Tagliabracci, M. Battino, R. Giorgetti, J. Chromatogr. B., 2008, 863(1), 177-180.

[11] D. Croci, A. Salmaggi, U. de Grazia, G. Bernardi, *Ther. drug monit.*, **2001**, 23(6), 665-8.

[12] M.M. Castel-Branco, A.M. Almeida, A.C. Falcao, T.A. Macedo, M.M. Caramona, F.G. Lopez, J. Chromatogr. B Biomed. Sci. Appl., 2001, 755(1-2), 119-127.

[13] M. Torra, M. Rodamilans, S. Arroyo, J. Corbella, Ther. drug monit., 2000, 22(5), 621-625.

[14] E. Vidal, C. Pascual, L. Pou, J. Chromatogr. B Biomed. Sci. Appl., 1999, 736(1-2), 295-8.

[15] P. Angelis-Stoforidis, D.J. Morgan, T.J. O'Brien, F.J.E. Vajda, J. Chromatogr. B Biomed. Sci. Appl., 1999, 727(1-2), 113-118.

- [16] K.M. Matar, P.J. Nicholls, S.A. Bawazir, M.I. Al-Hassan, A. Tekle, J. Pharm. Biomed. Anal., 1998, 17(3), 525-31.
- [17] M. Cociglio, R. Alric, O. Bouvier, J. Chromatogr. B Biomed. Sci. Appl., 1991, 572(1-2), 269-276.

[18] C.L. Cheng, C.H. Chou, O.P. Hu, J. Chromatogr. B., 2005, 817(2), 199-206.