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Studies of novel heterocyclic compounds

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ABSTRACT

Novel heterocyclic compound namely, 4-((1-((4-((dialkylamino)methyl)-5-thioxo-4,5-dihydro-1,3,4-oxadiazol-2-yl)methyl)-1H-benzo[d]imidazol-2-yl)methylamino)-N-(pyrimidin-2-yl)benzenesulfonamide(4a-e) have been prepared by mannich reaction, a N-(pyrimidin-2-yl)-4-((1-((5-thioxo-4,5-dihydro-1,3,4-oxadiazol-2-yl)methyl)-1H-benzo[d]imidazol-2-yl)methylamino) benzenesulfonamide(3) react with formaldehyde and different secondary amines. The compound (3) prepared from 4-((1-(2-hydrazinyl-2-oxoethyl)-1H-benzo[d]imidazol-2-yl)methylamino)-N-(pyrimidin-2-yl)benzenesulfonamide(2) with CS₂/KOH. 4-((1H-benzo[d]imidazol-2-yl)methylamino)-N-(pyrimidin-2-yl)benzene sulfonamide(1) react with chloro acetic acid and hydrazine hydrate gives compound (2). All the structures of novel synthesized compounds were established on basis of analytical and spectral data. The newly synthesized compounds were studied for their antibacterial and antifungal activities.

Keywords: Mannich reaction, Sulfapyrimidine, Benzimidazole, Spectral studies, antibacterial and antifungal activities.

INTRODUCTION

The heterocyclic compounds, especially nitrogen-containing heterocycles with a sulfur atom are an important class of compounds in medicinal chemistry.[1-4] Hydrazinolysis of esters is the conventional method for preparing acyl hydrazides [5,6]. However, when this method was applied to an α , β -unsaturated ester, the predominant product was the corresponding pyrazolidinone, the result of hydrazinolysis and an undesired subsequent intramolecular Michael-type addition [7]. In current era, the research on hydrazides and its derivatives are carried out due to their heterocyclised products display diverse biological activities including antibacterial, antifungicidal, analgesic, anti-inflammatory properties [8-10]. Other heterocyclic compounds says, oxadiazoles and their condensed products play a vital role in medicinal chemistry and exhibit diverse biological activities such as antibacterial, antifungal, anti-inflammatory, analgesic and anticancer activity[11-15]. Hence, it was thought of interest to combine Sulfapyrimidine containing benzimidazole with oxadiazole moieties which may enhance the drug activity of compounds to some extent, or they might possess some of the above mentioned biological activities. Hence in continuation of our earlier work [16], the current communication covers the study of 4-((1-((4-((dialkylamino)methyl)-5-thioxo-4,5-dihydro-1,3,4-oxadiazol-2-yl)methyl)-1H-benzo[d]imidazol-2-yl)methyl amino)-N-(pyrimidin-2-yl)benzenesulfonamide (4a-e). The synthetic approach is shown in scheme-1.

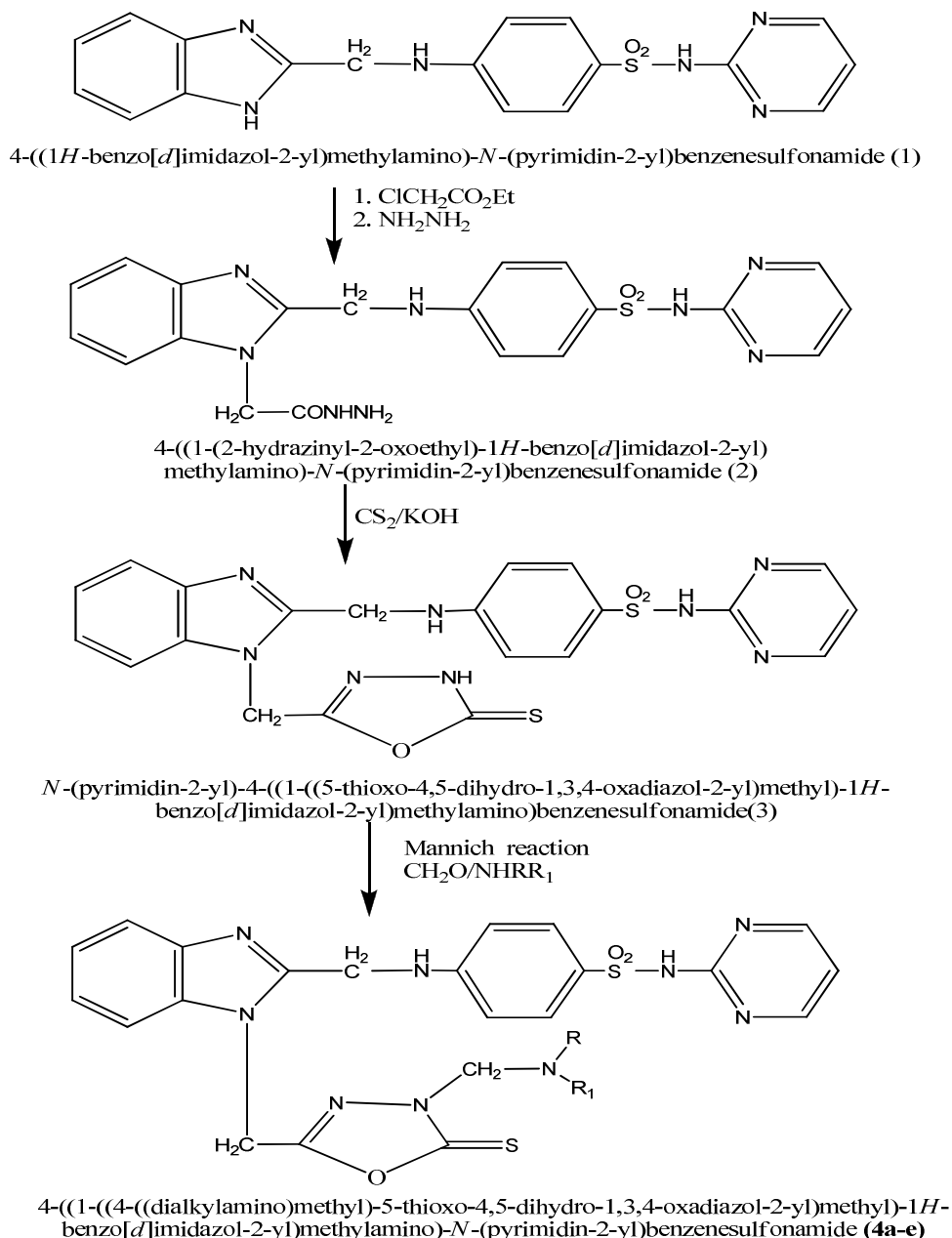
MATERIALS AND METHODS

Materials:

All the chemicals are used analytical grade. Sulfapyrimidine prepared by reported method. [17] 4-((1H-benzo[d]imidazol-2-yl)methyl amino)-N-(pyrimidin-2-yl)benzene sulfonamide (1) was prepared by method reported [18].

Measurement:

The IR spectra were recorded in KBr pellets on a Nicolet 400D spectrometer and ^1H NMR spectra were recorded in DMSO with TMS as internal standard on a Bruker spectrometer at 400 MHz and 100 MHz, respectively. Elemental analysis was carried out by Thermo finnigan CHN analyzer (Italy). Melting points were determined in open capillary tubes and were uncorrected.



Where,

	4a	4b	4c	4d	4e
R	CH_3	CH_3	Et	Et	Ph
R1	CH_3	Et	Et	Ph	Ph

SCHEME - 1

Preparation of 4-((1*H*-benzo[*d*]imidazol-2-yl)methylamino)-*N*-(pyrimidin-2-yl)benzene sulfonamide (1):-

2-(chloromethyl)-1*H*-benzo[*d*]imidazole (0.001 mole) was added to suspension of the Sulfapyrimidine (0.001 mole) and anhydrous potassium carbonate (0.005 mole) in ethyl methyl ketone (15 ml). The reaction mixture was stirred for 7 – 8.5 hrs at ambient temperature and ethyl methyl ketone was then evaporated. Distilled water was added to the residue and the formed precipitate was filtered, washed with water. The excess of solvent was removed and the product crystallized from methanol to give compound (1) yield is about 69%, m.p. 168°C. IR cm^{-1} : 3150(NH)1620-1648(C=N), 3020-3080(C-H, of Ar.), 2950, 2885, 1370(CH_2), 1175, 1390(SO_2). ^1H NMR: 7.21–7.68(m, 8H, Ar-H), 6.96-

8.32(m,3H,PyrimidoneAr-H),5.64-5.78(s,2H,NH),4.46 (s,2H, CH₂). Anal. Calcd for C₁₈H₁₆N₆O₂S (380): C, 56.83; H, 4.24; N, 22.09; S, 8.43. Found: C, 56.81; H, 4.22; N, 22.07; S, 8.41.

Preparation of 4-((1-(2-hydrazinyl-2-oxoethyl)-1H-benzo[d]imidazol-2-yl)methylamino)-N-(pyrimidin-2-yl)benzenesulfonamide (2):-

A solution of 4-((1H-benzo[d]imidazol-2-yl)methylamino)-N-(pyrimidin-2-yl)benzene sulfonamide (1) (0.1 mole) in the dry acetone (60 ml) and ethylchloroacetate (0.1 mole) in the presence of anhydrous K₂CO₃ (5 g) was refluxed for 10-12 hrs., cooled and the solid thus obtained was filtered, dried and crystallized from ethanol yield is about 66%. m.p. 234°C, and this compound (0.05 mole) and hydrazine hydrate (0.05mole) in 1,4-dioxane (35 ml) was refluxed on heating coil for 6-7 hrs. The excess of solvent was removed and the product crystallized from methanol to give (2), yield is about 69%, m.p.209°C. IR cm⁻¹: 3350-3150 (NH₂,NH)1620-1648(C=N),3020-3080(C-H of Ar.),2950,2885,1370(CH₂),1660-1670(CONH),1175,1390(SO₂).¹HNMR:7.21-7.68(m,8H,Ar-H),6.96-8.32(m,3H,PyrimidoneAr-H), 3.6(s,2H,NH₂),5.64-5.78(s,2H,NH), 4.86-4.38(s,4H,CH₂),7.8 (s,1H, CONH). Anal. Calcd for C₂₀H₂₀N₈O₃S(452): C, 53.09; H, 4.46; N, 24.76; S, 7.09. Found: C, 53.07; H, 4.45; N, 24.74; S, 7.08.

Preparation of N-(pyrimidin-2-yl)-4-((1-((5-thioxo-4,5-dihydro-1,3,4-oxadiazol-2-yl) methyl) -1H-benzo[d]imidazol-2-yl)methylamino)benzenesulfonamide(3):-

To a cold stirred solution of 4-((1-(2-hydrazinyl-2-oxoethyl)-1H-benzo[d]imidazol-2-yl)methylamino)-N-(pyrimidin-2-yl)benzenesulfonamide(2) (0.01 mole) in ethanol (50 ml) containing potassium hydroxide (0.01 mole), carbon disulphide (0.05 mole)was added gradually. The reaction mixture was heated under reflux on a steam-bath until hydrogen sulphide evolution ceased. Ethanol was removed by distillation under reduced pressure and the residue was stirred with water, filtered and the filtrate was neutralized with dilute hydrochloric acid. The product was filtered, washed with water and recrystallized from ethanol to get the compound N-(pyrimidin-2-yl)-4-((1-((5-thioxo-4,5-dihydro-1,3,4-oxadiazol-2-yl)methyl)-1H benzo[d]imidazol-2-yl)methylamino)benzenesulfonamide(3), which were obtained in 64% yield. IR cm⁻¹: 1620-1648(C=N), 3020-3080 cm⁻¹(C-H, of Ar.), 2950,2878, 1370 cm⁻¹(CH₂), 1185 (C=S),765(C-O-C ring), 1175,1390(SO₂).¹H NMR : 7.21-7.68(m, 8H, Ar-H), 6.96-8.32(m,3H,PyrimidoneAr-H),9.40 (s,1H, NH), 4.86-4.38,4.78 (s,4H,CH₂), 5.64-5.78(s,2H, NH). Anal. Calcd for C₂₁H₁₈N₈O₃S₂(494): C, 51.00; H, 3.67; N, 22.66;S, 12.97. Found: C, 50.95; H, 3.64; N, 22.65;S, 12.95.

Preparation of 4-((1-((4-((dialkylamino)methyl)-5-thioxo-4,5-dihydro-1,3,4-oxadiazol-2-yl)methyl)-1H-benzo[d]imidazol-2-yl)methylamino)-N-(pyrimidin-2-yl)benzenesulfonamide (4a-e):-

In a round bottom flask, the mixture of N-(pyrimidin-2-yl)-4-((1-((5-thioxo-4,5-dihydro-1,3,4-oxadiazol-2-yl)methyl)-1H-benzo[d]imidazol-2-yl)methylamino)benzene sulfonamide (3)(0.1mole) in THF (100ml), formaldehyde (0.1mole) and secondary amine (a-e) (0.12mol) was reflux on water bath for 3-3.5 hrs. The resultant mixture was concentrated, cooled, poured into ice-cold water, and then air-dried. Recrystallization from n-hexane gave 4-((1-((4-((dialkylamino)methyl)-5-thioxo-4,5-dihydro-1,3,4-oxadiazol-2-yl)methyl)-1H-benzo[d]imidazol-2-yl)methylamino)-N-(pyrimidin-2-yl)benzenesulfonamide (4a-e), which was obtained in 62-73% yield. The yields, melting points and other characterization data of these compounds are given in Table-1.

Table:-1 Analytical Data and Elemental Analysis of Compounds (4a-e)

Compd.	Molecular formula	Yield %	M.P. °C	Elemental Analysis							
				%C		%H		%N		%S	
				Found	Calcd.	Found	Calcd.	Found	Calcd.	Found	Calcd.
4a	C ₂₄ H ₂₅ N ₉ O ₃ S ₂	68	195	52.24	52.25	4.56	4.57	22.83	22.85	11.61	11.63
4b	C ₂₅ H ₂₇ N ₉ O ₃ S ₂	73	189	53.06	53.08	4.79	4.81	22.27	22.29	11.32	11.34
4c	C ₂₆ H ₂₉ N ₉ O ₃ S ₂	72	192	53.85	53.87	5.02	5.04	21.72	21.75	11.05	11.06
4d	C ₃₀ H ₂₉ N ₉ O ₃ S ₂	68	194	57.38	57.40	4.64	4.66	20.06	20.08	10.20	10.22
4e	C ₃₄ H ₂₉ N ₉ O ₃ S ₂	67	187	60.41	60.43	4.31	4.33	18.63	18.65	9.47	9.49

* Uncorrected

The structures assigned to 4-((1-((4-((dialkylamino)methyl)-5-thioxo-4,5-dihydro-1,3,4-oxadiazol-2-yl)methyl)-1H-benzo[d]imidazol-2-yl)methylamino)-N-(pyrimidin-2-yl)benzene- sulfonamide (4a-e) were supported by the elemental analysis , IR and NMR spectra showing an absorption bands at 1620-1648(C=N),3020-3080cm⁻¹(C-H of Ar.), 2950,2878,1370cm⁻¹(-CH₂),1185(C=S), 765 (C-O-C ring) 1175,1390(SO₂).¹HNMR: 7.21-7.68(m, 8H, Ar-H), 6.96-8.32(m,3H,PyrimidoneAr-H),4.86-4.38,4.78(s,4H,CH₂),5.64-5.78(s,2H,NH), 4a;2.17(s, 6H, CH₃), 4b; 2.26 (s,3H,CH₃), 1.08 (t,3H,CH₃), 2.67(q,2H,CH₂),4c;1.08(t,6H,CH₃), 2.67(q, 4H, CH₂),4d;1.08 (t,3H,CH₃),2.67 (q,2H,CH₂), 6.82-7.27(m,5H,Ar-H),4e;6.82-7.27(m, 10H, Ar-H).The C, H, N, S analysis data of all compounds are presented in Table-1.

BIOLOGICAL SCREENING**Antibacterial activities**

The antibacterial activities of all the compounds were studied against gram-positive bacteria (*Bacillus subtilis* and *Staphylococcus aureus*) and gram-negative bacteria (*klebsiella promioe* and *E.coli*) at a concentration of 50µg/ml by agar cup plate method. A methanol system was used as control in this method. Similar conditions using tetracycline as a control was used standard for comparison. The area of inhibition of zone measured in cm. The antibacterial activities of all the compounds are shown in Table-2.

Table:-2 Antibacterial Activity of Compounds (4a-e)

Compounds	Gram +Ve		Gram -Ve	
	<i>Bacillus subtilis</i>	<i>Staphylococcus aureus</i>	<i>Klebsiella promioe</i>	<i>E.coli</i>
4a	55	49	68	62
4b	54	51	67	61
4c	57	50	59	66
4d	65	59	70	72
4e	66	58	72	73
Tetracycline	68	60	77	80

Antifungal Activities

The fungicidal activity of all the compounds was studied at 1000 ppm concentration in vitro. Plant pathogenic organisms used were *Rhizopus nigricum*, *Nigrospora Sp.* and *Aspergillus niger*. The antifungal activities of all the compounds (4a-e) were measured on each of these plant pathogenic strains on a potato dextrose agar (PDA) medium. Such a PDA medium contained potato 200g, dextrose 20g, agar 20g and water 1c. Five days old cultures were employed. The compounds to be tested were suspended (1000ppm) in a PDA medium and autoclaved at 120° C for 15 min. at 15atm. pressure. These media were poured into sterile Petri plates and the organisms were inoculated after cooling the Petri plates. The percentage inhibition for fungi was calculated after five days using the formula given below:

$$\text{Percentage of inhibition} = 100(X-Y) / X$$

Where, X = Area of colony in control plate

Y = Area of colony in test plate

The fungicidal activity displayed by various compounds (4a-e) is shown in Table-3.

Table:-3 Antifungal Activity of Compounds (4a-e)

Compounds	Zone of Inhibition at 1000 ppm (%)		
	<i>Rhizopus Nigricum</i>	<i>Nigrospora Sp.</i>	<i>Aspergillus Niger</i>
4a	61	66	65
4b	59	65	63
4c	65	64	62
4d	69	72	66
4e	72	70	69

RESULTS AND DISCUSSION

The examination of elemental analytical data reveals that the elemental contents are consistence with the predicted structure shown in Scheme-1. The IR and NMR data also direct for assignment of the predicted structure.

The antibacterial activities of all the compounds were studied against gram-positive bacteria (*Bacillus subtilis* and *Staphylococcus aureus*) and gram-negative bacteria (*klebsiella promioe* and *E.coli*). All compounds were found toxic for Bacteria. Compounds 4d and 4e were found more toxic , Other compounds found to be less active than tetracycline are shown in Table-2.

The fungicidal activity of all the compounds was studied at 1000 ppm concentration in vitro. Plant pathogenic organisms used were *Rhizopus nigricum*, *Nigrospora Sp.* and *Aspergillus niger*. The percentage inhibition for fungi was calculated after five days using the formula given. The fungicidal activity displayed by various compounds (4a-e) is shown in Table-3. Compounds 4d and 4e were found more active, other compounds found to be less or moderate active.

CONCLUSION

The novel heterocyclic compound i.e. Oxadiazole-benzimidazole with pyrimidine fused derivatives (4a-e) were successfully synthesized. All the synthesized compounds structure were established on basis of analytical and spectral data. The newly synthesized compounds were evaluated for their antibacterial and antifungal activities. Among all the synthesized compounds 4d and 4e showed more active as antibacterial and antifungal agent.

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