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Synthesis and biological evaluation of friedlander annulation approach for the diversity oriented of functionalized quinolines

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ABSTRACT

Synthesis, spectral analysis and bioactivity of new inexpensive and diversity oriented method for the synthesis of various polysubstituted quinolines through the condensation of *o*-amino aryl carbonyls with ketones containing an active methylene group. We have synthesized twenty Quinoline derivatives were synthesized in moderate to good yields. The structures of all the newly synthesized molecules were assigned by elemental analysis and spectral data. The synthesized compounds were screened for their antibacterial activities strains using Cup plate method.

Keywords: Click chemistry, functionalized quinolines, Suzuki coupling reaction, triethyl amine.

INTRODUCTION

Quinoline is a common structural unit found in many natural products with remarkable pharmacological properties. Members of this extraction have broad applications in medicinal chemistry. The classical methods for the synthesis of quinolines, such as the Skraup[1], Doebner-von Miller[2], Combes[3], Pfitzinger[4] Quinoline synthesis, require harsh reaction conditions and the yields are unsatisfactory in most cases. The Friedlander annulations is the simplest, most straightforward synthetic method for the synthesis of quinoline derivatives, to a great extent for the highly substituted 3-quinolinecarboxylic esters [5]. This method usually involves acid- or base-catalyzed or thermal (up to 250°C) condensation between a 2-aminoaryl ketone or aldehyde and a second carbonyl compound possessing a reactive α -methylene group, followed by cyclodehydration.

Friedlander reported Quinoline synthesis in 1882 by the condensation of *o*-aminobenzaldehyde with acetaldehyde in the presence of sodium hydroxide. Acid catalysts such as hydrochloric acid, sulfuric acid, *p*-toluenesulfonic acid and phosphoric acids are extensively used for this transformation[6]. However, many of these classical methods require high temperatures, lengthy reaction times and extreme conditions and the yields reported are unsatisfactory due to the formation of different side reactions. Therefore, produced catalytic systems are being continuously explored in search of improved efficiencies and cost effectiveness [7]. However in recent times iodine[8], Lewis acids such as ZnCl₂ and AuCl₃·3H₂O[9], a combination of acidic catalysts [e.g., NaAuCl₄, Bi(OTf)₃, Nd(NO₃)₃·6H₂O][10], and microwave reactions[11]. Even some of these methods also suffer from harsh reaction conditions, low yields, high temperature, tedious work-up and the use of stoichiometric and relatively costly reagents[12]. Since Quinoline

derivatives are increasingly useful and important in drugs and pharmaceuticals, the development of easily understanding, suited and considerable yielding protocols are useful. As a part of our continuing efforts towards the development of useful synthetic activities[13], we planned to develop a new method for the synthesis of quinolines and its novel derivatives via Friedlander annulations approach catalyzed by copper sulphate. We observed the high efficiency of $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ (20 mol %) in sequential condensation / annulation reactions of *o*-amino aryl carbonyls and ketones containing an active methylene group for the synthesis of substituted quinolines.

MATERIALS AND METHODS

General Conditions: All the used reactants, reagents and solvents were obtained from commercial sources and were of analytical grade. Melting points were determined by open capillary method. ^1H NMR (CDCl_3 , 300 MHz) and ^{13}C NMR (CDCl_3 , 75 MHz) were recorded on spectrometer TMS as internal standard (chemical shifts and ppm). Mass spectra were recorded on a VG micromass 70-70H instrument. The purity of the compounds was checked by TLC on silica gel plates using a mixture of *n*-hexane and ethyl acetate.

I. General procedure for the synthesis of 6-chloro-N-(4-methoxyphenyl)-4-phenyl-2-((4-phenyl-1H-1,2,3-triazol-1-yl)methyl) quinoline-3-carboxamide (3a-e)

To a solution of quinoline-3-carboxylic acid **1** (0.20 g, 0.454 mmol) in dry CH_2Cl_2 (10 mL) was added EDCI (0.084 g, 0.545 mmol, dissolved in 3 mL) at 0°C under nitrogen atmosphere and stirred for 20 min. To this well stirred solution *p*-methoxy aniline (0.055 g, 0.454 mmol, dissolved in 3 mL CH_2Cl_2) was added slowly over 5 min at the same temperature and reaction mixture was allowed to stir for 6-8 h (monitored by TLC). After completion of reaction, reaction mixture was diluted with CH_2Cl_2 (50 mL) and washed with water (2x30 mL), dried over sodium sulphate and solvent was evaporated under reduced pressure resulted crude product, which was purified by silica gel column chromatography to get triazole coupled quinoline-3-carboxamide **3a** (0.19 g, 78 %). m. p. 220-223 $^\circ\text{C}$; IR (KBr): ν_{max} 3258, 3133, 3059, 3009, 2954, 2925, 2853, 1649, 1603, 1537, 1509, 1482, 1464, 1441, 1414, 1374, 1334, 1300, 1239, 1177, 1149, 1127, 1110, 1080, 1032, 960 cm^{-1} ; ^1H NMR (300 MHz, CDCl_3): δ 3.70 (s, 3H), 5.97 (s, 2H), 6.66 (d, 2H, $J = 8.87$ Hz), 6.98 (d, 2H, $J = 8.87$ Hz), 7.28-7.48 (m, 4H), 7.47-7.57 (m, 3H), 7.61 (d, 1H, $J = 2.07$ Hz), 7.64-7.75 (m, 4H), 8.01 (d, 1H, $J = 9.06$ Hz), 8.05 (s, 1H); ^{13}C NMR (75 MHz, CDCl_3): δ 54.44, 55.33, 113.92, 121.94, 122.80, 125.35, 125.65, 126.97, 128.09, 128.68, 128.92, 129.76, 129.23, 129.47, 130.22, 131.27, 131.53, 133.85, 134.27, 145.87, 146.01, 147.49, 151.02, 157.00, 164.97; ESI-MS: m/z 546 (M^+H); HRMS calculated for $\text{C}_{32}\text{H}_{25}\text{N}_3\text{O}_2\text{Cl}$ (M^+H) calculated 546.1691 found 546.1694.

This procedure was followed for the preparation of all triazole coupled quinoline-3-carboxamide **3(b-e)**.

6-chloro-4-phenyl-2-((4-phenyl-1H-1,2,3-triazol-1-yl)methyl)-N-(3-(trifluoromethyl)phenyl) quinoline-3-carboxamide (3b)

m.p. 231-233 $^\circ\text{C}$; IR (KBr): ν_{max} 3747, 3523, 3458, 3395, 3256, 3141, 3062, 3032, 2924, 2853, 1729, 1667, 1604, 1560, 1484, 1444, 1394, 1374, 1331, 1284, 1246, 1205, 1166, 1126, 1070, 1000, 961, 920 cm^{-1} ; ^1H NMR (300 MHz, CDCl_3): δ 5.96 (s, 2H), 7.17-7.39 (m, 6H), 7.39-7.57 (m, 6H), 7.58-7.66 (m, 3H), 7.70 (dd, 1H, $J = 9.06$ Hz, 2.26 Hz), 7.98-8.04 (m, 2H), 8.13 (bs, 1H); ^{13}C NMR (75 MHz, CDCl_3): δ 54.69, 113.97, 120.12, 120.39(7-C-Cl), 125.36, 125.57(-CF₃), 125.69, 126.64, 126.85, 128.02, 128.56, 128.70, 129.02, 129.18, 129.28, 130.15, 130.35, 130.77, 131.48, 132.45, 133.42, 134.54, 145.74, 146.51, 147.93, 148.16, 150.05, 165.84; ESI-MS: m/z 606 (M^+Na), 584 (M^+H); HRMS calculated for $\text{C}_{32}\text{H}_{22}\text{N}_3\text{OCIF}_3$ (M^+H) calculated 584.1459 found 584.1461.

6-chloro-4-phenyl-2-((4-phenyl-1H-1,2,3-triazol-1-yl)methyl)-N-(4-(trifluoromethoxy) phenyl) quinoline-3-carboxamide (3c)

m. p. 212-214 $^\circ\text{C}$; IR (KBr): ν_{max} 3306, 3135, 3063, 3018, 2925, 2854, 1727, 1657, 1609, 1534, 1509, 1482, 1443, 1412, 1394, 1329, 1249, 1214, 1160, 1110, 1080, 1047, 1018, 957, 920 cm^{-1} ; ^1H NMR (300 MHz, CDCl_3): δ 5.94 (s, 2H), 6.97 (d, 2H, $J = 8.49$ Hz), 7.18 (d, 2H, $J = 8.49$ Hz), 7.29-7.46 (m, 4H), 7.46-7.56 (m, 3H), 7.57-7.73 (m, 4H), 7.98 (d, 1H, $J = 8.87$ Hz), 8.02 (s, 1H), 8.27 (s, 1H); ^{13}C NMR (75 MHz, CDCl_3): δ 54.45, 121.43, 122.02, 121.91(-OCF₃), 125.38, 125.56, 126.95, 128.29, 128.55, 128.75, 128.91, 129.22, 129.32, 129.47, 129.92, 131.24, 131.66, 133.94, 134.11, 135.31, 145.89, 146.25, 147.50, 150.71, 165.26; ESI-MS: m/z 622 (M^+Na), 600 (M^+H); HRMS calculated for $\text{C}_{32}\text{H}_{22}\text{N}_3\text{O}_2\text{ClF}_3$ (M^+H) calculated 600.1408 found 600.1451.

6-chloro-N-(3-chloro-4-fluorophenyl)-4-phenyl-2-((4-phenyl-1H-1,2,3-triazol-1-yl)methyl)quinoline-3-carboxamide (3d)

m. p. 229-231 °C; IR (KBr): ν_{max} 3246, 3135, 3059, 3031, 2926, 2854, 1732, 1662, 1606, 1542, 1498, 1442, 1330, 1259, 1226, 1149, 1129, 1080, 1055, 1029, 961, 879, 834, 814 cm^{-1} ; ^1H NMR (300 MHz, CDCl_3): δ 5.94 (s, 2H), 6.81-6.88 (m, 2H), 7.28-7.45 (m, 6H), 7.48-7.56 (m, 3H), 7.60-7.68 (m, 3H), 7.69-7.74 (m, 1H), 7.89 (s, 1H), 8.02 (s, 1H), 8.05 (d, 1H, $J = 8.87$ Hz); ^{13}C NMR (75 MHz, CDCl_3): δ 54.63, 116.28, 116.46, 120.24, 120.29, 120.90, 121.04(-Cl), 122.14, 122.73, 125.39, 125.53, 126.95, 128.31, 128.74, 128.93, 129.21, 129.31, 129.35, 129.80, 131.31, 131.71, 133.33, 133.99, 134.12, 145.92, 146.22, 147.47, 150.79, 154.13, 156.09(-F), 165.24; ESI-MS: m/z 568 ($\text{M}^+\text{+H}$); HRMS calculated for $\text{C}_{31}\text{H}_{20}\text{N}_5\text{OCl}_2\text{F}$ ($\text{M}^+\text{+H}$) calculated 568.4325 found 568.4351.

6-chloro-N-(furan-2-ylmethyl)-4-phenyl-2-((4-phenyl-1H-1,2,3-triazol-1-yl)methyl)quinoline-3-carboxamide (3e)

m. p. 213-214 °C; IR (KBr): ν_{max} 3269, 3138, 3060, 3014, 2924, 2853, 1726, 1648, 1607, 1563, 1482, 1465, 1442, 1375, 1327, 1215, 1149, 1077, 1049, 1010, 975, 919, 881, 834, 810, 750, 695, 665, 640, 599 cm^{-1} ; ^1H NMR (300 MHz, CDCl_3): δ 4.24 (d, 2H, 5.28 Hz), $J = 5.77$ -5.87 (m, 1H), 5.90 (s, 2H), 6.10-6.22 (m, 1H), 6.36-6.50 (m, 1H), 7.14 (s, 1H), 7.27-7.70 (m, 10H), 7.76 (d, 2H, $J = 7.55$ Hz), 7.94 (d, 1H, $J = 9.06$ Hz), 8.07 (s, 1H); ^{13}C NMR (75MHz, CDCl_3): δ 36.55, 54.09, 107.57, 110.16, 121.43, 121.70, 125.29, 125.60, 126.94, 128.13, 128.76, 128.97/129.11, 129.78, 130.28, 131.14, 131.40, 133.73, 133.88, 142.05, 145.74, 145.85, 147.37, 149.75, 150.95, 166.40; ESI-MS: m/z 542 ($\text{M}^+\text{+Na}$), 520 ($\text{M}^+\text{+H}$); HRMS calculated for $\text{C}_{30}\text{H}_{23}\text{N}_5\text{O}_2\text{Cl}$ ($\text{M}^+\text{+H}$) calculated 520.1534 found 520.1543.

II. General procedure for the synthesis of ethyl 2-((4-benzoylpiperazin-1-yl) methyl)-6-chloro-4-phenylquinoline-3-carboxylate (6a-e).

To a mixture of Boc-protected product **4** (5.62 g, 25.5 mmol) and Et_3N (2.83 g, 28.0 mmol) in dry CH_2Cl_2 (20 mL) at 0°C under nitrogen atm. was added catalytic amount of DMAP and stirred for 30 min. To this reaction mixture benzoyl chloride (taken in 5 mL CH_2Cl_2) was added dropwise over 10 min. The resulting reaction mixture was further stirred for 6 h at room temperature, diluted with CH_2Cl_2 (50 mL) and water. Organic layer was separated and aqueous phase was extracted with CH_2Cl_2 (2x30 mL). The combined organic layers were washed with water, brine, dried over Na_2SO_4 , and concentrated under reduced pressure. The residue was purified by silica gel column chromatography, by eluting with 30% ethyl acetate/petroleum ether to afford 7.48 g of **6a** (98%) as a colorless solid, m.p. 239-241 °C; IR (KBr): ν_{max} 2976, 2928, 2865, 2809, 2312, 1723, 1691, 1637, 1565, 1481, 1457, 1423, 1365, 1340, 1315, 1285, 1259, 1241, 1222, 1163, 1103, 1082, 1071, 1001, 956, 645, 612, cm^{-1} ; ^1H NMR (300 MHz, CDCl_3): δ 0.86 (t, 3H, $J = 6.98$ Hz, CH_3), 2.35-2.68 (m, 4H, 2- CH_2), 3.22-3.42 (m, 2H, CH_2), 3.60-3.77 (m, 2H, CH_2), 3.91-4.06 (m, 4H, - COCH_2 + Ar- CH_2), 7.29-7.43 (m, 7H, ArH), 7.45-7.54 (m, 3H, ArH), 7.57 (d, 1H, ArH), 7.67 (dd, 1H, $J = 8.87$ Hz, $J = 2.26$ Hz, ArH), 8.04 (d, 1H, $J = 8.87$ Hz, ArH); ^{13}C NMR (75 MHz, CDCl_3): δ 13.46, 60.86, 63.35, 79.94, 125.26, 126.92, 127.42, 128.34, 128.64, 129.18, 129.60, 130.75, 131.08, 132.91, 134.95, 135.56, 145.36, 146.30, 154.59, 156.13, 167.82, 170.23; ESI-MS: m/z 536 ($\text{M}^+\text{+Na}$), 514 ($\text{M}^+\text{+H}$); HRMS calculated for $\text{C}_{30}\text{H}_{29}\text{N}_3\text{O}_3\text{Cl}$ ($\text{M}^+\text{+H}$) calculated 514.1892 found 514.1873.

This procedure was followed for the preparation of all piperazinamides of quinolines **6(b-e)**.

Ethyl 2-((4-(benzo[d][1,3]dioxole-5-carbonyl)piperazin-1-yl)methyl)-6-chloro-4-phenylquinoline-3-carboxylate (6b)

m.p. 219-221 °C; IR (KBr): ν_{max} 3061, 2982, 2922, 2811, 1721, 1632, 1565, 1482, 1458, 1441, 1400, 1342, 1315, 1281, 1247, 1221, 1169, 1141, 1103, 1081, 1070, 1037, 1000, 956, 931, 875, 833, 751, 701, 665, 646, 612, 570, cm^{-1} ; ^1H NMR (500 MHz, CDCl_3): δ 0.86 (t, 3H, $J = 7.17$ Hz, CH_3), 2.42-2.62 (m, 4H, 2- CH_2), 3.22 -3.72 (m, 4H, 2- CH_2), 3.93-4.04 (m, 4H, - COCH_2 + Ar CH_2), 5.98 (s, 2H, OCH_2O), 7.79 (d, 1H, $J = 7.93$ Hz, ArH), 6.88 (d, 1H, $J = 1.37$ Hz, ArH), 6.90 (dd, 1H, $J = 7.93$ Hz, $J = 1.37$ Hz, ArH), 7.31-7.36 (m, 2H, ArH), 7.49-7.53 (m, 3H, ArH), 7.57 (d, 1H, $J = 2.13$ Hz, ArH), 7.67 (dd, 1H, $J = 9.00$ Hz, $J = 2.88$ Hz, ArH), 8.04 (d, 1H, $J = 9.00$ Hz, ArH); ^{13}C NMR (75 MHz, CDCl_3): δ 63.43, 60.92, 52.74, 52.70, 13.53, 101.36, 108.02, 108.12, 121.53, 125.32, 127.02, 127.49, 128.36, 128.67, 129.25, 130.84, 131.15, 132.97, 135.04, 145.45, 146.36, 147.54, 148.77, 156.18, 167.89, 169.78; ESI-MS: m/z 558 ($\text{M}^+\text{+H}$); HRMS calculated for $\text{C}_{31}\text{H}_{29}\text{N}_3\text{O}_5\text{Cl}$ ($\text{M}^+\text{+H}$) calculated 558.1790 found 558.1772.

Ethyl-6-chloro-4-phenyl-2-((4-(4-(trifluoromethyl) benzoyl) piperazin-1-yl) methyl) quinoline -3-carboxylate (6c)

m.p. 230-232 °C; IR (KBr): ν_{max} 3063, 2984, 2919, 2812, 1722, 1637, 1566, 1481, 1463, 1444, 1426, 1400, 1330, 1255, 1220, 1165, 1126, 1070, 1025, 999, 956, 917, 883, 834664, 647, 610 cm^{-1} ; ^1H NMR (500 MHz, CDCl_3): δ 0.87 (t, 3H, $J = 7.17$ Hz, CH_3), 2.39-2.69 (m, 4H, 2 $-\text{CH}_2$), 3.22 -3.36 (m, 2H, CH_2), 3.64 -3.78 (m, 2H, CH_2), 3.94-4.05 (m, 4H, $-\text{COCH}_2 + \text{ArCH}_2$), 7.30-7.38 (m, 4H, ArH), 7.47-7.60 (m, 6H, ArH), 7.62-7.70 (m, 3H, ArH), 8.04 (d, 1H, $J = 9.00$ Hz, ArH); ^{13}C NMR (125 MHz, CDCl_3): δ 13.54, 42.18, 47.63, 52.41, 52.79, 60.94, 63.34, 124.02(- CF_3), 125.31, 126.45, 127.00, 127.43, 128.37, 128.70, 129.05, 129.22, 130.31, 130.83, 131.18, 133.01, 134.99, 136.40, 145.44, 146.43, 155.95, 167.87, 168.63; ESI-MS: m/z 582 ($\text{M}^+ + \text{H}$); HRMS calculated for $\text{C}_{31}\text{H}_{28}\text{N}_3\text{O}_3\text{ClF}_3$ ($\text{M}^+ + \text{H}$) calculated 582.1765 found 582.1750.

Ethyl-6-chloro-2-((4-(6-chlorobenzoyl)piperazin-1-yl)methyl)-4-phenylquinoline-3-carboxylate (6d)

m.p. 239-241 °C; ^1H NMR (300 MHz, CDCl_3): δ 0.87 (t, 3H, $J = 7.17$ Hz, CH_3), 2.41-2.69 (m, 4H, 2- CH_2), 3.25 -3.42 (m, 2H, CH_2), 3.61 -3.78 (m, 2H, CH_2), 3.93-4.01 (q, 2H, $J = 7.17$ Hz, CH_2), 4.02 (s, 2H, $-\text{COCH}_2$), 7.31-7.36 (m, 2H, ArH), 7.36-7.41 (m, 1H, ArH), 7.49-7.54 (m, 3H, ArH), 7.57 (d, 1H, $J = 2.13$ Hz, ArH), 7.68 (dd, 1H, $J = 8.85$ Hz, $J = 2.28$ Hz, ArH), 7.71 (dd, 1H, $J = 8.08$ Hz, $J = 2.28$ Hz, ArH), 8.05 (d, 1H, $J = 9.00$ Hz, ArH), 8.42 (d, 1H, $J = 1.98$ Hz, ArH); ^{13}C NMR (75 MHz, CDCl_3): δ 13.56, 42.24, 17.68, 52.37, 52.68, 60.94, 63.21, 124.22, 124.28, 125.32, 128.38, 128.73, 129.19, 130.15, 130.82, 131.25, 133.86, 134.95(- CCl), 137.86, 145.43, 146.50, 148.00, 148.10, 152.61, 155.75, 167.88; ESI-MS: m/z 549 ($\text{M}^+ + \text{H}$). HRMS calculated for $\text{C}_{30}\text{H}_{27}\text{Cl}_2\text{N}_3\text{O}_3$ ($\text{M}^+ + \text{H}$) calculated 549.1756 found 549.1750.

Ethyl 6-chloro-2-((4-(4-fluorobenzoyl) piperazin-1-yl) methyl)-4-phenylquinoline-3-carboxylate (6e)

m.p. 216-218 °C; IR (KBr): ν_{max} 2984, 2920, 2811, 1721, 1629, 1605, 1564, 1510, 1481, 1458, 1431, 1400, 1340, 1315, 1279, 1261, 1219, 1157, 1127, 1069, 999, 955, 882, 836, 747, 701, 664, 647, 612, 577 cm^{-1} ; ^1H NMR (300 MHz, CDCl_3): δ 0.86 (t, 3H, $J = 6.79$ Hz, CH_3), 2.32-2.72 (m, 4H, 2- CH_2), 3.20-3.82 (m, 4H, 2- CH_2), 3.91-4.11 (m, 4H, $-\text{COCH}_2 + \text{ArCH}_2$), 7.07 (t, 2H, $J = 8.30$ Hz, ArH), 7.30-7.46 (m, 4H, ArH), 7.47-7.63 (m, 4H, ArH), 7.68 (dd, 1H, $J = 9.06$ Hz, $J = 2.26$ Hz, ArH), 8.05 (d, 1H, $J = 9.06$ Hz, ArH); ^{13}C NMR (75 MHz, CDCl_3): δ 13.54, 52.59, 52.69, 60.93, 63.38, 115.41, 115.59, 125.32, 127.00, 127.45, 128.36, 128.69, 129.23, 129.28, 129.34, 130.82, 131.18, 131.59, 133.00, 135.00, 145.44, 146.40, 156.03, 162.35(- CF), 164.32, 167.87, 169.36; ESI-MS: m/z 532 ($\text{M}^+ + \text{H}$); HRMS (ESI) calculated for $\text{C}_{30}\text{H}_{28}\text{ClFN}_3\text{O}_3$ ($\text{M}^+ + \text{H}$) calculated 532.1797 found 532.1776.

III. General procedure for the synthesis of (E)-3-(4-bromophenyl)-1-(6-chloro-2-methyl-4-phenylquinolin-3-yl) prop-2-en-1-one (11a-e)

To a well stirred solution of acetylquinoline(9) (5 g, 16.94 mmol) and *p*-bromo-benzaldehyde (3.41 g, 18.64 mmol) in ethanol (50 mL) and water (10 mL) was added slowly 40% KOH solution over 2 min at room temperature, stirring was continued for 12 h. After completion of reaction, water (40 mL) was added to the reaction mixture and extracted with ethyl acetate (3x60 mL) and the collected organic layers were dried over Na_2SO_4 and evaporated under reduced pressure, to obtain crude residue which was purified by column chromatography using EtOAc/hexane (2:8) as an eluent to give yellow colored solid **11a** (6.87 g, 88 %). m.p. 214-216 °C; IR (KBr): ν_{max} 3609, 3542, 3395, 3059, 2957, 2922, 2852, 2310, 1906, 1709, 1671, 1646, 1622, 1601, 1582, 1562, 1482, 1441, 1385, 1362, 1312, 1278, 1220, 1197, 1179, 1157, 1125, 1073, 1046, cm^{-1} ; ^1H NMR (300 MHz, CDCl_3): δ 2.60 (s, 3H, ArCH_3), 6.47 (d, 1H, $J = 16.24$ Hz, =CH), 6.97 (d, 1H, $J = 16.24$ Hz, =CH), 7.11 (d, 2H, $J = 8.30$ Hz, ArH), 7.16-7.24 (m, 2H, ArH), 7.26-7.41 (m, 6H, ArH), 7.49 (d, 1H, $J = 2.26$ Hz, ArH), 7.60 (dd, 1H, $J = 8.87$ Hz, 2.26 Hz, ArH), 7.97 (d, 1H, $J = 8.87$ Hz, ArH); ^{13}C NMR (75 MHz, CDCl_3): δ 23.91, 125.04, 125.41, 126.01, 127.87, 128.63, 128.97, 129.62, 129.87, 130.51, 131.12, 132.21, 132.53, 132.79, 133.16, 134.45, 144.65, 145.08, 146.12, 155.27, 196.84; ESI-MS: m/z 462 ($\text{M}^+ + \text{H}$).

This procedure was followed for the preparation of all chalcones of quinolines **11b-e**.

(E)-3-(benzo[d][1,3]dioxol-5-yl)-1-(6-chloro-2-methyl-4-phenylquinolin-3-yl)prop-2-en-1-one (11b)

m.p. 199-201 °C; IR (KBr): ν_{max} 3062, 3012, 2900, 2782, 1664, 1640, 1620, 1598, 1580, 1501, 1485, 1446, 1389, 1359, 1340, 1309, 1255, 1214, 1158, 1125, 1102, 1079, 1036, 976, 929, 880, 862, 834, 807, 752, 704 cm^{-1} ; ^1H NMR (300 MHz, CDCl_3): δ 2.69 (s, 3H, ArCH_3), 5.99 (s, 2H, $-\text{OCH}_2\text{O}$), 7.42 (d, 1H, $J = 16.17$ Hz, =CH), 6.75 (d, 1H, $J = 8.08$ Hz, ArH), 6.78-6.85 (m, 2H, ArH), 7.00 (d, 1H, $J = 16.17$ Hz, =CH), 7.27-7.31 (m, 2H, ArH), 7.38-7.45 (m, 3H, ArH), 7.58 (d, 1H, $J = 2.28$ Hz, ArH), 7.68 (dd, 1H, $J = 9.00$ Hz, 2.28 Hz, ArH), 8.06 (d, 1H, $J = 9.00$ Hz, ArH); ^{13}C NMR (125 MHz, CDCl_3): δ 23.87, 101.69, 106.44, 108.57, 125.03, 125.36, 125.63, 126.09, 128.27, 128.52,

128.81, 129.79, 130.43, 130.93, 132.37, 133.42, 134.48, 144.47, 145.98, 146.81, 148.38, 150.34, 155.34, 196.92; ESI-MS: m/z 428 (M^+H).

(E)-1-(6-chloro-2-methyl-4-phenylquinolin-3-yl)-3-(4-(trifluoromethyl) phenyl) prop-2-en-1-one (11c)

m.p. 181-183 °C; IR(KBr): ν_{max} 3063, 3019, 2926, 2853, 1650, 1608, 1565, 1480, 1443, 1414, 1387, 1323, 1280, 1218, 1201, 1166, 1126, 1067, 1046, 1015, 981, 919, 860, 832, 756, 706 cm^{-1} ; 1H -NMR (300 MHz, $CDCl_3$): δ 2.70 (s, 3H, $ArCH_3$), 6.60 (d, 1H, $J = 16.24$ Hz, =CH), 7.12 (d, 1H, $J = 16.24$ Hz, =CH), 7.20-7.83 (m, 10H, ArH), 7.70 (dd, 1H, $J = 8.87$ Hz, 1.7 Hz, ArH), 8.08 (d, 1H, $J = 8.87$ Hz, ArH); ^{13}C NMR (125 MHz, $CDCl_3$): δ 18.87, 101.69, 106.44, 118.57, 125.03, 125.36, 125.63, 126.09, 128.27, 128.52, 128.81, 129.72, 130.43, 130.93, 131.37, 133.42, 134.48, 144.47, 145.28, 146.81, 148.38, 150.34, 157.34, 187.0; ESI-MS: m/z 452 (M^+H).

(E)-1-(6-chloro-2-methyl-4-phenylquinolin-3-yl)-3-(naphthalen-1-yl)prop-2-en-1-one (11d) m.p. 186-188 °C; IR (KBr): ν_{max} 3058, 3014, 2956, 2924, 2853, 1698, 1644, 1598, 1566, 1510, 1479, 1442, 1388, 1346, 1310, 1277, 1240, 1217, 1202, 1157, 1125, 1079, 1049, 1029, 1003, 973, 704, 665, 624, 589 cm^{-1} ; 1H NMR (300 MHz, $DMSO-d_6$): δ 2.77 (s, 3H, ArH_3), 6.86 (d, 1H, $J = 16.05$ Hz, =CH), 7.33-7.57 (m, 9H, ArH), 7.64 (d, 1H, $J = 2.07$ Hz, ArH), 7.70 (dd, 1H, $J = 8.87$ Hz, 2.26 Hz, ArH), 7.74-7.92 (m, 3H, ArH), 7.98 (d, 1H, $J = 16.05$ Hz, =CH), 8.10 (d, 1H, $J = 8.87$ Hz, ArH); ^{13}C NMR (75 MHz, $CDCl_3$): δ 23.96, 122.87, 124.85, 125.08, 125.32, 125.51, 126.32, 127.07, 128.67, 128.81, 128.97, 129.98, 130.06, 130.50, 131.11, 131.25, 132.51, 133.55, 134.47, 143.37, 144.66, 146.09, 155.36, 1196.96; ESI-MS: m/z 434 (M^+H).

(E)-1-(6-chloro-2-methyl-4-phenylquinolin-3-yl)-3-(4-(dimethylamino) phenyl) prop-2-en-1-one (11e)

m.p. 207-208 °C; IR (KBr): ν_{max} 2922, 2853, 1631, 1586, 1523, 1480, 1436, 1367, 1334, 1311, 1269, 1225, 1180, 1165, 1123, 1079, 1046, 1030, 1016, 977, 945, 879, 833, 814, 756, 703, 665, 599 cm^{-1} ; 1H NMR (300 MHz, $CDCl_3$): δ 2.70 (s, 3H, $ArCH_3$), 3.02 (s, 6H, 2- CH_3), 6.45 (d, 1H, $J = 16.13$ Hz, =CH), 6.59 (d, 2H, $J = 8.80$ Hz, ArH), 7.00 (d, 1H, $J = 16.13$ Hz, =CH), 7.19-7.33 (m, 4H, ArH), 7.35-7.44 (m, 3H, ArH), 7.56 (d, 1H, $J = 2.20$ Hz, ArH), 7.66 (dd, 1H, $J = 8.80$ Hz, 2.20 Hz, ArH), 8.06 (d, 1H, $J = 8.80$ Hz, ArH); ^{13}C NMR (75 MHz, $CDCl_3$): δ 23.32, 39.53, 111.52, 120.94, 121.65, 124.25, 125.96, 128.20, 128.51, 129.50, 130.19, 130.68, 130.81, 130.94, 133.70, 134.35, 143.29, 145.24, 148.84, 152.15, 155.08, 195.62; ESI-MS: m/z 427 (M^+H).

IV. General procedure for the synthesis of (E)-3-([1,1'-biphenyl]-4-yl)-1-(6-chloro-2-methyl-4-phenylquinolin-3-yl)prop-2-en-1-one (13a-e)

To a stirred solution of compound **11a** (0.2 g, 0.4 mmol) and phenyl boronic acid (0.052g, 0.4 mmol) in toluene: EtOH (10 ml) were added $Pd(PPh_3)_4$ (0.046 g, 0.04 mmol), K_2CO_3 (0.11 g, 0.8 mmol) and reaction mixture was heated at 90 °C for 10 h. Reaction mixture was worked up with water and extracted with ethyl acetate (3 x 20 ml). Combined ethyl acetate layer was dried over sodium sulphate, concentrated under reduced pressure to give solid product which was purified by silica gel chromatography to obtain compound **13a** (0.163 g, 89%). m.p. 222-224 °C; IR (KBr): ν_{max} 3059, 3029, 2924, 2852, 1699, 1643, 1620, 1596, 1557, 1516, 1482, 1445, 1409, 1388, 1338, 1312, 1278, 1219, 1185, 1158, 1124, 1079, 1046, 1005, 980, 946, 832, 761, 701, 667, 645 cm^{-1} ; 1H -NMR (300 MHz, $CDCl_3$): δ 2.71 (s, 3H, $ArCH_3$), 6.63 (d, 1H, $J = 16.61$ Hz, =CH), 7.14 (d, 1H, $J = 16.61$ Hz, =CH), 7.28-7.50 (m, 10H, ArH), 7.52-7.63 (m, 5H, ArH), 7.69 (dd, 1H, $J = 9.06$ Hz, 2.26 Hz, ArH), 8.07 (d, 1H, $J = 9.06$ Hz, ArH); ^{13}C -NMR (125 MHz, $CDCl_3$): δ 23.91, 125.06, 126.11, 126.97, 127.34, 127.54, 128.02, 128.58, 128.90, 129.86, 130.50, 131.00, 132.43, 132.80, 133.34, 134.48, 139.78, 143.82, 144.58, 146.08, 146.45, 155.32, 197.08; ESI-MS: m/z 460 (M^+H); HRMS calculated for $C_{31}H_{23}NOCl$ (M^+H) calculated 460.1462 found 460.1465.

(E)-1-(6-chloro-2-methyl-4-phenylquinolin-3-yl)-3-(4'-chloro-[1,1-biphenyl]-4-yl)prop-2-en-1-one (13b)

m.p. 232-234 °C; 1H NMR (300 MHz, $CDCl_3$): δ 2.71 (s, 3H, $ArCH_3$), 6.63 (d, 1H, $J = 16.43$ Hz, =CH), 7.13 (d, 1H, $J = 16.43$ Hz, =CH), 7.28-7.65 (m, 14H, ArH), 7.69 (dd, 1H, $J = 9.25$ Hz, 2.45 Hz, ArH), 8.07 (d, 1H, $J = 9.25$ Hz, ArH); ^{13}C NMR (75 MHz, $CDCl_3$): δ 23.93, 125.04, 126.06, 127.36, 127.52, 128.17, 128.22, 128.58, 128.88, 128.95, 129.07, 129.86, 130.51, 131.01, 132.43, 133.11, 133.27, 134.18, 134.46, 138.20, 142.44, 144.56, 146.10, 155.29, 197.04; ESI-MS: m/z 494 (M^+H); HRMS calculated for $C_{31}H_{22}NOCl_2$ (M^+H) calculated 494.1073 found 494.1068.

(E)-1-(6-chloro-2-methyl-4-phenylquinolin-3-yl)-3-(3'-fluoro-[1,1'-biphenyl]-4-yl)prop-2-en-1-one (13c)

m.p. 212-214 °C; IR (KBr): ν_{max} 3062, 3030, 2958, 2924, 2852, 1644, 1589, 1558, 1517, 1480, 1442, 1388, 1338, 1313, 1292, 1259, 1220, 1205, 1186, 1158, 1124, 1079, 1046, 1010 cm^{-1} ; 1H NMR (300 MHz, $CDCl_3$): δ 2.71 (s, 3H, $ArCH_3$), 6.63 (d, 1H, $J = 16.61$ Hz, =CH), 7.02-7.08 (m, 1H, ArH), 7.13 (d, 1H, $J = 16.61$ Hz, =CH), 7.27-7.49

(m, 11H, ArH), 7.55 (d, 2H, $J = 8.30$ Hz, ArH), 7.60 (d, 1H, $J = 2.26$ Hz, ArH), 7.70 (dd, 1H, $J = 9.06$ Hz, 3.02 Hz, ArH), 8.07 (d, 1H, $J = 9.06$ Hz, ArH); ^{13}C NMR (75 MHz, CDCl_3): δ 2396, 114.68, 114.97, 122.59, 122.62, 125.06, 126.07, 127.53, 127.63, 128.60, 128.91, 128.94, 129.87, 130.36, 130.47, 130.54, 131.03, 132.45, 133.26, 133.37, 134.47, 144.57, 146.07, 146.13, 155.31, 161.51, 197.09; ESI-MS: m/z 478 (M^+H); HRMS calculated for $\text{C}_{31}\text{H}_{22}\text{NOClF}$ (M^+H) calculated 478.1368 found 478.1366.

(E)-1-(6-chloro-2-methyl-4-phenylquinolin-3-yl)-3-(4'-hydroxy-[1,1'-biphenyl]-4-yl)prop-2-en-1-one (13d)

m.p. 202-204 °C; ^1H NMR (300 MHz, CDCl_3): δ 2.72 (s, 3H, ArCH_3), 6.62 (d, 1H, $J = 16.61$ Hz, =CH), 6.91 (d, 2H, $J = 9.06$ Hz, ArH), 7.12 (d, 1H, $J = 16.61$ Hz, =CH), 7.28-7.56 (m, 11H, ArH), 7.60 (d, 1H, $J = 2.26$ Hz, ArH), 7.69 (dd, 1H, $J = 9.06$ Hz, 2.26 Hz, ArH), 8.09 (d, 1H, $J = 9.06$ Hz, ArH); ^{13}C NMR (75 MHz, CDCl_3): δ 23.53, 115.97, 125.16, 126.25, 126.85, 126.93, 128.28, 128.62, 128.99, 129.80, 129.96, 130.22, 131.27, 131.92, 131.98, 132.67, 133.50, 134.30, 143.63, 145.08, 145.61, 147.00, 155.40, 156.45, 196.99; ESI-MS: m/z 476 (M^+H); HRMS calculated for $\text{C}_{31}\text{H}_{23}\text{NO}_2\text{Cl}$ (M^+H) calculated 476.1411 found 476.1421.

(E)-1-(6-chloro-2-methyl-4-phenylquinolin-3-yl)-3-(4'-methoxy-[1,1'-biphenyl]-4-yl)prop-2-en-1-one (13e)

m.p. 225-227 °C; IR (KBr): ν_{max} 3060, 3028, 3009, 2956, 2928, 2837, 1641, 1596, 1556, 1525, 1497, 1480, 1442, 1388, 1296, 1275, 1248, 1225, 1181, 1158, 1124, 1080, 1042, 999, 980, 705, 668, 644 cm^{-1} ; ^1H NMR (300 MHz, CDCl_3): δ 2.71 (s, 3H, ArCH_3), 3.85 (s, 3H, ArOCH_3), 6.62 (d, 1H, $J = 16.05$ Hz, =CH), 6.97 (d, 2H, $J = 8.68$ Hz, ArH), 7.12 (d, 1H, $J = 16.05$ Hz, =CH), 7.28-7.47 (m, 7H, ArH), 7.48-7.56 (m, 4H, ArH), 7.59 (d, 1H, $J = 2.26$ Hz, ArH), 7.69 (dd, 1H, $J = 9.06$ Hz, 2.26 Hz, ArH), 8.07 (d, 1H, $J = 9.06$ Hz, ArH); ^{13}C NMR (125 MHz, CDCl_3): δ 23.93, 55.33, 114.34, 125.06, 126.12, 126.97, 127.02, 128.05, 128.56, 128.85, 128.93, 129.85, 130.51, 130.95, 132.14, 132.18, 132.40, 133.37, 134.49, 143.45, 144.54, 146.09, 146.65, 155.33, 159.73, 197.12; ESI-MS: m/z 490 (M^+H). HRMS calculated for $\text{C}_{32}\text{H}_{24}\text{O}_2\text{ClN}$ (M^+H) calculated 490.1471 found 490.1491.

RESULTS AND DISCUSSION

To a solution of quinoline-3-carboxylic acid **1** in dry CH_2Cl_2 was added EDCI at 0°C under nitrogen atmosphere and stirred for 20 min. To this well stirred solution *p*-methoxy aniline **2** was added slowly over 5 min at the same temperature and reaction mixture was allowed to stir for 8 h. After completion of reaction, reaction mixture was diluted with CH_2Cl_2 and washed with water, dried over sodium sulphate and solvent was evaporated under reduced pressure resulted crude product 75-78%, which was purified by silica gel column chromatography to get triazole coupled quinoline-3-carboxamide [14] **3a**. Physical properties of **3(a-e)** are tabulated in Table-1

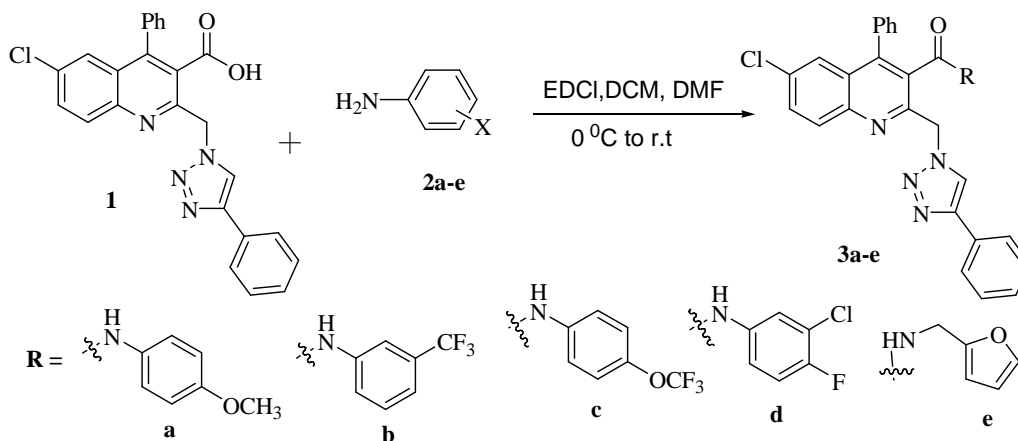


Table-1 Physical data of synthesized derivatives (3a-e)

S.No	compound	Time (hours)	Yield (%)	Color	M.p.°c
1	3a	8	78	White solid	220-223
2	3b	6	75	White solid	231-233
3	3c	6	76	White solid	212-214
4	3d	7-8	78	White solid	229-231
5	3e	6-7	75	White solid	213-214

To a mixture of Boc-protected product **4** and Et₃N in dry CH₂Cl₂ at 0°C under nitrogen atmosphere was added catalytic amount of DMAP and stirred for 30 min. To this reaction mixture benzoyl chloride **5** was added drop wise over 10 min. The resulting reaction mixture was further stirred for 6hrs at room temperature[15], diluted with CH₂Cl₂ and water. Organic layer was separated and aqueous phase was extracted with CH₂Cl₂. The combined organic layers were washed with water and brine solution, dried over Na₂SO₄, and concentrated under reduced pressure. The residue was purified by silica gel column chromatography, by eluting with 30% ethyl acetate/petroleum ether to afford to **6a** (95-98%), Physical properties of **6(a-e)** are tabulated in Table-2.

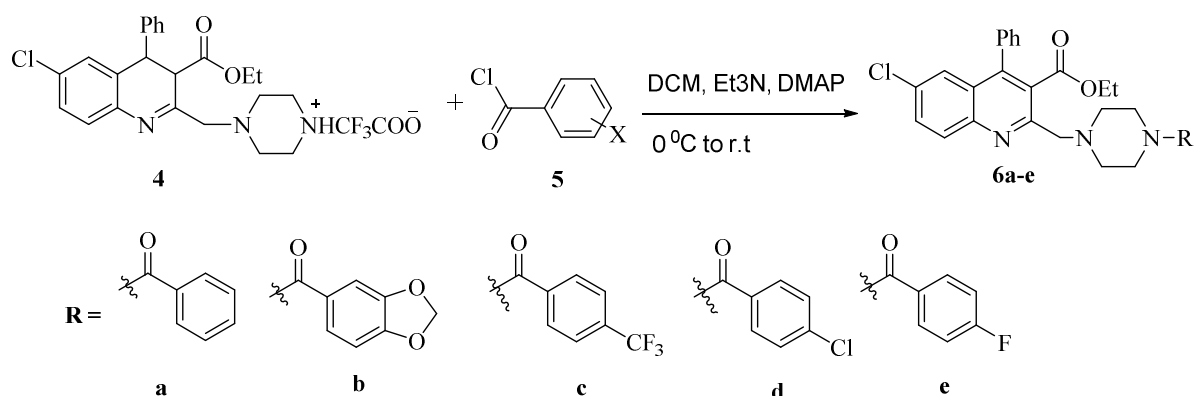
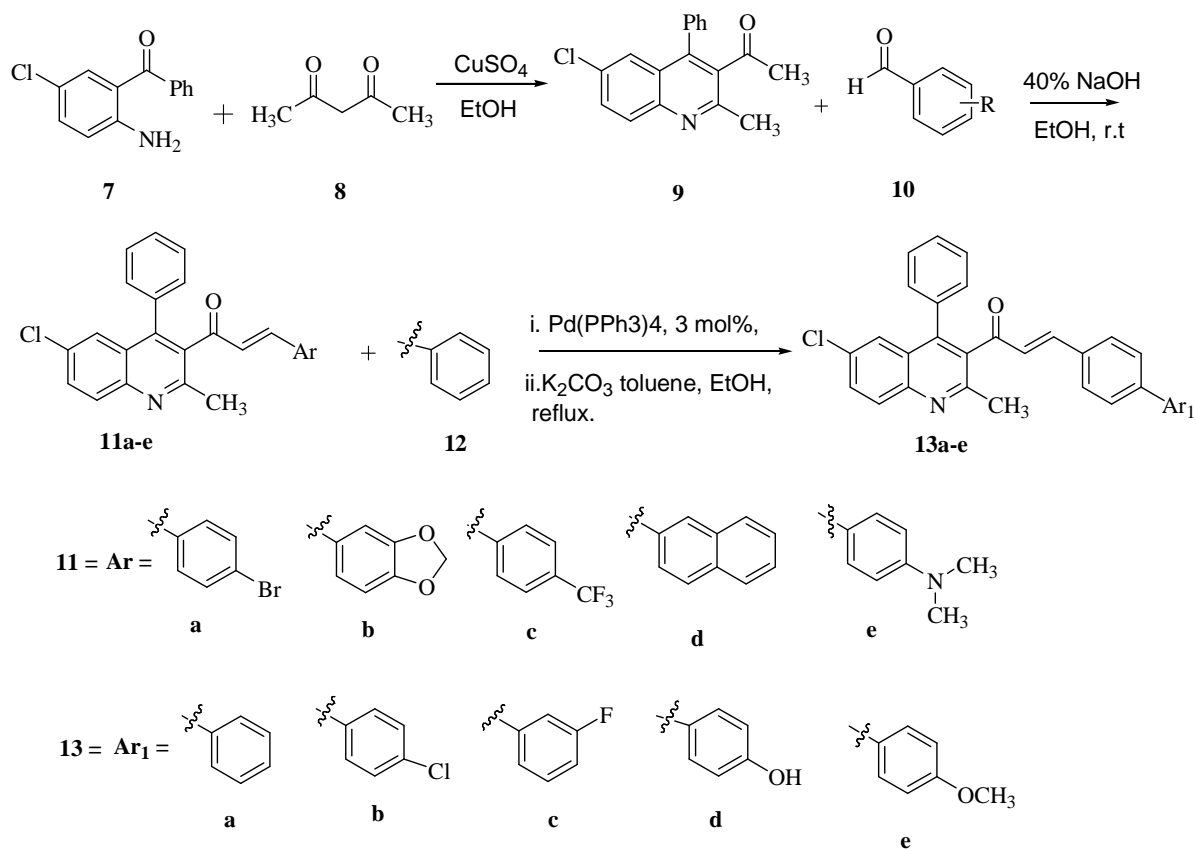


Table-2 physical data of synthesized derivatives (6a-e)

S.No	compound	Time (hours)	Yield (%)	Colour	M.p ^o c
1	6a	6	98	Colorless solid	239-241
2	6b	6	95	Colorless solid	219-221
3	6c	5	96	Colorless solid	230-232
4	6d	6	98	Colorless solid	239-241
5	6e	6	95	Colorless solid	216-218

To a well stirred solution of acetylquinoline **9** and *p*-bromo-benzaldehyde **10** in ethanol and water was added slowly 40% KOH solution over 2 min at room temperature, stirring was continued for 12 h. After completion of reaction, water was added to the reaction mixture and extracted with ethyl acetate and the taken organic layers were dried over Na₂SO₄ and evaporated under reduced pressure, to obtain crude residue, it was purified by column chromatography using Ethyl acetate/hexane (2:8) as an eluent to give yellow colored solid **11** (85-88%). To a stirred solution of compound **11** and phenyl boronic acid **12** in toluene:EtOH were added Pd(PPh₃)₄, K₂CO₃ and reaction mixture was heated at 90°C for 10 h. Reaction mixture was worked up with water and extracted with ethyl acetate. Combined ethyl acetate layer was dried over sodium sulphate, concentrated under reduced pressure to give solid product it was purified by silica gel chromatography to obtain compound **13** (85-89%). Physical properties of **11(a-e)** and **13(a-e)** are tabulated in Table-3.

**Table-3** physical data of synthesized derivatives (11a-e and 13a-e)

S.No	compound	Time (hours)	Yield (%)	Colour	M.p ^o c
1	11a	12	88	yellow color solid	214-216
2	11b	12	85	yellow color solid	199-201
3	11c	11	86	yellow color solid	181-2183
4	11d	12	88	yellow color solid	186-188
5	11e	12	85	yellow color solid	207-208
6	13a	10	89	yellow color solid	222-224
7	13b	10	88	yellow color solid	232-234
8	13c	9	85	yellow color solid	212-214
9	13d	10	88	yellow color solid	202-204
10	13e	9	85	Yellow color solid	225-227

Antibacterial Activity:

It is well known that Quinoline derivatives possess antibacterial activity[17-20]. Furthermore, halogen substituent was introduced into the basic structure anticipating an improvement of biological activity because their incorporation was proved to be influencing the biological activity in various heterocycles as well as coumarins[21]. Electron withdrawing groups like halogens will increase bactericidal potential as they alter the nature of the compound in such a way as to promote binding to the target(s)[22]. According to Rajendra Prasad et al in designing the compounds bearing electron withdrawing substituents (with high degree of binding linearity) results in high molecular weights to exhibit an improved antibacterial activity [23]. Similarly, the significant inhibition shown by substituted Quinoline was attributed to substituents like hydroxymethyl / hydroxy / methoxy / ethyl ester groups[24]. Tejaskumar Shah and Vikas Desai also reported that the presence of methoxy-, chloro-, and fluoro groups enhanced the antibacterial activity in isoxazoline derivatives [25].

The antibacterial screening data showed that most of the compound 3a-e, 6a-e, 11a-e and 13a-e showed moderate to excellent activities against the used microorganisms (**Table-4**) compared to the reference drug as *Chloramphenicol*.

These results suggested that the introduction of halogen substituent increased the hydrophobicity of the synthesized compounds and lead to the increase of the antibacterial activity [26]. In the present study, it is observed that good activity was shown by the prepared derivatives against the studied Gram positive bacteria and very poor activity against Gram negative bacteria. The good and poor antibacterial activities of the prepared derivatives against Gram positive and Gram negative bacteria can be explained based on their cell outer layers [27]. Gram positive bacteria have an ineffective and permeable outer barrier made of peptidoglycan layer, which is responsible for permeability of drug constituents. However, Gram negative bacteria have an impermeable outer membrane to drug constituents, as cell wall contains multilayered peptidoglycan and phospholipidic [28]. Among the compounds screened, 3b, 6a, 11c and 13c showed high activity. The observed antibacterial activity profile suggested that the presence of halogen functional group – bromine had enhanced the activity.

Table-4: Antibacterial activity

Compound	Zone of inhibition (mm)*					
	Escherichia coli (MTCC 40) (Gram-negative) (Conc. µg/ml)			Staphylococcus aureus (MTCC 96) (Gram-positive) (Conc. µg/ml)		
	200	100	50	200	100	50
3a	10	13	14	24	15	25
3b	24	27	21	14	15	15
3c	10	15	12	8	11	12
3d	8	6	18	11	6	18
3e	9	-	-	25	-	-
6a	25	22	32	15	22	32
6b	15	12	12	23	14	12
6c	23	22	21	19	22	21
6d	20	20	32	17	20	30
6e	13	12	12	20	12	12
11a	20	20	21	16	20	20
11b	11	-	-	22	-	-
11c	22	11	17	13	11	17
11d	15	15	20	15	10	11
11e	14	11	11	26	29	19
13a	12	6	8	3	6	9
13b	3	6	9	22	6	9
13c	22	18	19	15	18	20
13d	11	12	12	-	12	12
13e	15	11	14	24	25	20
Chloramphenicol	31	30	21	33	30	23

CONCLUSION

We have accomplished an efficient, convenient, and inexpensive and diversity oriented method for the synthesis of various polysubstituted quinolines through the condensation of o-amino aryl carbonyls with ketones containing an active methylene group by using catalytic amount of CuSO₄.5H₂O at room temperature. Further by applying this protocol we have synthesized the new derivatives of quinoline-2-yl-1, 2, 3-triazoles carrying amides at 3rd position of quinoline ring. Again we prepared different piperazine amides of quinolines at 2nd position with the important pharmacophoric group's methoxy, trifluoromethyl and fluoro group also we have been synthesized phenylsulfonylpiperazine quinolines with different groups on phenyl ring. Further we used acetylquinoline to prepare chalcones and biphenyl chalcones.

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