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Synthesis, characterization & antimicrobial screening of 1-(substituted phenyl)-2,5-bis(ethylsulfanyl)-1H-pyrrole-3,4-dicarbaldehydes

*A. P. Rajput and *A. R. Kankhare

P. G. Research Centre, Department of Chemistry, JET's Z. B. Patil College, Dhule (MS) India

ABSTRACT

The heterocyclic compounds are very much importance due to their biological activities. The formyl group attached to pyrrole molecules make them promising precursors for further synthetic transformation. Thioether compounds possess broad spectrum biological activities such as fungicidal[8], insecticidal, herbicidal & plant growth regulating activities. Some heterocyclic thioether derivatives have been also reported to exhibit good antiviral activities[7]. Due to these uses many workers are interested in the synthesis of thioether compounds with heterocycles[1-3]. To achieve our aim we diformylated succinimides using Vilsmeier- Haack reagent to get 2,5 dichloro - 3,4 diformyl (N- substituted phenyl) pyrroles. The succinimides in turn were obtained from succinic acid and substituted anilines. The diformylated compounds were then converted into thioether derivatives by treating with sodium sulphide, ethyl bromide in dry DMF as solvent[12-13]. All the compounds were characterized and screened for antimicrobial activities.

Key words: Vilsmeier- Haack reaction ,Formylation, Thioether derivatives

INTRODUCTION

A great variety of thioether compounds with heterocycles have been synthesized due to their broad spectrum biological activity such as fungicidal, insecticidal, herbicidal, & plant growth regulative activities. Some heterocyclic thioether derivatives exhibit good antiviral activities[4-6].

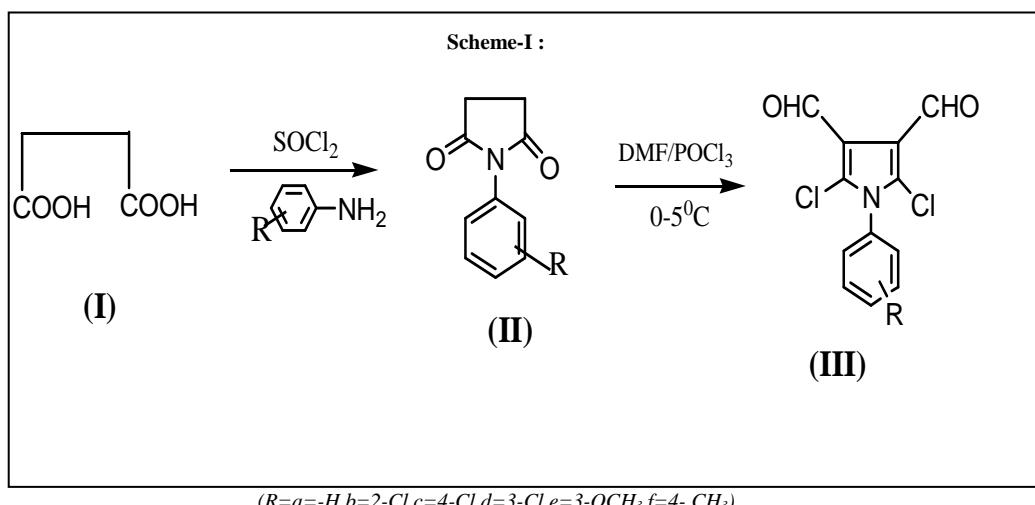
For this purpose, succinimides where diformylated using Vilsmeier-Haack reaction .to form 2, 5-dichloro -3, 4-difomyl (N-substituted phenyl) pyrroles. the succinimides in tern were synthesized from succinic acid & substituted anilines. the dichlorodifomyl pyrroles having formyl group & chlorine atom at ortho position to each other show promising precursors for other novel pyrrole derivatives.

MATERIALS AND METHODS

All melting points were determined in open capillary & are uncorrected. I.R. spectra were Recorded on Perkin-Elmer spectrum. ¹HNMR were recorded on Bruker DRX 500 MHz. NMR spectrometer with DMSO-d6 as a solvent using TMS as internal reference . (chemical shift in δ ppm).

General procedure for synthesis of 1-(N-substituted phenyl) -pyrrolidine -2,5-dione II (a-f)

A mixture of succinic acid (11.8 gm,0.1 m) & thionyl chloride (26.18 gm,2.2m) was refluxed for 30 minutes in 5 ml benzene , different aromatic amines (0.1 m) were dissolved. This solution of aromatic amines was added slowly in above reaction mixture. Then the reaction mixture was refluxed till complete HCl gas was evolved. The product obtained was cooled & Recrystallized from Ethanol to give pure 1 -(N- substituted phenyl)-pyrrolidine-2,5- dione II(a-f)

**(IIa) 1-(phenyl) pyrrolidine-2,5-dione.**

Mol. Formula: $\text{C}_{10}\text{H}_9\text{O}_2\text{N}$, Physical nature: shiny green, M.P: 150°C , Yield (%): 95%, IR (KBr) cm^{-1} : 2900 ($>\text{CH}_2$), 1726 ($>\text{C=O}$), 1464 (ArC=C), 1287 (C-N). H^1NMR (300 MHz, DMSO d₆, δ ppm): 2.78 (s,4H), 7.5-6.98 (m,3H,Ar-H). C^{13}NMR : 27.6(CH_2), 176($>\text{C=O}$), 135(C-N), 121(Ar-H). Elemental analysis Calculated for $\text{C}_{10}\text{H}_9\text{O}_2\text{N}$: C-68.57, H-5.14, N-8.05. Found: C-68.50, H-5.10, N-8.00.

(IIb) 1-(2-chloro phenyl) pyrrolidine-2,5-dione

Mol. Formula: $\text{C}_{10}\text{H}_8\text{O}_2\text{NCl}$, Physical nature: Yellowish, M.P: 150°C , Yield (%): 92%. IR (KBr) cm^{-1} : 2994 ($>\text{CH}_2$), 1702 ($>\text{C=O}$), 1419 (ArC=C), 1202(C-N). H^1NMR : (300MHz, DMSO-d₆, δ ppm): 2.70 (s,4H), 7.4-6.80 (m,3H,Ar-H). C^{13}NMR : 27.20 ($>\text{CH}_2$), 170 ($>\text{C=O}$), 134 (C-N), 120 (Ar-H) Elemental Analysis; Calculated For $\text{C}_{10}\text{H}_8\text{O}_2\text{NCl}$: C-57.00,H-4.27,N-6.65, Found : C-56.98, H-4.25, N-6.50,

(IIc) 1-(4-chloro phenyl) pyrrolidine -2,5-dione.

Mol. Formula: $\text{C}_{10}\text{H}_8\text{O}_2\text{NCl}$, Physical nature: light pink, M.P: 164°C , yield (%): 90%. IR (KBr) cm^{-1} : 2990 ($>\text{CH}_2$), 1710 ($>\text{C=O}$), 1420 (ArC=C), 1200 (C-N). H^1NMR (300 MHz,DMSO-d₆, δ ppm): 2.72 (s,4H), 7.3-6.90 (m,3H,Ar-H). C^{13}NMR : 26.2($>\text{CH}_2$), 170 ($>\text{C=O}$), 138(C-N), 128 (C-Cl). Elemental Analysis: Calculated For $\text{C}_{10}\text{H}_8\text{O}_2\text{NCl}$: C-57.00, H-4.27, N-6.65. Found: C-56.90, H-4.20, N-6.45.

(IId) 1-(3-Chlorophenyl) pyrrolidine -2,5 - dione

Mol. Formula : $\text{C}_{10}\text{H}_8\text{O}_2\text{NCl}$, Physical nature: yellowish, M.P: 115°C , Yield (%): 90%, IR (KBr) cm^{-1} : 2990 ($>\text{CH}_2$), 1730.32 ($>\text{C=O}$), 1425 (ArC=C), 1206 (C-N). H^1NMR : (300 MHz,DMSO-d₆, δ ppm): 2.78 (s,4H), 7.3- 6.90 (m,3H,Ar-H). C^{13}NMR : 27.2 ($>\text{CH}_2$), 174 ($>\text{C=O}$), 132 (C-N), 123 (Ar-H). Elemental Analysis: Calculated For $\text{C}_{10}\text{H}_8\text{O}_2\text{NCl}$: C-57.00, H-4.27, N-6.65. Found: C-56.90, H-4.15, N-6.50.

(IIe) 1-(3-methoxy phenyl) pyrrolidine -2,5-dione

Mol. Formula: $\text{C}_{11}\text{H}_{11}\text{O}_3\text{N}$, Physical nature:shiny black, M.P: 170°C , Yield (%): 95%, IR (KBr) cm^{-1} : 2924 ($>\text{CH}_2$), 1723 ($>\text{C=O}$), 1419 (ArC=C), 1202 (C-N). H^1NMR : (300 MHz, DMSO-d₆, δ ppm): 2.78 (s,4H), 3.648 (s,3H,CH₃), 7.6-7 (m,3H,Ar-H). C^{13}NMR : 26.5 ($>\text{CH}_2$), 137 (C-N), 120 (Ar-H), 172 ($>\text{C=O}$). Elemental Analysis: Calculated For $\text{C}_{11}\text{H}_{11}\text{O}_3\text{N}$; C-64.39, H-5.36, N-6.82. Found: C-64.30, H-5.25, N-6.50.

(II f) 1-(4- methyl phenyl) pyrrolidine -2,5- dione

Mol. Formula: $\text{C}_{11}\text{H}_{11}\text{O}_2\text{N}$, Physical nature: shiny brown, M.P: 150°C , Yield (%): 90%. I.R (KBr) cm^{-1} : 2928($>\text{CH}_2$), 1700 ($>\text{C=O}$), 1400 (Ar C=C), 1205 (C-N). H^1NMR : (300MHz, DMSO-d₆, δ ppm): 2.60 (s,4H), 2.8 (s,3H,CH₃), 7.5-7 (m,3H,Ar-H). C^{13}NMR : 25.6($>\text{CH}_2$), 132 (C-N), 120 (Ar-H), 170 ($>\text{C=O}$). Elemental analysis: Calculated For $\text{C}_{11}\text{H}_{11}\text{O}_2\text{N}$; C-69.84, H-5.82, N-7.40. Found: C-68.90, H-5.60, N-7.10.

General procedure for the synthesis of 2,5-dichloro-1-(N-substituted phenyl)-1H- pyrrole -3,4-dicarbaldehyde III(a-f)

To a cooled dimethyl formamide (0.24 moles) freshly distilled POCl_3 (0.12 moles) was slowly added in a dropwise fashion with constant stirring at $5-10^\circ\text{C}$. Then the succinimides II (a-f) (0.12 moles) were slowly added to a cooled Vilsmeier - Haak reagent in small aliquots at a time with constant stirring using magnetic stirrer. The reaction mixture was heated at $60-70^\circ\text{C}$ For 6 hrs. This mixture was kept overnight & was then slowly added to crushed ice

with stirring & stirred for another 30 min .Then the resulting clear coloured solution was reacted with 40% NaOH (50ml) maintaining the temp below 50°C.The reaction mixture was then heated at 50-60°C. for half an hour which after cooling in an ice bath coloured compounds were obtained. These compounds were recrystallized with aqueous methanol as solvent to give pure product III(a-f)

(IIIa) 2,5-dichloro-1-(phenyl) -1H- pyrrole-3,4-dicarbaldehyde.

Mol. Formula: C₁₂H₇O₂NCl₂, Physical nature: yellowish, M.P: 162°C, Yield (%): 85%. IR (KBr) cm⁻¹: 2860 (-CHO), 1705.93 (>C=O), 1507.61 (ArC=C), 1187 (C-N), H¹NMR: (300MHz,DMSO-d₆, δppm): 7.8-7.28 (s,4H, Ar-H), 10.38 (br.s, 2H,2-CHO). C¹³NMR:141 (C-N), 129 (Ar-H), 119 (C-Cl), 191 (-CHO) Elemental Analysis: Calculated For C₁₂H₇O₂NCl₂; C-53.73, H 2.61, N-5.22. Found : C-53.70, H-2.58, N-5.19.

(IIIb) 2,5-dichloro-1-(2-chloro phenyl)-1H-Pyrrole -3,4-dicarbaldehyde.

Mol. Formula: C₁₂H₆O₂NCl₃, Physical nature: yellowish, M.P: 90°C, Yield (%): 80%. IR: (KBr)cm⁻¹: 2850 (-CHO), 1705.80 (>C=O), 1500 (ArC=C), 1190 (C-N), H¹NMR: (300 MHz,DMSO-d₆, δppm): 7.5-7.20 (s,4H, Ar-H), 10.20 (br.s, 2H,2-CHO). C¹³NMR: 139 (C-N) , 125 (Ar-H), 115 (C- Cl), 190 (-CHO). Elemental Analysis: Calculated For C₁₂H₆O₂NCl₃ C-47.60, H-15.20, N-4.62,. Found; C-47.55, H 15.18, N-4.60.

(IIIc) 2,5- dichloro-1-(4-chloro phenyl)-1H- pyrrole -3,4- dicarbaldehyde.

Mol. Formula: C₁₂H₆O₂NCl₃, Physical Nature: brownish, M.P: 242°C, yield (%): 82%, IR (KBr) cm⁻¹: 2865 (-CHO), 1700.93 (>C=O), 1507.61 (ArC=C), 1188 (C-N), H¹NMR: (300 MHz,DMSO-d₆, δppm): 7.4-7.20 (s,4H, Ar-H), 10.33 (br.s, 2H,2-CHO). C¹³NMR: 142 (C-Cl), 195 (-CHO), 133 (Ar-H), 135(C-N). Elemental Analysis; Calculated For C₁₂H₆O₂NCl₃ C-47.60, H-15.20, N-4.62. Found: C-47.58, H-15.10, N-4.57.

(IIId) 2,5-dichloro -1- (3- chloro phenyl) -1H-pyrrole -3,4- dicarbaldehyde

Mol. Formula: C₁₂H₆O₂NCl₃, Physical nature: brownish, M.P: 84°C, yield (%): 85%. IR (KBr) cm⁻¹: 2869 (-CHO), 1700 (>C=O), 1500 (ArC=C), 1180 (C-N), H¹NMR (300 MHz,DMSO-d₆, δppm): 7.7-7.20 (s,4H, Ar-H), 10.28 (br.s, 2H,2-CHO). C¹³NMR: 145 (C-N), 129 (Ar-H), 190 (-CHO), 119 (C-Cl), Elemental Analysis: Calculated For C₁₂H₆O₂NCl₃ C-47.60, H-15.20, N-4.62, Found: C-47.58, H-15.18, N-4.60.

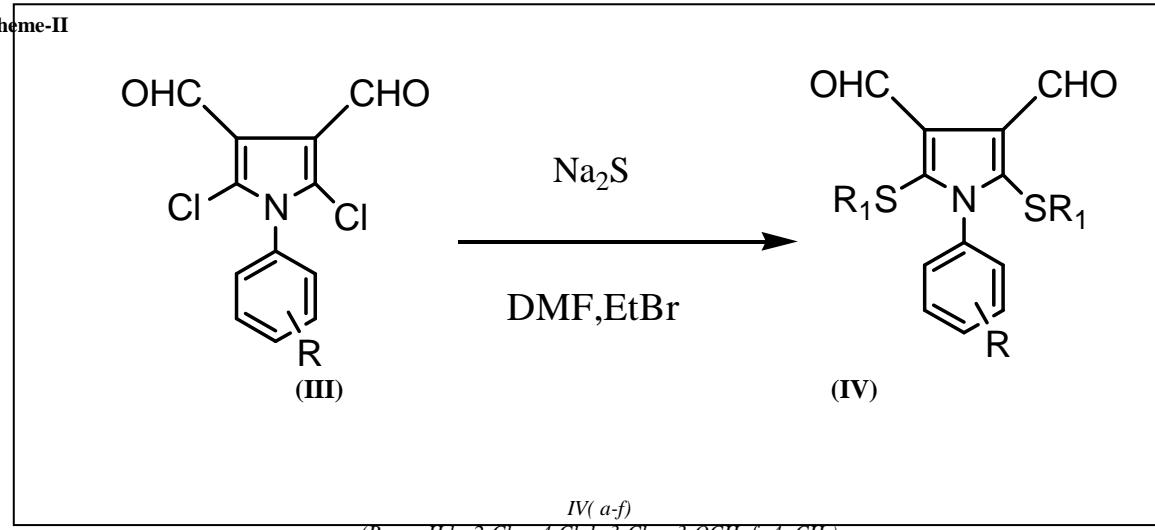
(IIIe)2,5-dichloro-1-(3-methoxyphenyl)-1H-pyrrole-3,4-dicarbaldehyde.

Mol. Formula: C₁₃H₉O₃NCl₂, Physical nature: faint white, M.P: 119°C, yield (%): 85% . IR (KBr) cm⁻¹: 2861 (-CHO), 1705 (>C=O), 1590 (ArC=C), 1189 (C-N), H¹NMR: (300 MHz, DMSO-d₆, δppm): 3.58 (s,3H-OCH₃), 7.6-7.33 (S,4H, Ar-H), 10.40 (br.s, 2H,2-CHO). C¹³NMR: 120 (C-Cl), 130 (Ar-H), 190 (-CHO), 135 (C-N). Elemental Analysis: Calculated For C₁₃H₉O₃NCl₂ C-52.34, H-3.02, N-4.69. Found: C-52.30, H-3.00, N-4.65.

(IIIIf) 2,5- dichloro-1- (4-methyl phenyl) -1H- pyrrole -3,4- dicarbaldehyde.

Mol. Formula: C₁₃H₉O₂NCl₂, Physical nature: brownish, M.P: 186°C, yield (%): 70%. IR: (KBr) cm⁻¹: 2860 (-CHO), 1708.99 (>C=O), 1507.69 (ArC=C), 1193 (C-N), H¹NMR: (300 MHz,DMSO-d₆, δppm): 7.1-7.20 (s,4H, Ar-H), 10.39 (br.s, 2H,2-CHO). C¹³NMR: 140 (C-Cl), 190 (-CHO), 135 (Ar-H), 134 (C-N). Elemental Analysis: Calculated For C₁₃H₉O₂NCl₂ C-55.31, H-3.19, N-4.96. Found: C-55.28, H-3.13, N-4.90.

Scheme-II



General procedure for synthesis of 2,5-bis (ethyl sulfany1)-1- (N-substituted phenyl-1H-pyrrole-3,4-dicarbaldehydes. IV (a-f)

To a solution of III (1 mmol) in dry DMF (5ml) sodium sulphide (3mmol, fused flakes) was added & stirred for 2-3 hr at rt. On completion of the reaction, the corresponding halo compound (ethyl bromide) was added & stirred for another 30-45 min.& poured in to crushed ice. The ppt obtained was filtered, dried & purified by recrystallisation from aq. ethanol to give corresponding pure 2,5-bis (ethyl sulfanyl) -1-(N-substituted phenyl)-1H- pyrrole-3,4-dicarbaldehydes .(IVa-f)

(IVa) 2,5-bis (ethyl thio) -1- phenyl-1H- pyrrole -3,4- dicarbaldehyde.

Mol. Formula: $C_{16}H_{17}O_2NS_2$, Physical nature: whitish, Mol. Wt. 287, M.P: 222^0C , yield (%): 85%. IR (KBr) cm^{-1} 2852 (-CHO), 1703 ($>\text{C=O}$), 1448 (Ar C=C), 1236 (C-N), 1016 (-SC₂H₅). H¹NMR (300 MHz,DMSO-d₆ δppm): 1.25 (t,6H, J=7Hz, 2SCH₂CH₃), 7.33-7.13 (m, Ar-H), 10.15 (s,2H,2CHO), 3.08 (q, 4H,J= 7Hz, 2S CH₂CH₃). C¹³NMR 125 (Ar-H), 140 (CN), 191 (-CHO), 30.4 ($>\text{CH}_2$), 14.4 (-CH₃). Elemental Analysis: Calculated For $C_{16}H_{17}O_2NS_2$ C-66.89, H-5.92, N-4.87. Found: C-66.80, H-5.88, N-4.82.

(IVb) 1-(2-chloro phenyl) -2,5- bis (ethyl thio) -1H-pyrrole -3,4- dicarbaldehyde

Mol. Formula: $C_{16}H_{16}O_2NS_2Cl$, Physical nature: whitish, Mol. Wt 321.5, M.P: 92^0C , yield (%): 87% . IR(KBR) cm^{-1} 2850 (-CHO), 1700 ($>\text{C=O}$), 1440 (ArC=C), 1235 (C-N), 1050(SC₂H₅). H¹NMR (300MHz,DMSO-d₆ δppm) 1.20 (t,6H,J=7Hz,2SCH₂CH₃), 7.30-7.10 (m,Ar-H), 10.10 (s,2H,2CHO), 3.00 (q,4H,J=7Hz,2s CH₂CH₃). C¹³NMR: 120-122 (Ar-H), 139 (C-N), 134 (C-Cl), 190 (-CHO), 29.4 (S-CH₂), 14.2 (-CH₂-CH₃). Elemental Analysis : Calculated For $C_{16}H_{16}O_2NS_2Cl$, C-59.72, H-4.97, N-4.35. Found, C-59.68, H-4.92, N-4.30.

(IVc) 1-(4-chloro phenyl)-2-5- bis (ethyl thio)-1H-pyrrole -3,4- dicarbaldehyde

Mol. Formula: $C_{16}H_{16}O_2NS_2Cl$, M.P: 162^0C , Yield (%): 82%. IR (KBR) cm^{-1} 2850 (-CHO), 1700 ($>\text{C=O}$), 1440 (Ar-C=C), 1230 (C-N), 1015 (-SC₂H₅). H¹NMR: (300MHz,DMSO-d₆ δppm) 1.22 (t,6H,J=7Hz,2SC₂H₅), 7.30-7.10 (m,Ar-H), 10.10 (s,2H,2CHO), 3.05 (q,4H,J=7Hz,2SC₂H₅). C¹³NMR: 122 (Ar-H), 142 (C-N), 190 (-CHO), 30.3 ($>\text{CH}_2$), 14.2 (-CH₃). Elemental Analysis: Calculated For $C_{16}H_{16}O_2NS_2Cl$. C-59.72, H-4.97, N-4.35. Found C-59.70, H-4.90, N-4.30.

(IVd) 1-(3- chloro phenyl)- 2,5-bis (ethyl thio) -1H- pyrrole-3,4- dicarbaldehyde

Mol formula: $C_{16}H_{16}O_2NS_2Cl$, Physical nature: Greyish, Mol wt. 321.5, M.P: 172^0C , Yield (%) : 88%. IR (KBr) cm^{-1} 2855 (-CHO), 1702 ($>\text{C=O}$), 1445 (Ar-H), 1233 (C-N), 1107 (-SC₂H₅). H¹NMR (300MHz,DMSO-d₆, δ ppm) 1.25 (t,6H,J=7Hz,2SC₂H₅), 7.25-7.5 (m,Ar-H), 10.15 (s,2H,2CHO), 3.05 (q,4H,J=7Hz,2SCH₂CH₃). C¹³NMR 128 (ArH), 139 (CN), 192 (-CHO), 30.2 ($>\text{CH}_2$), 14.3 (-CH₃) Elemental Analysis :Calculated For $C_{16}H_{16}O_2NS_2Cl$, C-59.72, H-4.47, N-3.35, Found: C-59.70, H-4.40, N-3.30.

(IVe) 1(3- methoxy phenyl) -2,5- bis (ethyl thio)-1H- pyrrole -3,4- dicarbaldehyde

Mol. Formula: $C_{17}H_{19}O_3NS_2$, Physical nature: whitish, Mol. Wt. 317, M.P; 119^0C , Yield(%): 82%. IR (KBr) cm^{-1} 2852 (-CHO), 1703 ($>\text{C=O}$), 1442 (ArC=C), 1236 (C-N), 1110 (-SCH₂CH₃). H¹NMR (300 MHz, DMSO-d₆, δ ppm) 1.28 (t,6H, J=7Hz,2SCH₂CH₃), 7.33- 7.13 (m,ArH), 10.15 (s,2H,2CHO), 3.08 (q,4H,J=7Hz,2SC₂H₅). C¹³NMR: 125(Ar-H), 140(C-N), 191(-CHO), Elemental Analysis: Calculated For $C_{17}H_{19}O_3NS_2$, C-64.35, H-5.99. found: C-64.30, H-5.90.

(IVf) 1-(4-methyl phenyl) -2,5-bis (ethyl thio) -1H- pyrrol -3,4- dicarbaldehyde

Mol. Formula $C_{17}H_{19}O_2NS_2$ Physical nature yellowish Mol. Wt. 301, M.P. 164^0C yield(%) 87% IR (KBr) cm^{-1} 2857(CHO),1705 ($>\text{C=O}$),1441(ArC=C),1230(C-N),1099 (-SCH₂CH₃). H¹ NMR (300 MHz, DMSO-d₆, δ ppm) 1.35(t,6H,J=7Hz,2SCH₂CH₃),7.30-7.13 (m,Ar-H),10.18(s,2H,2CHO),3.08(q,4H,J=7Hz,2SC₂H₅). C¹³NMR: 122(Ar-H),135(C-N),192(CHO). Elemental Analysis: Calculated For, $C_{17}H_{19}O_2NS_2$ C-67.77,H-6.31. Found C-67.70,H-6.30.

Table-1 shows physical data of compound

| Comp | R | M.F | M.P (°C) | Yield (%) |
|------|--------------------|---|-------------|-----------|
| IIa | -H | C ₁₀ H ₉ O ₂ N | 150 | 95 % |
| IIb | 2-Cl | C ₁₀ H ₈ O ₂ NCl | 150 | 92 % |
| IIc | 4-Cl | C ₁₀ H ₈ O ₂ NCl | 164 | 90 % |
| IId | 3-Cl | C ₁₀ H ₈ O ₂ NCl | 115 | 90 % |
| IIe | 3-OCH ₃ | C ₁₁ H ₁₁ O ₃ N | 170 | 95 % |
| IIIf | 4-CH ₃ | C ₁₁ H ₁₁ O ₂ N | 150 | 90 % |
| IIIa | -H | C ₁₂ H ₇ O ₂ NCl ₂ | 162 | 85 % |
| IIIb | 2-Cl | C ₁₂ H ₆ O ₂ NCl ₃ | 90 | 80 % |
| IIIc | 4-Cl | C ₁₂ H ₆ O ₂ NCl ₃ | 242 | 82 % |
| IID | 3-Cl | C ₁₂ H ₆ O ₂ NCl ₃ | 84 | 85 % |
| IIIe | 3-OCH ₃ | C ₁₃ H ₉ O ₃ NCl ₂ | 119 | 85 % |
| IIIf | 4-CH ₃ | C ₁₃ H ₉ O ₂ NCl ₂ | 186 | 70 % |
| IVa | -H | C ₁₆ H ₁₇ O ₂ NS ₂ | 222 | 85 % |
| IVb | 2-Cl | C ₁₆ H ₁₆ O ₂ NS ₂ Cl | 92 | 87 % |
| IVc | 4-Cl | C ₁₆ H ₁₆ O ₂ NS ₂ Cl | 162 | 82 % |
| IVd | 3-Cl | C ₁₆ H ₁₆ O ₂ NS ₂ Cl | 172 | 88% |
| IVe | 3-OCH ₃ | C ₁₇ H ₁₉ O ₃ NS ₂ | 119 | 82 % |
| IVf | 4-CH ₃ | C ₁₇ H ₁₉ O ₂ NS ₂ | 164 | 87 % |

Biological Testing of compounds.

Heterocyclic thioether compounds IV(a-f) were evaluated for antibacterial against fungi Escherichia coli (EC), pseudomonas S. aeruginosa (PA), staphylococcus aureus (SA), Bacillus subtilis (BS), candida albicans (CA), Aspergillus niger (AN).

The result were obtained in the form of clearing zone and were after the period of incubation (37°C for 24 hrs). The zone of inhibition was measured in mm and data is presented in table 2.

CULTURE USED:

| Culture abbreviation | Culture name | Culture code |
|----------------------|-------------------------------|--------------|
| ES | <i>Escherichia coli</i> | NCIM 1209 |
| PA | <i>Pseudomonas aeruginosa</i> | NCIM 2036 |
| SA | <i>Staphylococcus aureus</i> | NCIM 2079 |
| BS | <i>Bacillus subtilis</i> | NCIM 2250 |
| AN | <i>Aspergillus niger</i> | NCIM 545 |

Table-2 Antimicrobial activity of compounds(Zone of inhibition in mm)

| Sr. No. | Compound | EC | PA | SA | BS |
|---------|-----------------|-------|-------|-------|-------|
| 1 | IIa | 10.36 | 11.21 | 10.14 | 10.12 |
| 2 | IIb | 10.14 | 12.21 | 12.21 | 11.14 |
| 3 | IIc | 10.22 | 11.23 | 10.12 | 11.12 |
| 4 | IId | 11.89 | 11.88 | 10.88 | 11.45 |
| 5 | IIe | 11.33 | 11.02 | 11.14 | 10.45 |
| 6 | IIIf | 10.25 | 12.23 | 10.47 | 10.98 |
| 7 | IIIa | 10.23 | 10.12 | 10.23 | 10.56 |
| 8 | IIIb | 13.11 | 10.52 | 10.01 | 10.89 |
| 9 | IIIc | 12.55 | 10.41 | 11.11 | 10.54 |
| 10 | IIId | 10.10 | 11.02 | 10.88 | 10.87 |
| 11 | IIIe | 10.89 | 11.08 | 11.12 | 11.25 |
| 12 | IIIf | 10.12 | 10.48 | 10.78 | 10.78 |
| 13 | IVa | - | - | 13.66 | 12.36 |
| 14 | IVb | - | - | 12.97 | 13.56 |
| 15 | IVc | - | - | 7.47 | - |
| 16 | IVd | - | - | 7.76 | 8.15 |
| 17 | IVe | - | - | 7.78 | 9.23 |
| 18 | IVf | - | - | 9.83 | 10.12 |
| 19 | Chloramphenicol | 28.67 | 24.44 | 29.63 | 26.30 |
| 20 | Ciprofloxacin | 21.11 | 22.23 | 22.33 | 21.34 |

Media used :

For bacteria : Nutrient agar (Hi-media)

For yeast : MGYP

Inoculum size :

Bacteria : 1 x 10 bacteria per ml.

Yeast : 1 x 10 cells per ml.

❖concentration of compound

(Prepared in ethanol) 100 μ gm 1 disc method used(disc method, disc size 6mm) “__” means no zone of inhibition.

CONCLUSION

In present work we have developed a general method for the synthesis of succinimides with good yield. which can be used for the synthesis of various heterocyclic system. The dicarbaldehydes formed (IIIa-f) are unknown synthones and may be used for the synthesis of various heterocyclic system.

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